

Energy, Mines and Resources Canada

Energie, Mines et -Ressources Canada

Direction de la physique du globe

Earth Physics Branch

1 Observatory Crescent Ottawa Canada K1A 0Y3

1 Place de l'Observatoire Ottawa Canada K1A 0Y3

Geothermal Service of Canada

Service géothermique du Canada

INVESTIGATION OF FREEZING SOILS USING TIME DOMAIN REFLECTOMETRY

M.W. Smith and D.E. Patterson

Geotechnical Science Laboratories, Carleton University

Earth Physics Branch Open File Number 81 - 6

Ottawa, Canada, 1981

54 p.

NOT FOR REPRODUCTION

R

Price/Prix: \$18.00

EPB Open File 81-6

This document was produced by scanning the original publication.

Ce document est le produit d'une numérisation par balayage de la publication originale.

Abstract

The report describes the use of the time-domain reflectometry technique to determine soil freezing characteristic curves. In addition the influence of partial saturation, bulk density variation and percent clay on those characteristic curves is examined. Preliminary evaluation suggests that excess ice contents may also be inferred by comparing the apparent dielectric constant for the icy soil with that measured at normal field density.

Résumé

Ce rapport décrit l'utilisation de la technique de réflectométrie par intervalles de temps pour déterminer les courbes caractéristiques de congélation de sols. En plus l'influence de la non-saturation, des variations de la densité globale et du pourcentage d'argile sur ces courbes caractéristiques out été examinés. Les résultats préliminaires suggèrent qu'une teneur en glace excessive peut être déduite en comparant la constante diélectrique apparente d'un sol contenant de la glace avec la constante mesurée sur le sol à sa densité normale sur le terrain.

FINAL REPORT

INVESTIGATION OF FREEZING SOILS USING

TIME DOMAIN REFLECTOMETRY

for the

Department of Energy, Mines and Resources

Earth Physics Branch

by

Michael W. Smith Principal Investigator

and

Daniel E. Patterson Research Associate

Geotechnical Science Laboratories Department of Geography Carleton University Ottawa, Ontario K1S 5B6

DSS File No. 14SU-23235-0-0584 Serial No. 0SU80-00059

1

1

Table of Contents

			rage		
1.	Intr	oduction	1		
2.	Expe	erimental Procedures	3		
	2.1 2.2 2.3	Freezing Characteristic Curves Freezing Column Experiments K Determination for Low Density Samples a	3 3 7		
3.	Resu	lts and Discussion	10		
	3.1 3.2	Freezing Characteristic Data Freezing Column Experiments	10 16		
	2.2	Excess Ice	24		
4.	Conc	lusions	40		
Acknowledgements					
Ref	erenc	es	42		
App	endix	I Tabulation of the Emperical Relationship Given by Topp Davis and Annan (1980)	43		
App	endix	II Freezing Characteristic Data	45		
App	endix	: III Inferring Maximum Ice Content from Measured θ_{uf} : Program Listing	51		

Page

List of Figures

Figure	1	Parallel Transmission Line Freezing Cell	4
Figure	2	Design for Freezing Column Experiments	6
Figure	3	Coaxial Line Freezing Cell	8
Figure	4	Freezing Characteristic Data: Sandy Soils	12
Figure	5	Freezing Characteristic Data: Clayey Soils	13
Figure	6	Freezing Characteristic Data: Norwegian Soils.	15
Figure	7	Unfrozen Water Content Versus Clay Content	17
Figure	8	Freezing Characteristic Data for Allendale Silty Clay at Various Degrees of Saturation	18
Figure	9	Results from Freezing Column Experiment: Ellwood Clay Loam	19
Figure	10	Results from Freezing Column Experiment: Cavan Loam	21
Figure	11	Theoretical Freezing Characteristic Curves for Various Dry Bulk Densities	25
Figure	12	Theoretical Changes in K with Dry Bulk Density • at Three Temperaturesa	26
Figure	13	Variation in Freezing Characteristic Curve with Bulk Density (Manchester Silt)	28
Figure	14	w - T [°] C for Manchester Silt ·····	29
Figure	15	Variation in Freezing Characteristic Curve with Bulk Density (Castor Sandy Loam)	30
Figure	16	w - T [°] C for Castor Sandy Loam	31
Figure	17	Variation in Freezing Characteristic Curve with Bulk Density (Allendale Silty Clay)	32
Figure	18	w - T ^o C for Allendale Silty Clay	33
Figure	19	Nomogram for Inferring Maximum Ice Content from Measured θ_{uf}	35

1

Page

Page

Figure	20	K for Frozen Ottawa Sand at Different Initial Water Contents	37
Figure	21	TDR Traces for Allendale Silty Clay with Excess Ice	38

2

ł

• 1

1

List of Tables

. 1

I

I

3

1 1

. 1

¢

			Page
Table	1	Grain Size Characteristics	11
Table	2	Freezing Column Results (Ellwood Clay Loam)	20
Table	3	Freezing Column Results (Cavan Loam)	22

Introduction

An empirical relationship between the apparent dielectric constant, K_a , and the volumetric liquid water content of a mineral soil, determined by Topp et al. (1980), was found previously to apply to freezing soils (Smith and Patterson, 1980). Experiments have been continued to obtain further data on the use of the TDR technique to determine soil freezing characteristic curves (θ_{uf} vs. T ^oC). Data have been obtained for a variety of soils, and compared to values determined by other methods where possible. These results lend further support to the applicability to frozen soils of the empirical relationship determined for unfrozen soils by Topp et al. (1980).

Data from the TDR experiments have also been used to evaluate the influence of the percent saturation, the variation in bulk density, and the percent clay content on the θ_{uf} vs. T ^oC relationship.

Preliminary instrumentation of a laboratory soil column has been carried out to examine whether the TDR technique can be used to monitor changes in soil hydrology during freezing experiments. For closed system freezing, which involves only the internal redistribution of water in the soil profile, the TDR technique was used to measure the water loss in the unfrozen zone. This was then compared to the water gain (by ice accumulation) in the frozen zone, determined gravimetrically at the end of the experiment.

Clearly, it would be more desirable to be able to directly measure the total water content (ice plus water) in the frozen zone, during such experiments. Further, in open system freezing experiments there may not be any desiccation of the unfrozen zone (due to a constant water supply), and one clearly needs a measure of the total water content in the frozen zone. To accomplish this, the TDR technique could be combined with a gammaray apparatus, for example. However, a single system, if possible, would seem preferable.

No simple extension to the TDR technique has yet been found to permit direct measurement of either ice content or ice plus water content. Other methods do exist for determining the electrical properties of ice and water (e.g., loss via tangent, capacitance, or resistivity measurement); however, incorporating them with the TDR technique may not be possible. Investigation of this is continuing.

In the meantime, a method of inferring maximum possible ice content, from TDR measurement of K_a , has been devised. Since K_a is highly sensitive to the volumetric liquid water content, then the value will decrease as ice replaces water in a unit volume of soil. Therefore, the effect of excess ice accumulation in a soil will be to lower the value of K_a , measured at some temperature, compared to that for the same soil at normal field density. This difference in K_a can be used to calculate a "maximum possible ice content".

1

SECTION 2

EXPERIMENTAL PROCEDURES

2.1 Freezing Characteristic Curves

When parallel transmission lines are used, a slurried sample is placed in a PVC tube 20 cm long and 5 cm I.D., and allowed to consolidate over a few days. A balanced parallel line probe of 0.3 cm (1/8 in.) stainless steel rod is then inserted in the sample; a line length of 17.5 cm was used, with a line spacing of 2.5 cm (Figure 1). The PVC tube is covered with a latex rubber membrane and immersed in a circulating ethylene glycol bath, which permits temperature control to $\pm 0.01^{\circ}$ C for extended periods of time. Freezing is initiated by subjecting the sample to a temperature of about -4° C for 30 to 45 minutes after it has stabilised near 0° C for 24 hours. The sample is then ramped through a temperature cycle and K_a (hence θ_{uf}) determined at each temperature. This type of freezing cell can be used for both freezing and thawing cycles, since tube rupture due to freezing is not a problem.

To compile a freezing characteristic curve takes about 2-3 days for a silt loam, allowing 4 or 5 hours for each point to reach equilibrium. One can check for equilibrium by monitoring K_a a number of times at each temperature until a steady TDR trace is obtained. For a heavy clay at temperatures between 0° and -1°C, it can take up to 24 hours to reach equilibrium in the sample. A complete curve down to -5°C can be obtained in 4-6 days, depending on the number of points desired.

2.2 Freezing Column Experiments

One procedure used to obtain data on the rates and amount of water





movement toward the freezing front during closed system soil freezing is to replicate soil profiles, freeze them for various time periods and then section the profiles to obtain water content with depth gravimetrically (e.g., Torrance and Williams, 1976; Mageau and Morgenstern, 1980). This replication process is time-consuming and can introduce experimental errors which can make the interpretation of results difficult. Also, this approach cannot be used on undisturbed soil cores obtained from the field. Since the TDR technique is a non-destructive method, the need for duplicating soil profiles is removed and the possibility of instrumenting undisturbed soil cores exists (if the core diameter is large enough, say about 12.5 - 15 cm).

A freezing column, shown in Figure 2, was constructed to permit controlled sample freezing and water content measurement via the TDR technique. Two cooling plates were used to control the end temperatures, with the top plate at sub-freezing temperatures. Balanced parallel transmission lines, installed horizontally at 2.5 cm intervals along the column length, were used for water content measurement. Closed-system freezing experiments have been carried out initially, to examine the use of the TDR technique for monitoring the desiccation of the unfrozen zone as water migrates towards the freezing front.

Soil samples (of Ellwood clay loam and Cavan loam) were slurried, placed in the column and allowed to consolidate for several days. The column was then insulated, placed in a controlled temperature chamber (at ~ 1° C) and the temperature of both cooling plates set to 1° C. Once the sample temperature had stabilized at 1° C, the temperature of the top plate was stepped down to about -4° C to initiate sample freezing. The



Figure 2 Design for Freezing Column Experiments

temperature of the top plate was then set to a temperature of -1° C. K_a (and hence θ_v) was measured throughout the unfrozen soil at the beginning and end of each experiment (which lasted about 120 hours). Also, at the end of the experiment the whole column was cored, using a brass ring corer (2.5 cm in length, 5.1 cm I.D.), to determine the water content profile gravimetrically. Some problems were encountered with this where the soil tended to crumble.

Test results are presented in Section 3.2.

2.3 K Determination for Low Density Samples

Experiments were performed to obtain K_a vs T^oC data for samples with dry bulk densities less than 1 g cm⁻³, for example, as in the case of soils containing excess ice. A small coaxial freezing cell was designed for these experiments; it is 15 cm long, has an inside diameter of 2.5 cm and a 0.64 cm ($\frac{1}{4}$ in.) diameter centre conductor (Figure 3). With coaxial freezing cells, care must be taken that the centre conductor is not heaved, thereby possibly breaking the connection at the BNC connector (see Figure 3). To avoid this, a soil slurry is cooled to about 0^oC and added in small amounts to the cell. After each addition of soil, the cell is placed in a freezer at -18° C to rapidly freeze the sample. This is repeated until the coaxial cell is filled. The cell is then covered with a latex membrane and placed in the temperature bath.

A sample containing excess ice, but without banded ice, can be obtained if a fine-grained super-saturated slurry is used and is added in small amounts to the cell, since freezing will occur before the particles settle out of suspension. Total water content and bulk density





Aluminium Top Plate 0.150" Thick 1.322" O.D.

. 1

- A 43 Drill Tapped for 4-40 Screws
- B 3/32 " Clearance Hole
- C 7/16" Centre Hole



Plexiglass Bottom Plate

Figure 3 Coaxial Line Freezing Cell

were known since all masses and volumes were determined prior to and following each experiment. By comparing K values for low bulk density samples to K values for samples of "normal" bulk density, some inference about ice content can be made. Experimental results are presented in Section 3.3.

In a number of other experiments, discrete ice lenses were formed by adding water to the cell between soil layers. The ice lenses form impedance contrasts along the transmission line, and produce a signature on the TDR trace (see Section 3.3).

SECTION 3

RESULTS AND DISCUSSION

3.1 Freezing Characteristic Data

Ţ

Freezing characteristic data were determined for a variety of soils using the TDR technique. θ_{uf} was determined from K_a measurements using an empirical relationship obtained by Topp <u>et al</u>. (1980) for unfrozen mineral soils (Appendix I), which has been found to apply also to freezing soils (Smith and Patterson, 1980).

Two broad groups of soils were selected, sandy and clayey, with several different soils within each group. The results have been examined with respect to grain size characteristics, dry bulk density, and degree of saturation. Grain size characteristics for all samples tested are presented in Table 1. Freezing characteristic data are presented in Figures 4 and 5 for the sandy and clayey soils respectively. The points shown represent the mean value if more than one sample was used. A complete tabulation of all test results is given in Appendix II.

Upon examining Figure 4, there is a great deal of variability in $\theta_{\rm uf}$ at temperatures above -0.5° C for the sandy soils (a range of about 20% at -0.1° C). However, the $\theta_{\rm uf}$ data at temperatures colder than -0.5° C generally agree to within $\pm 2\%$. Previous experiments on the reproducibility of freezing characteristic data for samples of the same soil type showed agreement to well within $\pm 2\frac{1}{2}\%$ in $\theta_{\rm uf}$ at all temperatures (Smith and Patterson, 1980). Therefore, it is concluded that the scatter portrayed in Figure 4 at temperatures above -0.5° C are due to variations in properties between soil types (e.g., pore size distribution).

TABLE 1

GRAIN SIZE CHARACTERISTCS

	% SAND	% SILT	% CLAY	SOIL DESIGNATION*
Castor	66.1	30.4	3.5	sandy loam
Castor <74u	55.4	40.6	4.0	sandy loam
Garden	64.6	25.1	10.3	sandy loam
Agri	56.9	24.7	17.6	sandy loam
Field	84.8	7.1	8.1	loamy sand
Loeb	35.0	31.7	33.3	clay loam
Allendale	17.7	33.9	48.5	clay
Agri	12.0	45.7	41.7	silty clay
LE2	2.9	58.3	38.8	silty clay loam
LE7	10.8	29.7	59.5	clay
Ellwood	39.8	24.5	35.7	clay loam

* U.S. Department of Agriculture Classification System



1

ŝ

I.

Figure 4 Freezing Characteristic Data: Sandy Soils



Figure 5 Freezing Characteristic Data: Clayey Soils

The agreement in freezing characteristic curves for various clayey soils (Figure 5) is satisfactory. The Ellwood and Loeb soils, which have very similar grain size characteristics (see Table 1), agree to within 1% in θ_{uf} at all temperatures. Overall, the range in θ_{uf} values at -0.25° C is about 5\% for all the soils; it decreases to about 3\% at -2.5° C. TDR freezing characteristic data for LE2 and LE7 Norwegian clay samples were compared to similar data derived from suction-moisture tests by Williams (1975). These are shown in Figure 6 and agreement is within 2% in θ_{uf} .

Suction-moisture data for the Agri sandy loam and Agri silty clay at -0.54°C, obtained from Stalker (in progress), also agree favourably with TDR determined θ_{uf} (Figures 4 and 5, Appendix II).

The freezing characteristic data were examined on the basis of grain size to see if any relationship exists. Dillon and Andersland (1966) and Anderson and Tice (1972) attempted to predict unfrozen water content from specific surface area, which can be measured or estimated; since the clay size fraction contributes most to the surface area of a soil, it was decided to compare TDR measurements of $\theta_{\rm uf}$ to the clay content of the soil.

The data for finer textured materials are for warming cycles; those for coarser textured samples are for freezing cycles. Because freezethaw hysteresis occurs in soils, the warmest temperature examined was -0.5° C (below this, the differences in θ_{uf} for warming and cooling cycles would be small in the coarser textured soils). θ_{uf} data previously determined by the TDR technique, found in Smith and Patterson (1980), were also used.

1



Figure 6 Freezing Characteristic Data: Norwegian Soils

ş

1

The results, shown in Figure 7, show a strong correlation between θ_{uf} and the percentage of soil particles finer than 2μ . Although the data are limited, they serve to emphasize the role that clay content has upon the soil freezing characteristic curve. However, any relationship could be affected by clay mineralogy (e.g., expanding clays) or extremes in soluble salt content.

Freezing characteristic data for Allendale silty clay were obtained for a variety of samples with dry bulk densities in the range 1.18 to 1.31 g cm⁻³ and at degrees of saturation from 66 to 96%. The test results are presented in Figure 8 (see also Appendix II). As shown, θ_{uf} at any given temperature falls within $\pm 2\%$ for all samples, suggesting that such ranges of density variations and degree of saturation have a minimal effect upon the freezing characteristic curve. Similar results were also found for Garden sandy loam (see Appendix II). The effect of soil bulk density will be discussed further in Section 3.3, where results for low density samples are presented.

3.2 Freezing Column Experiments

The test results from two experiments are shown in Figures 9 and 10, and Tables 2 and 3. The data for Ellwood clay loam show that the θ_v profile was essentially uniform at 0 hours but displayed marked desiccation in the unfrozen zone after 122 hours, implying water movement from the unfrozen to the frozen zone during freezing. Gravimetric sampling of the column at 122 hours indicated an increase in total water content in the frozen section. Table 2 shows that the calculated water balance (i.e., water loss from the unfrozen zone vs. water gain in the



Figure 7 Unfrozen Water Content Versus Clay Content



Figure 8 Freezing Characteristic Data for Allendale Silty Clay at Various Degrees of Saturation

18



19

1

TABLE 2

3

1

1

FREEZING COLUMN RESULTS (ELLWOOD CLAY LOAM)

WATER CONTENT DATA

Soil	L	θ (cm ³	cm^{-3} , %) by TDR	θ from gravimetric
(cm)		0 hr	122 hrs	122 hrs
0.5 -	3.0	42.5	Frozen	
3.0 -	5.5	43.7	Frozen	58.4 (0-5.0 cm)
5.5 -	8.0	43.7	26.8	27.2 (5.0-8.0 cm)
8.0 -	10.5	43.7	37.2	41.6
10.5 -	13.0	45.0	41.9	42.5
13.0 -	15.5	43.7	43.7	44.2
15.5 -	18.0	43.7	43.7	44.7
18.0 -	20.5	42.8	43.7	44.7

WATER BALANCE DATA

Volume of H_2^0 (cm³)

Soil Layer (cm)	(1) O hr TDR	(2) 122 hrs TDR	(3) 122 hrs gravimetre	(4) mean of (2) and (3)	Change in H ₀ Volume (4)- ² (1)
0- 5.0	290.5	Frozen	393.6	393.6	+ 103.1
5.0- 8.0	176.7	108.4	109.9	109.2	- 67.5
8.0-10.5	147.2	127.4	140.2	133.8	- 13.4
10.5-13.0	151.6	141.2	143.2	142.2	- 9.4
13.0-15.5	147.2	147.2	148.9	148.1	-
13.5-18.0	147.2	147.2	150.6	148.9	-
18.0-20.5	144.2	147.2	150.6	148.9	-
Total		Total			

 $H_2^0 = 1204.6$ $H_2^0 = 1237.0$

Volume of H_2^0 moved into frozen zone = 103 cm³ Volume of H_2^0 moved out of unfrozen zone = 90.3 cm³



ł

Results from Freezing Column Experiment: Cavan Loam

TABLE 3

FREEZING COLUMN RESULTS (CAVAN LOAM)

1

1

WATER CONTENT DATA

Soil Laver	θ (cm ³ cm ⁻	⁻³ , %) by TDR	θ from gravimetric
(cm)	0 hr	72 hrs	72 hrs
0.5 - 3.0	36.4	Frozen	
-3.0 - 5.5	33.0	Frozen	
5.5 - 8.0	33.0	Frozen	45.0 (0-7.5 cm)
8.0 - 10.5	32.0	27.2	Unable
10.5 - 13.0	31.8	27.4	to
13.0 - 15.5	31.8	28.5	sample
13.5 - 18.0	33.0	30.1	31.6
18.0 - 20.5	38.9	32.2	30.5

WATER BALANCE DATA

Soil Layer	Vol. H ₂ 0 at 0 hrs	Vol. H ₂ 0 at 72 hrs	Change in H ₂ 0 Volume
0 - 7.5	350.7	454.8	104.1
7.5 - 10.5	129.4	109.9	-19.5
10.5 - 13.0	107.1	92.3	-14.8
13.0 - 15.5	107.1	96.0	-11.1
15.5 - 18.0	111.2	101.4	- 9.8
18.0 - 20.5	131.1	108.5	-22.6

Volume of H_2^0 moved into frozen zone = 104.1 Volume of H_2^0 moved out of unfrozen zone = 77.8 frozen zone) is in good agreement. It should also be noted that the TDR and gravimetric measurements of θ_v at 122 hours in the unfrozen zone were in good agreement.

The Cavan loam results also show desiccation of the unfrozen zone and an increase in water content in the frozen portion. The calculated water balance agrees fairly well although the agreement is not as good as that for the Ellwood sample, presumably because of larger experimental errors in the gravimetric sampling.

Although the results are preliminary in nature, they seem to indicate that the TDR technique can be a useful tool for monitoring the desiccation of the unfrozen zone in closed system soil freezing. Further experimentation is required to substantiate these preliminary findings. Unfortunately, in neither of these first experiments can anything definite be said about the changes in K_a over time in the frozen zone. This is because, with a spacing of 2.5 cm, the field of the probe overlapped both the frozen and unfrozen zones. For example, in the Ellwood sample a marked decrease in K_a was measured by the probe just behind the frost front; however, it is not possible to say whether this is due to excess ice accumulation in the frozen zone (see Section 3.3), desiccation in the unfrozen zone, or a combination of both effects.

A new freezing column has closer-spaced probes (1.5 cm), and closer attention will be paid to ensure that the probe just behind the freezing front is completely within the frozen zone. Thus we can monitor the effect of ice accumulation on K_a , and then be able to assess the use of the TDR technique to determine areas of excess ice accumulation in the frozen zone and desiccation of the unfrozen zone.

3.3 Freezing Characteristic Data for Samples with Excess Ice

The effect of dry bulk density (ρ_b) on the soil freezing characteristic curve was examined for densities which are representative of possible field values (see Section 3.1). However, when a soil freezes and water migration occurs, zones of excess ice form, effectively decreasing ρ_b . Although we cannot use the TDR technique to measure ice content directly, increases in ice content, by ice lensing, should result in a decrease in the measured K_a due to the displacement of soil and unfrozen water. By examining freezing characteristic curves for low density samples (generally less than 1 g cm⁻³) we have tried to assess whether the TDR technique could be used to delineate areas of excess ice formation in freezing soils. If successful, it could be used <u>in situ</u> or in laboratory freezing soil columns.

A number of authors have suggested that the gravimetric unfrozen water content, w $(gg^{-1}, \%)$, should not be affected by the ice content of the soil (e.g., Williams, 1967; Anderson and Morgenstern, 1973). If so, then θ_{uf} is directly proportional to ρ_{b} as follows:

$$\theta_{uf} = \frac{w \cdot \rho_b}{\rho_a}$$
(1)

where ρ_b is the density of water (1 g cm⁻³). Thus, as ρ_b decreases, so does θ_{uf} , and so should the measured value of K. These relationships are illustrated, hypothetically, in Figures 11 and 12. Starting with an assumed freezing characteristic curve, on a gravimetric basis (w vs. T^oC), this was converted to a volumetric basis, for various bulk densities, using equation (1). The K_a scale for Figure 12 was derived from the relationship between K_a and Ouf tabulated in Appendix I.



à

1

1

Figure 11 Theoretical Freezing Characteristic Curves for Various Dry Bulk Densities



Figure 12 Theoretical Changes in K with Dry Bulk Density at Three Temperatures

Since θ_{uf} decreases as ρ_b decreases, it is expected that measured values of K_a for samples containing excess ice will be less than K_a for the same soil at normal density. i.e.,

- i) If K_a , measured at a certain temperature, is less than the K_a known for the same soil at normal density, then ρ_b is lower, for example, because of excess ice (see Figure 11).
- ii) If, at a certain temperature, K_a is observed to decrease over time, this indicates a decrease in ρ_b (Figure 12), for example, because of excess ice formation.

To examine these ideas, K_a vs. T^oC data were determined for Manchester silt, Castor sandy loam and Allendale silty clay at a variety of bulk densities (i.e., excess ice contents). The results have been converted into unfrozen water contents, using the relationship in Appendix I for θ_{uf} and equation (1) for w; they are shown in Figures 13 to 18. The experimental results confirm that:

- i) At a given temperature, θ_{uf} (K_a) decreases as ρ_b decreases (Figures 13, 15, and 17).
- At a given temperature, w is "constant" (within + 2%)
 regardless of the ice content (Figures 14, 16, and 18).

To take a specific example, we can look at the effect of excess ice content on K_a for Allendale silty clay (Figure 17). At a temperature of -0.7°C, we have the following data at various bulk densities:

$\rho_{\rm b} (g \text{ cm}^{-3})$	Ka	θ_{uf} (cm ³ cm ⁻³ , %)	$\theta_{i} (cm^{3} cm^{-3}, \%)^{*}$
1.186	9.88	19.6	34.2
0.825	7.84	15.8	43.7
0.683	6.76	13.4	56.6
0.545	5.14	9.4	57.2

*Ice content determined gravimetrically



Figure 13 Variation in Freezing Characteristic Curve with Bulk Density (Manchester Silt)

ġ

29



Figure 14 W-T^oC for Manchester Silt



l

ŧ.

곗

Figure 15 Variation in Freezing Characteristic Curve with Bulk Density (Castor Sandy Loam)



Figure 16 w - T ^oC for Castor Sandy Loam

.

2 👖





-



Figure 18 W-T^oC for Allendale Silty Clay

ł

ï

Thus there appears to be some evidence for a relationship between K_{a} and θ_{i} and in view of the above results it might be possible to infer an ice content from TDR measurement of K . However, since the TDR technique cannot measure ice content directly, the more term of "maximum possible ice content" is used in the following discussion. A nomogram has been prepared showing the relationship between maximum possible ice content, measured. θ_{uf} (K_a) and bulk density (Figure 19).¹ The nomogram assumes that w is constant (at any temperature) and that the soil contains no air. This results in a maximum possible ice content. Actual values may be lower than this if air is present (see below). The only information required are the freezing characteristic data at "normal" bulk density, and the measured ${\tt K}_{\tt a}$ $({\boldsymbol \theta}_{\tt uf})$ at some known temperature for the sample in question. For example, Figure 19 shows that a soil at a given temperature below 0°C has a θ_{uf} of 21% and a dry bulk density of 1.4 g cm⁻³; therefore w is 15% (Point A) and the maximum possible ice content is about 28.5% (Point B). If a sample later shows a $\theta_{\rm uf}$ of 10% (at the same temperature), $\rho_{\rm b}$ is about 0.63 g cm $^{-3}$ (Point C, assuming constant w) and the maximum possible ice content is about 65% (Point D). Applying the same logic to the example of the Allendale silty clay, we get the following results:

ŧ

ρ _b	θ _i	(Actual)	θ	(Nomogram)
1.186		34.2		35.8
0.825		43.7		55.2
0.683		56.6		66.2
0.545		57.2		70.0

¹ A full description of the nomogram is given in Appendix III.



Figure 19 Nomogram for Inferring Maximum Ice Content from Measured $\boldsymbol{\theta}_{uf}$

3

ŧ

Apart from the first sample (which contained virtually no air), there is some difference in the ice contents as a result of air in the samples. However, the inferred (nomogram) values do give some indication of the build-up of excess ice. Unfortunately, one cannot really improve on the nomogram since the TDR cannot resolve between air and ice in soil. Figure 20 (from Smith and Patterson, 1980) shows only a very slight change in K_a as ice replaces air in the soil, and certainly not enough of a difference to be exploited in the present context. Therefore we can only talk in terms of a maximum possible ice contect.

However, the potential exists for using TDR measurements of K_a to identify zones of excess ice. For example, ice content can be inferred if K_a changes over time whilst temperature does not change. Such conditions can occur in a freezing soil column where temperature can be readily controlled and a water source (from the unfrozen zone) is available. The usefulness of the TDR approach for identifying areas of ice accumulation will have to be investigated further via open system freezing soil column experiments using large diameter columns which can be readily sectioned to independently obtain water contents and density changes in the frozen section.

Finally, we have looked at whether the nature of the excess ice has any effect on K_a measurements. Data for Allendale silty clay, at a density of 0.825 g cm⁻³, were obtained for different samples containing dispersed ice and ice lenses (see Figure 17). As can be seen, the nature of the excess ice has no appreciable effect upon the overall value of K_a (and hence (θ_{uf}). However, it is interesting to note that discrete ice lenses in a coaxial line were readily detected on the TDR trace (Figure 21). Experiments are currently in progress to examine the possibility of using



I

Figure 20 K for Frozen Ottawa Sand at Different Initial Water Contents



ł

Į



parallel transmission lines to identify zones of excess ice accumulation in freezing soils. The ability to identify ice lense growth would have many practical applications, such as round refrigerated pipelines or beneath highways.

K

SECTION 4

CONCLUSIONS

The TDR technique now shows great promise for routine measurement of volumetric liquid water contents in frozen soils as suggested by Smith and Patterson (1980). The extension of the TDR work to samples containing excess ice and to instrumenting soil freezing columns suggests that the TDR technique may prove to be a valuable nondestructive laboratory technique for obtaining hydrologic data for soil freezing experiments.

Specific findings from the experiments reported here include:

- 1. Values of θ_{uf} determined via the TDR technique and suction-moisture tests agree to within $\pm 2.5\%$ in θ_{uf} .
- 2. Freezing characteristic data obtained from samples of various field densities and degrees of saturation agree well ($\frac{+}{-}$ 2% in θ_{uf}).
- 3. A relationship between clay content and θ_{uf} , at various temperatures was found for soils tested via the TDR technique.
- The unfrozen water content, expressed gravimetrically, for a soil at a given temperature is within ⁺/₋ 2% regardless of ice content.
- 5. K_a , at a given temperature, is less in samples containing excess ice (dry bulk densities less than 1 g cm⁻³) than for samples without excess ice. A method has been devised for inferring a maximum possible ice content by comparing a measured value of K_a (θ_{uf}) with that expected for the same soil at normal field density (i.e., from the normal freezing characteristic curve).

6. Preliminary freezing column experiments indicate that the TDR technique may be a viable method of determining the rates and amount of desiccation in the unfrozen zone during closed system soil freezing.

Continued research should be undertaken to examine the θ_{uf} -T °C relationship for soil samples containing excess ice since these data will help to assess the use of the TDR technique to delineate zones of ice lens growth in freezing soils columns or perhaps even <u>in situ</u>. Experiments are currently being undertaken to examine the potential use of balanced parallel transmission lines to locate zones of excess ice in soil profiles. Continued freezing soil column experimentation is needed to examine the use of the TDR technique as a non-destructive method of providing data on heat and mass transfer in freezing soils. Field sites should also be instrumented to assess the use of the TDR technique for monitoring water movement and ice segregation in natural freezing ground.

ACKNOWLEDGEMENTS

Ē

1

This research has benefited from the continued interest and support of Dr. A.S. Judge (Earth Physics Branch, Department of Energy, Mines and Resources). Thanks also to C.H. Lewis (laboratory assistant), A. Pendlington and L. Boyle of the Geotechnical Science Laboratories, Carleton University.

References

Anderson, D.M. and N.R. Morgenstern. 1973. "Physics, chemistry and mechanics of frozen ground", <u>Permafrost 2nd International</u> Conference, Yakutsk, U.S.S.R., pp. 257-288.

and A.R. Tice. 1972. "Predicting unfrozen water contents in frozen soils from surface area measurements", Highway Research Record, No. 393, pp. 12-18.

- Dillon, H.B. and O.B. Andersland. 1966. "Predicting unfrozen water contents in frozen soils", <u>Canadian Geotechnical Journal</u>, Vol. 3, pp. 53-60.
- Mageau, D.W. and N.R. Morgenstern. 1980. "Observations on moisture migration in frozen soils", <u>Canadian Geotechnical Journal</u>, Vol. 17, pp. 54-60.
- Smith, M.W. and D.E. Patterson. 1980. <u>Investigation of frozen soils</u> using time domain reflectometry. Final Report to the Department of Energy, Mines and Resources, Ottawa, p. 59.
- Stalker, R.W. (in prep.). Study of water content variations in a natural field site, M.A. thesis, Geography Department, Carleton University.
- Topp, G.C., J.L. Davis and A.P. Annan. 1980. "Electromagnetic determination of soil water content: measurements in coaxial transmission lines", <u>Water Resources Research</u>, Vol. 16, No. 3, pp. 574-582.
- Torrance, J.K. and P.J. Williams. 1976. <u>Research into water migration</u> <u>phenomena in freezing soil columns</u>. Final Report to the Department of the Environment, Hydrology Branch, Inland Waters Directorate, p. 43.
- Williams, P.J. 1967. Properties and behaviour of freezing soils. Norwegian Geotechnical Institute, Oslo, Publication No. 72, p. 119.

1975. Determination of unfrozen water content temperature relationships for ten soil samples. Geotechnical Science Laboratories, Department of Geography, Carleton University, Ottawa, Internal Report IR-11.

Appendix I

		GIVEN BY TOPP, DAVIS	S AND ANNAN	(1980)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Ka	Vol. Water Cont. (decimal fraction)	Ka	Vol. Water Cont. (decimal fraction)
10.12 0.200 27.47 0.425 10.41 0.205 27.93 0.430 10.71 0.210 28.39 0.435 11.02 0.215 28.85 0.440	3.03 3.08 3.14 3.20 3.27 3.35 3.44 3.53 3.74 3.63 3.74 3.857 4.237 4.237 4.237 4.527 4.527 3.897 4.237 4.527 5.535 5.721 6.324 6.768 7.45 7.69 7.948 8.728 9.269 9.549 9.5432	(decimal fraction) 0.000 0.005 0.010 0.015 0.020 0.025 0.030 0.035 0.040 0.045 0.055 0.060 0.065 0.070 0.075 0.080 0.085 0.090 0.095 0.100 0.105 0.110 0.115 0.120 0.125 0.130 0.135 0.140 0.145 0.155 0.150 0.155 0.160 0.175 0.180 0.190 0.195	11.64 11.96 12.28 12.61 12.94 13.28 13.62 13.97 14.32 14.67 15.03 15.40 15.76 16.13 16.51 16.89 17.27 17.66 18.05 18.44 19.24 19.65 20.05 20.47 20.88 21.30 21.72 22.15 22.57 23.00 23.44 23.87 24.31 24.76 25.20 25.65 26.10 26.55 27.01	(decimal fraction) 0.225 0.230 0.235 0.240 0.245 0.250 0.265 0.260 0.265 0.270 0.275 0.280 0.285 0.290 0.295 0.300 0.305 0.310 0.315 0.320 0.325 0.320 0.325 0.330 0.335 0.340 0.345 0.355 0.360 0.365 0.370 0.375 0.380 0.385 0.390 0.395 0.400 0.415 0.420
1 1 1 2 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	10.41 10.71 11.02	0.205 0.210 0.215	27.93 28.39 28.85	0.430 0.435 0.440

TABULATION OF THE EMPIRICAL RELATIONSHIP GIVEN BY TOPP, DAVIS AND ANNAN (1980)

1

1

Appendix I Continued

Ka	Vol. Water Cont. (decimal fraction)	Ka	Vol. Water Cont. (decimal fraction)
29.79	0.450	32.65	0.480
30.26	0.455	33.13	0.485
30.74	0.460	33.62	0.490
31.21	0.465	34.10	0.495
31.69	0.470	34.59	0.500
32.17	0.475	35.08	0.505

APPENDIX II

FREEZING CHARACTERISTIC DATA

 $\begin{array}{l} \theta_{uf} \text{ is the volumetric water content (cm^3 cm^{-3}, %).} \\ \text{K}_a \text{ is the apparent dielectric constant.} \\ \rho_b \text{ is the dry bulk density in g cm}^{-3}. \\ \overline{\theta}_{uf} \text{ is the mean } \theta_{uf}. \end{array}$

CASTOR SANDY LOAM (freeze cycle)

San	ple (1)		(2)	(3	3)	
°c	Ka	θ_{uf}	Ka	$\theta_{\tt uf}$	Ka	$\theta_{\tt uf}$	$\overline{\theta}_{\mathtt{uf}}$
0 -0.20 -0.52 -0.70 -1.20 -1.85 -2.50 -3.30	26.0 8.31 5.84 5.76 5.60 5.29 4.69 4.41	41.0 16.8 11.4 11.1 10.7 9.8 8.0 7.2	27.6 9.40 6.25 5.60 5.29 4.91 4.55 4.55	42.6 18.8 12.4 10.7 9.8 8.8 7.6 7.6	24.3 9.00 5.84 5.68 5.29 5.21 4.91 4.91	39.0 18.0 11.4 10.9 9.8 9.6 8.8 8.8	40.9 17.9 11.7 10.9 10.1 9.4 8.1 7.9
From gr	avimetr	ic samp	ling:			- • •	,

РЪ	1.56	1.54	1.54
θ at $0^{\circ}C$. 39.2	39.2	39.2

CASTOR SANDY LOAM <74µ (freeze cycle)

Sar	nple (1)		(2)	(2	3)	
°c	Ka	$\theta_{\tt uf}$	Ka	$\theta_{\tt uf}$	Ka	$\theta_{\tt uf}$	$\bar{\theta}_{\mathtt{uf}}$
0 -0.10	25.0	39.8 33.3	25.1 20.9	39.9 35.0	25.3	40.2	40.0 34.4
-0.31 -0.46	10.9 7.47	21.2 15.0	11.1 8.02	21.6	12.5	23.8	22.2 15.7
-0.80 -1.60	5.44 4.69	10.3 8.1	6.08 5.44	11.7 10.2	5.60 4.98	10.7 9.0	10.9 9.1
-2.45	4.55 4.20	7.5 6.4	5.14 4.62	9.4 7.8	4.48 4.34	7.3 6.8	8.1 7.0

1

3

2

i

Ę.

ρ _b	1.62	1.58	1.65
0°C	38.4	39.2	39.0

FIELD LOAMY SAND (freeze cycle)

°c	Ka	$\theta_{\tt uf}$
0	17.1	30.2
-0.30	6.25	12.4
-0.78	5.60	10.6
-2.88	4.84	8.6
-3.94	4.55	7.6

From gravimetric sampling:

ρ		1.72
θat	0°	30.4

ANGRI SANDY LOAM (freeze cycle)

°c	Ka	$\theta_{\tt uf}$
0	23.5	38.1
-0.08	12.9	24.5
-0.30	9.20	18.4
-0.55	8.22	16.6
-0.70	7.47	15.0
-1.03	5.84	11.3
-1.47	5.76	11.1
-2.03	5.60	10.7
-2.55	5.44	10.2

From gravimetric sampling:

ρ		1.40
θat	0°	41.5

GARDEN SANDY LOAM (freeze cycle)

	(1)		(2)	(3)*	
°c	ĸ	$\theta_{\tt uf}$	Ka	θuf	Ka	$\theta_{\tt uf}$	$\overline{\theta}_{uf}$
0 -0.12 -0.25 -0.38 -0.60 -1.10 -1.92 -2.91	19.4 10.5 8.0 6.93 6.41 5.92 5.29 4.69	33.2 20.6 16.2 13.9 12.7 11.6 9.8 8.1	29.2 13.9 8.9 7.29 6.67 6.16 5.68 5.24	41.1 25.9 17.8 14.6 13.3 12.1 10.8 9.7	19.7 10.4 8.1 7.29 7.11 6.25 5.68 5.29	33.5 20.4 16.4 14.6 14.3 12.3 10.8 9.8	35.9 22.3 16.8 14.4 13.4 12 10.5 9.2
From gra	avimetri	c sampli	ing:				
b at 0°(C	1.42 32.1		1.30 42.2		1.51 33.9	

* undisturbed sample

1

in the second

1

ŧ.

1

1

ALLENDALE SILTY CLAY (thaw cycle)

(1)		(2))
ĸa	$\theta_{\texttt{uf}}$	Ka	$\theta_{\texttt{uf}}$
16.0 10.7 10.2 9.82 9.20 8.80 8.03 7.47	28.8 20.9 20.2 19.6 18.4 17.6 16.2 15.0	13.4 10.8 9.82 9.61 9.00 8.80 8.41 7.84	25.2 21.1 19.6 19.2 18.0 17.6 16.9 15.8
7.29	14.6	7.84	15.8
	(1) K _a 16.0 10.7 10.2 9.82 9.20 8.80 8.03 7.47 7.29	 (1) K_a θ_{uf} 16.0 28.8 10.7 20.9 10.2 20.2 9.82 19.6 9.20 18.4 8.80 17.6 8.03 16.2 7.47 15.0 7.29 14.6 	(1) (2) $K_a = \theta_{uf} = K_a$ 16.0 28.8 13.4 10.7 20.9 10.8 10.2 20.2 9.82 9.82 19.6 9.61 9.20 18.4 9.00 8.80 17.6 8.80 8.03 16.2 8.41 7.47 15.0 7.84 7.29 14.6 7.84

From gravimetric sampling:

р _р	1.18	1.19
θ [¯] at 0 [°] C	53.8	53.2
% sat (Vw/Vv)	96.4	95.8

	(3)		(4)	((5)
°c	Ka	$\theta_{\tt uf}$	K _a	$\theta_{\tt uf}$	Ka	$\theta_{\tt uf}$
-0.07 -0.25 -0.45 -0.63 -0.90 -1.08 -1.55 -2.08 -3.27	15.7 12.1 11.1 10.7 9.61 9.20 8.41 8.22 7.75	28.4 23.2 21.6 20.9 19.1 18.4 16.9 16.6 15.6	15.7 13.4 11.6 10.9 10.0 9.40 9.00 8.03 7.38	28.4 25.2 22.4 21.3 19.8 18.8 18.0 16.2 14.8	15.5 11.5 10.0 9.20 8.41 8.03 7.47 6.93 6.59	28.1 22.2 19.8 18.4 17.0 16.2 15.0 14.0 13.1
From gravimetric sampling:						
Ръ	_	1.23		1.31		1.22

AGRI SILTY CLAY (thaw cycle)

33.9

66.3

38.4

70.5

°c	Ka	$\theta_{\tt uf}$
0 -0.08 -0.30 -0.55 -0.70 -1.03 -1.47 -2.03 -2.55	31.9 15.3 10.5 9.61 9.61 9.40 9.00 8.80 8.22	47.2 27.9 20.6 19.1 19.1 18.8 18.0 17.6 16.6

Fram gravimetric sampling:

 θ at 0°C

% sat (Vw/Vv) 87.4

47.3

2

1

ŧ.

1

¢

ρ _ł	5		1.03
θ	at	0°c	50.3

°C	Ka	$\theta_{\tt uf}$
0 -0.08 -0.30 -0.55 -0.70 -1.03 -1.47 -2.03	35.0 15.2 11.6 9.92 9.10 8.80 8.03 7.47	50.4 27.8 22.4 19.8 18.2 17.7 16.2 15.0
-2.53	1.20	7.4.4

From gravimetric sampling:

I

12

ρ			1.20
0	at	0°	48.8

LE7 CLAY (thaw cycle)

°c	Ka	$\theta_{\tt uf}$
0	34.8	50.2
-0.08	14.7	27.0
-0.30	12.0	24.0
-0.70	11.1	21.6
-1.03	10.4	20.4
-1.47	9.71	19.3
-2.03	9.10 8.60	18.2

From gravimetric sampling:

ρ)		1.05
θ	at	00	51.3

°c	Ka	$\theta_{\tt uf}$
-0.20	15.07	27.6
-0.35	12.85	24.4
-0.77	10.61	20.8
-1.30	9.45	18.8
-2.12	8.67	17.40
-3.15	7.77	15.6

2

3

1

á

â

--

- and

1

1

1

1

1

LE2 and LE7 from suction moisture (Williams 1975)

LE7

LE2

°c	w (gg-1, %)	$\theta_{\tt uf}^{\star}$	₩ (gg-1,	%) θ [*] uf
-0.1 -0.2 -0.3 -0.4 -0.5	22.5 18.8 17.5 16.3 15.0	27.0 22.5 21.0 19.5 18.0	29.0 27.0 24.5 23.0 22.0	30.5 28.4 25.7 24.2 23.1
-0.9	13.0	15.6	19.0	20.0

* θ_{uf} = w x ρ_{b} where ρ_{b} = 1.2 for LE2 and 1.05 for LE7 w is best fit

LOEB CLAY LOAM (thaw cycle)

°c	Ka	$\theta_{\tt uf}$
0	31.2	46.6
-0.25	13.7	25.6
-0.45	12.0	23.1
-0.82	10.4	20.4
-1.25	9.30	18.4
-2.82	. 8.22	16.5

From gravimetric sampling:

ρ _b)		1.1	9
θ	at	0°c	. 47.5	,

APPENDIX III

As indicated in section 3.3, a maximum possible ice content for a soil can be inferred if temperature and the freezing characteristic for the soil at normal dry bulk density are known. A nomogram (Figure 19) shows the relationships between ice content, θ_{uf} and ρ_b . The program listed in this Appendix can also be used to determine ice content from a measured θ_{uf} (i.e., Ka); the required input information at a given temperature (T₁) is:

1. dry bulk density for the normal soil;

2. particle density;

3. θ_{uf} at T₁ for the normal soil

The program then calculates the volume of solids, ice and water for the normal soil at T_1 . By inputing θ_{uf} at T_1 for a sample with excess ice, the program will calculate the dry bulk density and ice content. It should be emphasized that the ice content determined via the nomogram or the program is the 'maximum possible' value and not an actual measured value.

A program listing (in BASIC) and a sample output follows.

M1= MASS OF SOLIDS M2= MASS OF WATER B1= BULK DENSITY V1= TOTAL VOLUME V2= VOLUME OF SOLIDS P1= PARTICLE DENSITY V3= VOLUME OF VOIDS V9= VOLUME OF H2O MAXIMUM ASSUMING %SAT OF 95% V8= VOLUME OF AIR T1= SAMPLE TEMPERATURE U1= UNFRCZEN VOL. H2O CONTENT FOR THE NATURAL SOIL V4= VOLUME OF H2O BASED ON U1 V4= VOLUME OF H2O BASED ON U1 V5= VOLUME OF ICE BASED ON U1 U2= VOL. ICE CONTENT BASED ON U1 W1= GRAV. UNMERCIAN DE CONTENT BASED ON U1

B2= UNKNOWN BULK DENSITY U9= MEASURED VOL. H20 CONTENT FOR UNKNOWN SOIL I.E. EXCESS ICE M9= MASS OF SOLIDS BASED ON B2 M8= MASS OF H20 BASED ON B2 V6= VOLUME OF SOLIDS BASED ON B2 V7= VOLUME OF ICE BASED ON B2 U8= V0L. ICE CONTENT BASED ON B2

THESE ARE THE SOIL CONDITIONS FOR THE SOIL FROM THE ØUF-TEMP RELATIONSHIP

1

BUL	K IENC.	1.3000	4	TOTAL=	1.0000
PAR		2.7000	V.	SOLIDS=	0.4815
11	SOLIDES	1.3000	1,1 1,1 1,1 1,1	H20 MAX=	0.4926
l i	VOIDS=	0.5185	i i n	HIR=	0.0259

THESE APE THE CONDITIONS FOR OUF FROM FREEZING CHARACTERISTIC

	1000 1000 1000 1000 1000	-0.1000	0UF=	0.2000
ti a		0.2000	₩. ICE=	0.3191
	ICE	CONT= 0.319	91 GRAV H2O CONT=	0.1538

 THE FOLLOWING VALUES ARE FOR THIS SAMPLE

 OUF - 0.1000

 BULK DENS.= 0.6500
 MASS OF SOLIDS= 0.6500

 MASS OF H20= 0.1000
 VOL. OF SOLIDS= 0.2407

 VOL. OF ICE= 0.6333
 ICE CONTENT= 0.6333

	OLLOWING	; VALUES	HRE	FOR	THIS	SAMPLE	
	0.0500						
		0.3250		MASS	ŨF	SOLIDS=	0.3250
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	06 420=	0.0500		VOL.	DF	SOLIDS=	0.1204
VOL.	OF 1CE-	0.8037		10 100 1000	CONT		0.8037

PRINT HLUES OF THE VOLUMETRIC UNFROZEN WATER CONTENT, OBTAINED FROM " PRINT THE TDR TECHNIQUE. THERE ARE SOME ASSUMPTIONS WHICH MUST BE MADE" SLEAR! THEY ARE:" PRINT 1. TOTAL SAMPLE VOLUME IS FIXED" 2. THE % SATURATION (95%) IS FIXED (APPROX FIELD CAPACITY" PRINT PRIMT PRINT THE ICE CONTENTS CALCULATED WILL REPRESENT PROBABLE MAXIMUM VALUES" PRINT 0 PRINT "M'= MASS OF SOLIDS" PRINT "M2= MASS OF WATER" Ø PRIN Ē 81= BULK DENSITY" 'V1= TOTAL VOLUME" 诏 PRINT '√2≕ VOLUME OF SOLIDS" 0 PRINT 'PI= PARTICLE DENSITY" 0 PRINT PRINT "V3= VOLUME OF VOIDS" 10 1/9= VOLUME OF H20 MAXIMUM ASSUMING %SAT OF 95%" 0 PRIN: 0 PRIM VS= VOLUME OF AIR" 0 PRIN Ū. PRINT TI= SAMPLE TEMPERATURE" 0 PRINT D.⇒ UNFROZEN VOL. H20 CONTENT FOR THE NATURAL SOIL" "V4= VOLUME OF H20 BASED ON U1" Ū. PRINT 10 PRINT /5= VOLUME OF ICE BASED ON U1" Ē PRINT 192= VOL. ICE CONTENT BASED ON U1" PRIMI W:= GRAV. UNFROZEN H20 CONTENT BASED ON U1" 0 9 PRINT PRINT "82= UNKNOWN BULK DENSITY" Ø Ø PRINT U9= MEASURED VOL. H20 CONTENT FOR UNKNOWN SOIL I.E. EXCESS ICE" PRIN. Ū. ™9= MASS OF SOLIDS BASED ON B2" 'M8= MASS OF H20 BASED ON B2" Ø PRINT V6= VOLUME OF SOLIDS BASED ON B2" Ø PRINT PRINT "V7= VOLUME OF ICE BASED ON B2" Ø PRINT Ø 'US= VOL. ICE CONTENT BASED ON B2" Ū. PRINT FIXED 4 0 0 DISP 'INPUT B DENS, VT, P DENS" 5 1 WAIT 2000 INPUT 31, VI.P1 Ø 0 M1=810-91 V2=M1/P1 Ø V3=V: 72 Ø Ø N9=V3.8.93 iA. V8=V3-V9 0 PRINT THESE ARE THE SOIL CONDITIONS FOR THE " SOIL FROM THE OUF-TEMP RELATIONSHIP" 0 PRINT 0 PRINT "BULK DENS.= ",B1,TAB30"V. TOTAL= ",V1 0 PRINT PART, DENSITY= ",P1,TAB30"V. SOLIDS= ",V2 0 PRINT M. SOLIDS= ",M1,TAB30"V. H20 MAX= ",V9 LA POINT L

3

```
LNPU: LISUI
   V4=U1-71
   V5=(v9-V4)*1.0905
   U2=V5 V1
   W1=V4 M1
   PRINT TAB5"THESE ARE THE CONDITIONS FOR OUF FROM FREEZING CHARACTERISTI
 · PRINT
 PRIMT THESTTEMP= ",T1,TAB30"0UF= ",U1
 1 PRINT TAB5"V. H20= ",V4,TAB30"V. ICE= ",V5
 I PRINT TAB5"VOL ICE CONT= "U2, TAB30"GRAV H20 CONT= ", W1
 I DISP GNTER ØUF MEASURED"
 I WAIT 1500
 I INPU, U9
 ) B2=U9x01
 ) M9=82+V1
 ) M8=M9*W1
 ) V6=M9/P1
 ) V7=V)--(M8+V8+V6)
 ) U8=V7/V1
 3 PRINT
 3 PRINT TABS"THE FOLLOWING VALUES ARE FOR THIS SAMPLE"
 3 PRINT LARS"OUF= "U9
 3 PRINT TAB5"BULK DENS. = "B2, TAB30"MASS OF SOLIDS= "M9
 3 PRINT THESTMASS OF H20= "M8, TAB30"VOL. OF SOLIDS= "V6
 3 PRINT 1985"VOL. OF ICE= "V7, TAB30"ICE CONTENT=
                                                        "118
 3 DISP 'MORE POINTS? YES=1 NO=2"
 3 WAIT 1500
 a INPUT I
 3 IF I=1 THEN 930
 3 DISP 'SOIL TYPE THE SAME? YES=1 NO=2"
-3 WAIT 1500
 3 INPUT J
 3 IF J=: THEN 880
  IF J=2 THEN 360
 9
 0 DISP 'IS TEMP THE SAME YES=1 NO=2"
 0 WAIT 1500
0 INPUT K
 0 IF K=: THEN 640
 0 IF K=2 THEN 520
. A FNT
```