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MINES BRANCH INVESTIGATION REPORT IR 58-149

COMPILATION OF TEST RESULTS AND STATISTICAL DATA ON PHASE 1 OF GALVANIZING RESEARCH PROJECT ZN-7 (Supplementary to Research Reports R5 and R6)

by

J. J. SEBISTY

PHYSICAL METALLURGY DIVISION

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SUMMARY

For record purposes, the results of work carried out on phase I of the galvanizing research project have been collected in the appendices which make up this report. This covers test work at the Mines Branch, Ottawa, accelerated corrosion tests at the Steel Company of Canada, Limited, Hamilton, Ontario, and statistical studies at the Consolidated Mining and Smelting Company of Canada, Limited, Trail, British Columbia.

With the exception of the statistical study given in appendix IV, the data tabulated have been fully discussed in reports previously issued (Research Reports R 5 and R 6) and in the corresponding papers prepared for presentation at the Fifth International Galvanizing Conference at Brussels, Belgium, June 1958.

Senior Scientific Officer, Physical Metallurgy Division, Mines Branch, Department of Mines and Technical Surveys, Ottawa, Canada.

- i -

CONTENTS

1

	Page
Summary	i
<u>Appendix I</u> - Data on Main Series of Tests in Phase I of Galvanizing Project Zn-7	1
Introduction. Tables 1 to 5	
Appendix II - Accelerated Corrosion Testing of Galvanized Panels	7
Introduction. Corrosion Tests, Corrosion Evaluation. Experimental Procedure. Tables 1 to 4. Recommendations.	
Appendix III(a) - Preliminary Statistical Analysis of Coating Test Data	17
Introduction. Headings Y1 to Y 12.	
<u>Appendix III(b)</u> - Preliminary Statistical Analysis of Accelerated Corrosion Test Data.	27
Introduction. Headings Y13 to Y18.	
<u>Appendix IV</u> - Canonical Analysis of Coating Test and Accelerated Corrosion Test Data	32
Introduction. Headings Y ₁ to Y ₁₈ . Figures 1 to 18.	

(62 pages, 9 tables, 18 illus.)

.

<u>APPENDIX I</u>

- 1 -

DATA ON MAIN SERIES OF TESTS IN PHASE I OF GALVANIZING PROJECT ZN-7

by

J. J. Sebisty

June 25, 1958

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INT'RODUCTION

In this appendix, data pertaining to the work carried out at the Mines Branch on phase I of the galvanizing research project are given in tabular form. In order to save space, only typical melt and galvanizing logs have been included and the coating test results have been averaged. For illustrative purposes, complete coating test results are given for two typical series of specimens. The various tables included are identified as follows:

- Table 1. Typical galvanizing melt log.
- Table 2. Typical galvanizing log.
- Table 3. Coating test results for typical series of specimens.
- Table 3 (a). Surface appearance rating codes.
- Table 4. Average coating test results.
- Table 5.Combined coating test results. (These were
derived from the findings of the preliminary
statistical study given in appendix III(a).
The values listed were used for graphical
presentation of the coating test results in
the paper and research report covering this
part of the work).

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TABLE 1 .

	Typi	lcal Galva	nizi	ng	<u>Melt I</u>	Log
MINES BRANCH PHYSICAL METALLURGY		NON-FERR GALVANIZ		Project Zn-7		
DIVISION		GUILV MULZ			Date: Nov. 21/57	
Melt No. CU (Bath No	<u>. 1)</u>				• .	
<u>Charge</u> 39.6 1b	· ·			•		
Metal.	Compos	sition			Fo	orm Amount
Zn Pb Fe-Zn master Al-Zn master	Pb 99 Fe-Zn master 0					ngot 34.35 1b heet 88.5 g hot 3.2 1b " 1.86 1b
	**** *********************************					
Procedure	1	limo			mp	Remarks
Furnace on Zinc charged Alloying	8	\$.45 a.m. \$.45 " .05 "		50	0°0	
Pb Fe-Zn Al-Zn extra Fe-Zn (130 g) extra Al-Zn (50 g)	8.45 " 11.05 " 11.15 " 1.30 p.m. 1.35 "			0°0 0°0 0°0 5°0) added after galvenizing of) each lot of 12 large specimens	
<u>Poured</u> to ingot after galvanizing run	4.30 p.m.			0°C		
			, <u>mene (mes</u>			A Traduct - Thanks - Mill Fried and a second se
Bath composition	· .	Fe %	Als	6	Pb 9	6
Nominal Actual	0.03	0.2	2 0.5			
1.35 p.m. Start of di 2.55 " 2nd sample 4.15 " End of dipp	0.032 0		9	0.50 0.50 0.49	0 (taken after 24 large specimens	
		, 				

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MINES BRANG PHYSICAL MI DIVISION							t Zn-7 Nov. 21/57		
Melt No.	<u>cu (</u>	Bath No. 1)	•						
Test No.	1, 2	, 3, 4.	48 specimens,				5 & No. 3 steel)		
Pickling Sample No.	T	Acid Conc	Inhibitor		<u>No. 5&</u> Temp	NO. 2	<u>steer</u> Rinse		
bumpito no.			TIMADAOOL		r romb		1(11)00		
ЛЦІ.	.5%	H ₂ 504 sol'n.	1/2% by volume of acid (Rodine 92)	5 min	at 71°C	rinse in co	bbed and then ed for 1 min old running r. Dried in one.		
Fluxing				, , , , , , , , , , , , , , , , , , ,					
Sample No.	[Flux	Density	Time	& Temp	Dryin	ng Time & Temp		
ALL.	Ammo	c chloride - onium chloride 27:1.35 ratio flux)	10.4° Baumé	l min	at 82°C		to 2 min 60 to 170°C		
Gelvenizing	a		Immersion	Twww.ow.cd.ou	TT# 1. 5	7	,		
COTABILITY TUB	<u> Б</u>								
Sample No.		Bath Temper	Speed	Immersion	Withdra Speed		Remarks		
Sample No. Large Speci	imens	Bath Tempod	Speed	Time	Speed		Romarks		
		466, 466, 465	Speed				Romarks No. 5 finish		
Large Speci	6	466, 466, 465 465, 465, 465	Speed 6 fpm	Time	Speed				
Large Spect	6 18	466, 466, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465	Speed . 6 fpm "	Time 30 sec	Speed 3 fpm		No. 5 finish		
Large Speci 1-1 to 1-0 1-13 to 1-1	6 18 6 [:]	466, 466, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465	Speed . 6 fpm " "	Time 30 sec II	<u>Speed</u> 3 fpm "		No. 5 finish No. 3 finish		
Large Spec: 1-1 to 1-0 1-13 to 1-3 2-1 to 2-0	6 18 6 ⁻ 18	466, 466, 465 465, 465, 465	Speed 6 fpm " " "	Time 30 sec u 2 min	<u>Speed</u> 3 fpm "		No. 5 finish No. 3 finish No. 5 finish		
Large Spec: 1-1 to 1-0 1-13 to 1-1 2-1 to 2-0 2-13 to 2-1	6 18 6 [:] 18 6	466, 466, 465 465, 465, 465 445, 445, 445	Speed 6 fpm 11 11 11 11 11	Time 30 sec u 2 min u	Speed 3 fpm " "		No. 5 finish No. 3 finish No. 5 finish No. 3 finish		
Large Spec: 1-1 to 1-(1-13 to 1-: 2-1 to 2-(2-13 to 2-: 3-1 to 3-(6 18 6 18 6 18	466, 466, 465 465, 465, 465 444, 444, 445 445, 445, 445	Speed 6 fpm 11 11 11 11 11 11 11	Time 30 sec u 2 min u 30 sec	Speed 3 fpm " " "		No. 5 finish No. 3 finish No. 5 finish No. 3 finish No. 5 finish		
Large Spec: 1-1 to 1-(1-13 to 1-: 2-1 to 2-(2-13 to 2-: 3-1 to 3-(3-13 to 3-:	6 18 6 18 6 18 6	466, 466, 465 465, 465, 465 445, 445, 445 445, 445, 445	Speed 6 fpm 11 11 11 11 11 11 11 11 11 11 11 11 11	Time 30 sec u 2 min u 30 sec u	Speed 3 fpm " " " " "		No. 5 finish No. 3 finish No. 5 finish No. 3 finish No. 5 finish No. 3 finish		
Large Spec: 1-1 to 1-6 1-13 to 1-7 2-1 to 2-6 2-13 to 2-7 3-1 to 3-6 3-13 to 3-7 4-1 to 4-6 4-13 to 4-7 Small Spec	6 18 6 18 6 18 6 18 18	466, 466, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 445, 445, 445 445, 445, 445 445, 445,	Speed 6 fpm 11 11 11 11 11 11 11 11 11 11 11 11 11	Time 30 sec u 2 min u 30 sec u 2 min u u	Speed 3 fpm " " " " " " " " " "		No. 5 finish No. 3 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish		
Large Spec: 1-1 to 1-6 1-13 to 1-7 2-1 to 2-6 2-13 to 2-7 3-1 to 3-6 3-13 to 3-7 4-1 to 4-6 4-13 to 4-7 Small Spec	6 18 6 18 6 18 6 18 18 18 19 000 of	466, 466, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 445, 445, 445 445, 445, 445 445, 445,	6 fpm " " " " " " " ght Loss Measur pairs prior to Manual-approx.	Time 30 sec u 2 min u 30 sec u 2 min u 2 min u ench ser:	Speed 3 fpm " " " " " " " " " "	large	No. 5 finish No. 3 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish specimens) 3 each of No. 5 & No. 3		
Large Spec: 1-1 to 1-0 1-13 to 1-1 2-1 to 2-0 2-13 to 2-1 3-1 to 3-0 3-13 to 3-1 4-1 to 4-0 4-13 to 4-1 <u>Small Spec</u> (each ground	6 18 6 18 6 18 6 18 18 18 18 18 30 30	466, 466, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 465, 465, 465 444, 444, 445 445, 445, 445 445, 445,	6 fpm 4 fpm 1 1 1 1 1 1 1 1 1 1 1 1 1	Time 30 sec u 2 min u 30 sec u 2 min u 2 min u ench ser:	Speed 3 fpm " " " " " " " " " " " " " " " " " " "	large	No. 5 finish No. 3 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish No. 5 finish No. 3 finish		

Test Number	Steel Finish	Coating Wt, oz/sq ft-sheet	Iron Content	Steel Wt	Results for Alloy Thickness, MM	% Alloy	Ductility **	Adherence **	Spangle Size		Brightness	Roughnes:
1	5 5	0.44 0.46	180 2.0 210 2.3	4.9 3.7	*	·	1	1	4	4	2 3	4
1 1 1	5 3 3 3	0.52 0.43 0.44 0.43	1701.82102.32102.32402.6	4•5 4•4 5•6	* * * * *	- - - -	1 1 1	1 1 1	4 4 4	4 4 4 4	3 2 2 2	4 4 4
5 5 5	5 5 5	1.32 1.38 1.94	2460 26.4 2500 26.9 2510 27.1	28.4 24.0 23.5	0.0240 0.0257 0.0236	59.1 59.4 59.5	4 4 4	5.5 5.5 5.5	4 4 4	4 4 4	3 4 4	3 3 3
5 5 5	3 3 3	1.66 1.67 1.67	2380 25.7 2440 26.3 2410 26.0	26.1 27.6 31.9	0.0236 0.0222 0.0217	60.0 58.5 60.3	3 3 3 3	5.5 5.5 5.5	222	2 2 2	222	222
	· · ·	*Impossible	to measure or esti	mate.	***See	codes in	Table 3(a).			<u>`</u>		······································
		•			TABLE	<u>3 (a)</u>						
Ductili	+ 12			Suri Spangle	ace Appearance	e Rating C	odes	Brightn	- (Phot	cometer res	dincel	
	ng: - 1. 2. 3. 4.	Poor, wide separa size blocks	fine cracks cking, with p into small block	Ratin	ng: - 1. Larg 2. Medi 3. Smal 4. No s	un 1			ng: - 1. 2. J 3.			
		in large blocks						······			· · ·	
<u>dheren</u>		·		Spangle	Contrast			Roughnes	<u>is</u>			
		ius causing flakin 30° reverse bend).										
Ratir	2. (3. (5. 0.192 in. 6. 0.252 " 7. 0.320 " 8. 0.400 "	Ratir	3. Low out	rate, spans	gles well def rast. Spang	ined	2. 3.	Moderately Fine to mo sandpaper Rough text		1

- 4 -

TABLE 3

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surface caused by various defects (ridges, dewetting, black spots, pimples)

ан такжа сулартарата (салас сулартаратар на раз в на така) нарактаратар су узародната на так с на на такар драж Паракта сулартарата (салас сулартаратар на раз в на така) нарактаратар су узародната на так с на на такар драж н

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TABLE	4

		·			Steel Wt	age Coating Alloy	Test Resul			Spangle	Spangle		
Test <u>Number</u>	Steel Finish	Coating Wt, oz/so_ft-sheet_	Iron C mg/so ft	ontent g/m ²	Loss g/m ²	Thickness	% Alloy	Ductility	Adherence **	5ize **	Contrast	Brightness	Roughness
l	5	0.47	187	2.0	4-4	-	-	1	1	4	4	2.7	4
1	3	0.43	220	2.4	5.0	-	-	1	1	- 4		2	4
2	5	0.55	393 473	4.2 5.1	6.5 8.7	-	-	1	1 1	4	4	3.3	4
2 3	2 5	0.55 0.43	133	1.4	2.9	-	-	i	ì	ž	4	2.3	ž
3	3	0.43	123	1.3	2.8	-	-	ī	ī	4	4	2.3	4
4	5	0.44	207	2.2	4.8	-	-	l	1	4	4	2	4
4	3	0.50	250	3.0	6.5	-	-	l	1	4	4	2.3	7
5	5	1.88	2890	26.8	25.3	0.0244	59.3	4	5.5	4	4	3.7	3 2
5	3	1.67	2410	26.0	28.5	0.0225	59.6 46.5	3	5.5 6	2 4	2 4	2 2.7	4
6 6	5	2.52 2.26	2680 2633	28.9 28.4	30.0 31.5	0.0300 0.0292	45.9	4 4	6	4	4	2.7	2
7	5	1.81	2360	25.4	23.2	0.0292	40.5 59.4	3.3	5.5	4	4	2	ŝ
7		1.66	2320	26.1	26.6	0.0223	63.C	3.0	5.5	2	2	2	2
8	3 5	2.25	2577	27.8	29.6	0.02/1	49.5	3	6	4	4	3.7	4
8	3	2.28	2627	28.4	28.1	0.0278	51.1	3	6	4	4	4 1.7	4
9	5	0.50	197	2.2	4.0	-	-	1	1	4	4		4
9	3	0.52	197	2.2	5.3		<u> </u>	1 3.3	1 5.5	. 4	4	2 3	4
10	5	1.80 1.76	2523 2567	27.3 27.7	25.0 25.3	0.0232 0.0220	59 . 1 62 .4	3	5.5	2	2	2	2
10 11	3 5	1.29	1946	21.0	18.5	0.0157	60.7	3	5.5	2.7	3.3	ž	3
11	3	1.30	1993	21.4	20.8	0.0174	61.2	3	5.5	1	3	2	2
12	5	3.02	406	47.4	48.7	0.0482	79.6	4	6	3		2.3	4
12	3	2.82	4376	47.1	48.3	0.0483	77.7	4	6	2.7	3 3	2	3
13	5	0.94	1053	13.2	13.4	0.0980	46.2	2	3.2	2	3	1.3	1
13 13	3	1.14	1386	14.9	10.2	0.0117	46.4	2	3.3	1	3	1	1
14 14	3 5 3	2.07	2896	31.2	26.5	0.0318	71.1	3	5.5 5.5	2 1	3 3	1	1
14	3 5	2.21 0.40	3096 166	33.3 1.8	38.9 2.7	0.0298	60.2	1	1 .	2	4	ĩ	1
15	3	0.46	206	2.2	2.3	-	-	i	i	2	4	i.3	4
16	5	1.65	2350	25.3	28.3	0.0231	63.5	3	5.5	Ĩ4	4	2.3	3
16	3	1.67	2436	26.2	25.7	0.0213	60.1	3	5.5	2	2	2.3	2
17	5	1.53	2096	22.5	19.1	0.0157	61.4	2	4.5	4	4	3.3	3
17	35	1.21	2126	22.9	19.7	0.0160	62.3	2	4.5	4	4	3.7	3
18	5	1.02	1676	18.0	18.8	0.0143	59.8	2	4	4	4	2.3	2 2
18	3 5	1.10	1696 3710	18.2 39.9	17.8 35.8	0.0142 0.0383	62.1 71.8	2 3	4	4	4	2	3
19 19	5	2.63 2.31	3570	39.9 38.4	38.3	0.0355	71.3	3	6	4	4	3	2
20	5	3.05	4530	48.S	46.2	0.0496	79.5	4	7	4	4	3.3	3.3
20	3	2.71	4500	48.5	48.2	0.0448	75.6	4	7	4	4	4	2
21	5	1.89	2516	27.1	27.1	0.0237	62.3	ġ	5.5	4	4	2.3	3.3
21	3	1.72	2493	26.8	24.3	0.0269	61.8	3	5.5	2	2	2	2
22	5	2.57	361.3	38.8	44.3	0.0350	58.1	4	6	4	4	4	3
22	3	2.80	3756	40.4	47.7	0.0343	59.1	4	6.5 6.2	4	4	4	4
23	5	2.48 2.70	3123 3216	33.6 34.5	35.4 38.3	0.0314 0.0283	49•5 44•3	4	6	4	4	4	4
23 24	3 5	1.93	2076	22.2	28.8	0.0240	59.7	4	ó	3.3	3.7	3.3	3.7
24 24	3	1.36	2033	21.8	22.6	0.0182	58.5	4	ĕ	4	3.3	3.7	3
25	3	1.77	2270	24.3	24-4	0.0231	60.5	3	6	4	4	4	3.3
25	3	1.90	2296	24.7	26.0	0.0248	60.3	3.7	6	4	4	4	3
26	5	1.18	1543	16.6	18.4	0.0136	49.7	2	4.5	2.7	2.7	2	2
26	3	1.34	1706	18.3	19.6	0.0188	49.4	2	4-5	2	2	2	2
27	5	1.72	2336	25.1	23.4	0.0212	58.2 57.0	3. 3	5.5 5.8	4	4 <u>.</u> 2	2	2
27 28	35	1.70 2.10	2453 3093	26.4 33.3	27.1 33.0	0.0315	77.6	3.7	6	4	4	4.	3
28	3	1.97	3176	34.2	32.0	0.0354	78.7	3	ĕ	3	3	3.7	2
29	5	0.94	1303	14.0	13.3	0.0111	53.3	2	4	ź	ź	2	2
29	3	1.07	1436	15.5	19.1	0.0094	37.8	2	4	2	2	2	1
30	5	4.81	6950	74.8	67.1	0.0837	81.5	5	8	4	4	3	4
30	3	4.54	6876	74.1	71.4	0.0799	81.5	5	8	4	4	3	4

* - each value shown is average of three determinations. ** - see codes in Table 3(a).

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			•		Combi	ned Averages				
Aluminium Content	Immersion Time min	Tests Combined	Number of Results Averaged	Coating wt, oz/so ft-sheet	Iron Content g/m ²	Steel Loss	Alloy Thickness	5 Allor	Ductility *	Adherenc *
C.2	0.5 1.0 2.0	1,3 9,15 2,4	12 12 12	0.44 0.48 0.52	1.6 2.1 3.6	3.S 3.5 6.4	- (2) - - -	-	1 1 1	고 고 王
°C.1	0.25 0.5 1.0	29 11,13,17,18 5,7,10,16,21,	6 24 48	1.01 1.20 1.71	14.8 19.1 26.9	16.2 17.1 25.6	0.0103 0.0168 0.0233	45.6 57.5 61.4	2 2.25 3.1	4 4•3 5•5
	2.0 4.0	27,26,28 12,14,19,20 30	24 6	2.51 4.68	39•4 74•5	41.8 69.3	0.0408	73.4 81.6	3.52 5	6.1 3
0	0.5 1.0 2.0	24,25 6, 8 22,23	12 12 12	1.87 2.32 2.64	23.6 28.4 35.8	24.6 29.8 41.4	0.0228 V 0.0226 X 0.0323 V	59.8 48.3 52.9	3.6 3.6 4.25	6 6 7 6.1

TABLE 5

- 6 -

* - see codes in Table 3(a).

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APPENDIX II

ACCELERATED CORROSION TESTING OF GALVANIZED PANELS

by

J. G. Sibakin

(The Steel Company of Canada, Limited, Hamilton, Ontario) January 21, 1958

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INTRODUCTION

Brief descriptions are given below of the accelerated corrosion tests performed on the galvanized test panels prepared at the Physical Metallurgy Division, Mines Branch, Ottawa. The results obtained are then listed in Tables 1 to 4 (see pages 12 to 15). These results include:

Table 1 - Diffusivity measurements for typical specimens. (Average results for all specimens are listed in Table 2.)
Table 2 - Corrosion results after humidity test.
Table 3 - Weight change data.
Table 4 - Water film test results.

CORROSION TESTS

(a) Humidity Test

The requirements for this test are galvanized panels cut to 4-in. by 5-in., clean 4-in. by 5-in. glass plates (twice as many as the number of galvanized panels), freezer, humidity cabinet, and spring clothes pins (four times as many as the number of galvanized panels). This test is completed in 48 hr and consists of the following cycle repeated twice:

- (i) 5 hr in freezer at O°F (both glass plates and galvanized panels).
- (ii) Moisten two glass plates and one galvanized panel simultaneously by condensation of moist air until all of the original frost disappears.
- (iii) Place the galvanized panel between the two glass plates and fasten the whole assembly with four spring clothes pins.
 - (iv) Place the assembly in the humidity cabinet at 95°F,
 96% R.H., for 16 hr.
 - (v) 4 hr in humidity cabinet at 95°F, 20% R.H., to dry the samples.

After the second cycle, corrosion index values are assigned and the "diffusivity" is determined by reflectometer.

(b) <u>Water Film Test</u>

This test requires twice the number of clean 4-in. by 5-in. glass plates as there are 4-in. by 5-in. galvanized panels to be tested. Two glass plates and one galvanized panel are submerged together in tap water, the galvanized panel being sandwiched between the glass plates under the surface of the water. The whole assembly is fastened together with four spring clothes pins, removed from the water, end stored in the flat position in warm air for 48 hr. Corrosion index values are assigned after 24 and 48 hr, for both white and black stain.

<u>NOTE</u>: In both of these tests, the galvanized samples must be initially flat in order to provide the most reliable results.

- 8 --

CORROSION EVALUATION

In general, corrosion by the humidity test is evaluated by reflectometer, whereas corrosion by the water film test is evaluated by corrosion index.

(a) Reflectometer

The reflectometer used at present at Stelco is the Photoelectric Reflection Meter, Model 610, manufactured by Photovolt Corp., New York.

The "search head" of the instrument is placed on the surface to be evaluated. The light source within the head directs a vertical beam of parallel light towards the sample surface. A circular (washer-shaped) photoelectric cell surrounding the light beam and facing downward picks up light diffused from the sample surface (hence the term diffusivity). The e.m.f. thus generated motivates the galvanometer of the reflectometer, giving a diffusivity reading.

The diffusivity of the galvanized panels in question was actually measured in percent compared to MgO as having a diffusivity of 100. (The standard used in the test was the white standard, which has a diffusivity of 73.0% compared to MgO.) The light from the head to the panel was passed through the tri-stimulus green filter.

(b) <u>Corrosion Index</u>

The corrosion index is a value assigned to the corroded surface after visual examination. Generally two corrosion index values appear together, e.g. ¹2, the upper being the index of the numbered side, the lower that of the unnumbered side. The corrosion index is indicative of corrosion in the following way:

- 9 -

 Index
 0
 1
 2
 3
 4

 % of Surface
 0
 1 to 25
 26 to 50
 51 to 75
 76 to 100

Both white and black staining were evaluated in this manner. In some cases the latter occurred as areas the size of pin-heads, and this is indicated by the index 1°.

EXPERIMENTAL PROCEDURE

The galvanized panels were sheared to 4-in. by 5-in. prior to testing. In the shearing operation, the rough bottom edge was cut off to about 1/4-in. from the bottom. The panel was cut 5-in. up from, and parallel to, the fresh bottom edge, then the sides were trimmed off to give a width of 4-in.

After being cut to size, each panel was

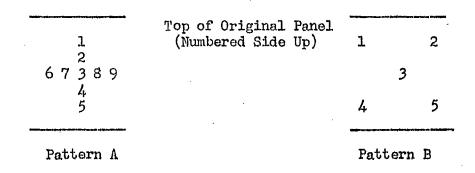
- (i) accurately weighed,
- (ii) tested for diffusivity by reflectometer, and
- (iii) subjected to the two-cycle humidity test.

Upon completion of the humidity test, each sample was

- (i) evaluated by corrosion index,
- (ii) weighed accurately to determine weight gain due to corrosion,
- (iii) tested again for diffusivity,
- (iv) treated with ammonium hydroxide to remove the corrosion products,
 - (v) reweighed to find weight loss due to corrosion,
- (vi) subjected to the water film test for 48 hr, and
- (vii) evaluated by corrosion index after 24 hr and 48 hr under the water film test.

- 10 -

Reflectometer readings of diffusivity were taken on the numbered sides of all of the panels, but were taken in two different patterns over the panels as illustrated below:



As indicated in Table 2, some specimens were tested using pattern A and the remainder using pattern B.

(Tak	les	1	to	4	follow,)
(on	page	эş	12	to	15.)

TABLE 1	

Test	Steel	Time of				Posi	tion				
Number	Finish	Test	1	2	3	4	5	6	7	8	·
1	5 3	B* A** B A	16 28.5 22 31	21.5 26 20 29.5	21.5 30 22 27.5	28 31 25 25	27 30 22 26,5				
5 5	5	B A B A	31.5 29 26.5 30	30 28 26.5 29	31 28•5 25 29	31 29•5 25 31•5	30 31 26 31	30.5 27 26 31	30.5 27.5 25.5 28	32 27 25•5 28	3:

* Specimens tested for diffusivity before humidity test. ** Specimens tested for diffusivity after . humidity test.

.

				n Results Afte		y Test	0	an Tadaa
NumberFindshCorrosionOatn% GainWhiteElack1522.829.16.327.64002522.329.36.025.843002522.329.36.025.844003521.426.45.023.444003322.129.37.232.63004321.226.65.425.53005*533.220.729.64.216.53006*520.833.612.861.544006*320.733.612.861.544006*320.733.612.861.544007*327.032.05.018.543007*327.032.05.018.543008*529.637.17.5618.043009515.330.515.299.33009515.330.515.2300010*530.832.11.34.230011526.931.5 <td>Test</td> <td>Stool</td> <td>Before</td> <td></td> <td>ivity</td> <td></td> <td>Corrosi</td> <td>on index</td>	Test	Stool	Before		ivity		Corrosi	on index
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$					Gain	% Gain	White	Black
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	وم جرونه في مرد الي م		hiren an ar an					
		5	22.8			27.6	4 4	<u> </u>
	1	3				25.7	2,3	~ ⁰
	2	5				25.8	4 3	~ ~
	2	3				-4.3	1 3 4 1	<u> </u>
	3	5					3 4	~ ~
	3	3					1 3	_ 0
	4	2					3 2	~ ^v
	4 5*	2 5			-27		3 3	
	5*	á			4.2		3 1	
		5					47	00
		á					4 2	0 0
		5					4 4	0 0
		3					4 3	
	8*	5					33	~
	8*	3		36.7		18.0	43	
	9	5	15.3	30.5	15.2	99•3	33	0
	9	3	15.5	29.0	13.5	87.1	33	_ ∼
	10*	. 5	30.8	32.1	1.3		33	~ ~
	10*	3		28.4	3.4		23	^ ^V
		5			4.6		43	~ ~
		3			9.1		4 3	~ ~
		5			-0.3		13	_ ∩ Ŭ
		3		32.8			3 3	~ ~
		2					3 3	<u> </u>
		د د	18.9		2.2		2 2	
		3					33	_ ∼
		5					3 3	
		ă					3 3	0 0
		5	36.3				3 3	00
		3					33	
		5					23	- 0
	17	3	49.6	40.1	-9.5	-19.1	4 4	
		5		30.4			1 2	~ ~
		3			2.2		<u> </u>	
		5.					2	∩ Ŭ
		3					23	~ ⁰
		5			-8.8	~18.6	23	^ V
		3					3 0	~ ⁰
		2					3 2	~ ⁰
		5					2 3	~ ^V
		2		•			2 3	~
		5					3 3	٥Õ
		ž					3 /	0 0
24 3 33.4 38.7 5.3 15.8 34 0 25 5 40.5 42.1 1.6 3.9 3 0 0 25 3 42.1 41.9 0.8 1.9 2 3 0 0 26 5 24.8 23.4 -1.4 -5.6 3 2 0 0 26 3 25.8 24.9 -0.7 -2.7 3 0 0 27 5 27.1 26.9 -0.2 -0.7 3 4 0 27 3 26.0 28.7 2.7 10.3 3 3 0 28 5 33.5 30.1 -3.4 -10.1 3 3 0 28 3 32.8 27.6 -5.2 -15.8 2 2 0 29 5 24.8 22.9 -1.9 -7.6 2 3 0 29 3 25.4 24.7 -0.7 -2.7 3 3 0 30 5 33.6 29.1 -4.5 -13.3 3 0 0 30 3 33.9 31.8 -2.1 -6.1 3 4 0		5		36.0			~ 7	0 0 .
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	24	3	33.4	38.7	5.3	15.8	3 1	0 0
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	25	5		42.1	1.6	3.9	3 3	0 0
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	25	3	42.1	41.9	0.8	1.9	23	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	5		23.4	-1.4		32	0 0
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	3	25.8	24.9	-0.7	-2.7	33	~ ~
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	27	5		26.9	-0.2	-0.7	3 4	~ 0
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	27	3		28.7	2.7	10.3	33	~ 0
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		5	33.5		-3.4		23	v
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	28	3	32.8				× 2	~ ~
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		5	24.8				â 3	~ ^v
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	29	3	25.4			-2.7	3 3	~ ~
· · · · · · · · · · · · · · · · · · ·		2			-4.5	ر ₊ر⊥- 1 ک⊷	3 1	
	0	ر	2207	0+1	-ו7	-0•T	- 4	- 0

*Pattern A diffusivity measurements; the results shown are averages of nine measurements. All others are Pattern B, and the results shown are averages of five measurements.

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TAE	

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Weight	Change	Data	(Humidity Test)
			. ~

$ \begin{array}{c c c c c c c c c c c c c c c c c c c $				(All woight	ts in grams)		
						Weight of Panel	Weight Loss
$ \begin{array}{c} 1 & 5 & 68.4729 & 68.5328 & 0.0599 & 68.3451 & 0.127 \\ 1 & 3 & 69.4385 & 69.635 & 0.0640 & 69.7457 & 0.155 \\ 2 & 5 & 68.930 & 69.0275 & 0.0455 & 68.9000 & 0.33 \\ 2 & 3 & 71.1273 & 71.1721 & 0.0448 & 71.1287 & 0.998 \\ 3 & 5 & 69.023 & 69.045 & 0.0522 & 68.9404 & 0.100 \\ 3 & 3 & 70.4530 & 70.4510 & 0.0380 & 70.3732 & 0.077 \\ 4 & 5 & 68.492 & 68.4891 & 0.0399 & 68.3711 & 0.079 \\ 4 & 5 & 70.2004 & 71.1009 & 0.3005 & 70.2108 & 0.0655 \\ 5 & 74.41001 & 74.1425 & 0.0424 & 74.0418 & 0.665 \\ 5 & 76.4383 & 70.8995 & 0.0422 & 76.7944 & 0.66 \\ 6 & 77.457 & 77.3995 & 0.0422 & 76.7944 & 0.66 \\ 6 & 3 & 76.6261 & 73.9220 & 0.0513 & 74.7602 & 0.100 \\ 7 & 3 & 73.6664 & 73.9220 & 0.0513 & 74.7602 & 0.100 \\ 7 & 3 & 76.6261 & 76.6304 & 0.0543 & 76.5744 & 0.66 \\ 8 & 5 & 76.6261 & 76.6304 & 0.0543 & 76.5744 & 0.06 \\ 8 & 3 & 76.6261 & 76.6304 & 0.0543 & 76.5744 & 0.05 \\ 9 & 5 & 69.0546 & 69.0935 & 0.0339 & 68.9760 & 0.039 \\ 9 & 5 & 69.0546 & 69.0935 & 0.0339 & 68.9760 & 0.039 \\ 10 & 5 & 75.2466 & 75.2833 & 0.0427 & 75.1763 & 0.05 \\ 11 & 5 & 72.7570 & 72.7833 & 0.0233 & 72.6399 & 0.067 \\ 12 & 5 & 79.4679 & 77.4420 & 0.0323 & 77.6379 & 0.06 \\ 12 & 5 & 79.4679 & 77.4833 & 0.0233 & 79.3200 & 0.77 \\ 13 & 5 & 71.4827 & 73.9102 & 0.0323 & 79.3209 & 0.07 \\ 13 & 5 & 71.4827 & 73.9102 & 0.0323 & 79.3290 & 0.07 \\ 13 & 5 & 71.4827 & 73.9046 & 0.0333 & 79.3290 & 0.77 \\ 13 & 5 & 71.4827 & 73.906 & 0.0226 & 75.6334 & 0.05 \\ 14 & 5 & 75.6310 & 75.7266 & 0.0326 & 73.4641 & 0.27 \\ 14 & 5 & 75.6310 & 75.726 & 0.0326 & 73.4541 & 0.77 \\ 14 & 5 & 75.6700 & 77.7570 & 72.7849 & 0.0473 & 73.8094 & 0.05 \\ 15 & 5 & 68.7914 & 68.3977 & 0.0473 & 68.6960 & 0.99 \\ 15 & 5 & 74.3518 & 74.3977 & 0.0473 & 73.8094 & 0.05 \\ 16 & 5 & 74.3518 & 74.3977 & 0.0473 & 73.8094 & 0.05 \\ 17 & 5 & 73.8422 & 73.9096 & 0.0279 & 77.3370 & 0.06 \\ 13 & 5 & 72.6503 & 72.6647 & 0.0134 & 72.6334 & 0.05 \\ 14 & 5 & 72.6503 & 72.6647 & 0.0134 & 72.6334 & 0.05 \\ 15 & 5 & 73.9382 & 77.6439 & 0.0234 & 77.5377 & 0.06 \\ 20 & 8 & 0.927 & 77.8493 & 0.0234 & 77.5373 & 0.023 \\ 7 & 74.579 & 74.6$							
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Number	Finish	Weight	Humidity Test	from Corrosion	Corroaion Products	Corrosion
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	· • . 7	~	64 1000	60 5200	0.0500	68, 3/51	0.1278
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$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	2	2					0.0986
6377, 1, 1, 1, 1, 1, 1, 1, 1, 2, 2, 2, 0, 0, 0, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,	~ ~ ~	ر ج					0,1019
6 3 77, 1, 1/3 77, 1, 1/3 70, 1322 0, 0377 77, 0937 0, 053 7 3 74, 7070 74, 74/42 0, 0372 74, 6/24 0, 056 8 5 76, 6261 76, 6304 0, 054, 3 76, 574, 5 0, 056 8 3 78, 184, 1 78, 2207 0, 04, 66 78, 1128 0, 077 9 5 69, 054, 6 69, 0335 0, 038, 4 69, 5950 0, 099 10 5 75, 23, 26 0, 038, 4 69, 5950 0, 099 10 5 75, 23, 23 0, 04, 27 75, 17, 1768 0, 066 11 5 72, 7570 72, 2493 0, 0323 72, 6399 0, 066 12 5 79, 3677 79, 39, 102 0, 0331 73, 2200 0, 077 13 5 71, 4, 452 71, 4403 0, 0249 77, 7062 0, 66 12 3 79, 4056 79, 44, 01 0, 0323 73, 2200 0, 077 13 5 71, 4527 71, 4603 0, 0249 77, 7062 <	2	2					0.0798
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6377, 1, 1, 1, 1, 1, 1, 1, 1, 2, 2, 2, 0, 0, 0, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,	5	2					0.0619
6377, 1, 1, 1, 1, 1, 1, 1, 1, 2, 2, 2, 0, 0, 0, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1, 1,	6	5					0.0639
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	· 6	á				77.0937	0.0538
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$. 7	5				74,7602	0.1082
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	7	3					0.0646
8 3 74.1841 74.2307 0.0466 78.1128 0.07 9 5 69.0546 69.059 0.0389 68.9750 0.08 9 3 69.6904 69.7288 0.0334 69.5950 0.09 10 5 75.2383 0.0427 75.1768 0.666 10 3 75.1932 75.2383 0.0426 75.1333 0.055 11 5 72.7370 72.7893 0.0233 72.6399 0.666 12 5 79.8679 79.8958 0.0279 79.3076 0.666 12 3 70.4563 77.4401 0.0333 79.3290 0.67 13 3 72.2400 75.7136 0.0226 75.6334 0.057 14 3 78.7700 78.7949 0.0249 77.7052 0.66 15 5 68.7914 68.83977 0.0473 68.6960 0.99 15 3 70.4117 70.4387 0.0479 74.2885 0.056 16 </td <td>8</td> <td>5</td> <td></td> <td></td> <td></td> <td>76.5745</td> <td>0.0516</td>	8	5				76.5745	0.0516
95 $69, 69, 66, 69, 69, 350.038368, 97600.08, 960, 0.08, 69, 72839369, 690, 690, 69, 728330.0384, 69, 59500.09610375, 192275, 23830.042775, 17630.06611373, 878773, 91020.032372, 63990.06612579, 687979, 49580.027979, 40760.06612379, 406379, 44010.033379, 32900.07713571, 452771, 46030.022673, 16410.07714575, 691073, 27260.022673, 16410.07714378, 770078, 79490.024977, 70520.06715568, 791468, 39770.047770, 31370.09716574, 483374, 52180.027770, 31370.09716574, 483374, 52180.027770, 31370.09716574, 483374, 52180.032574, 42160.06617573, 521073, 55230.031373, 44090.026418572, 650372, 65470.014774, 52790.04619577, 800977, 80490.0047774, 52790.06619577, 800977, 80490.0047774, 52790.066$	ŝ	3					0.0713
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	9				0.0389	68,9760	0.0845
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	ģ				0.0384	69.5950	0.0954
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		5		75.2893	0.0427		0.0698
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10	3		75.2388	0.0456		0.0549
113 73.8787 73.9102 0.0315 73.8034 0.060125 79.4663 79.4401 0.0333 79.3290 0.07135 71.4527 71.4603 0.0231 71.3919 0.06133 73.2400 73.2726 0.0326 73.1641 0.07145 75.6910 75.7136 0.0226 73.1641 0.07145 75.6910 75.7136 0.0226 77.6334 0.0515568.791468.3970.047368.69600.09153 70.4117 70.4587 0.0470 70.3137 0.09165 74.4833 74.5218 0.0325 74.4216 0.06175 73.8422 73.9096 0.0274 73.8303 0.05185 72.6503 72.6647 0.0147 74.5279 0.04195 77.8009 77.6049 0.0040 77.7357 0.06205 76.7393 76.7476 0.0037 76.6771 0.06205 76.7393 76.7476 0.0037 76.6771 0.06215 74.7852 74.7995 0.0433 74.6732 0.05225 77.3877 0.0230 74.6732 0.05215 74.7862 74.7995 0.0230 74.6732 0.05225 77.3877 0.0230 74.6732 0.05233 80.721080.72440.0232 <td></td> <td>5</td> <td>72.7570</td> <td>72.7893</td> <td>0.0323</td> <td></td> <td>0.0671</td>		5	72.7570	72.7893	0.0323		0.0671
12379.406879.401 0.0333 70.3290 0.07 13571.452771.4608 0.0281 71.3919 0.66 13373.240075.2726 0.0326 73.1641 0.07 14575.691075.7136 0.0226 75.6334 0.05 14378.770078.7949 0.0249 77.7052 0.66 15568.791468.8397 0.0473 68.6960 0.09 15370.411770.4587 0.0479 74.2835 0.66 16374.483374.5218 0.0335 74.4216 0.66 17573.882273.9096 0.0274 73.6303 0.65 17373.521073.5523 0.0313 73.4409 0.68 18572.650372.6647 0.0144 72.6261 0.022 18374.570274.5849 0.0147 74.5279 0.04 19576.7476 0.0037 60.6371 0.66 203 80.9394 80.9284 0.0151 80.6329 0.66 21574.756274.7969 0.0237 74.6732 0.05 22577.3587 0.066 79.5535 0.047 0.0237 77.3195 0.06 21374.728974.7269 0.0237 77.3276 0.05 22 $3.80.9137$ 80.9284 0.0251 80.6529 0.05 22577.33		3	73.8787				0.0693
12379.406879.401 0.0333 70.3290 0.07 13571.452771.4608 0.0281 71.3919 0.66 13373.240075.2726 0.0326 73.1641 0.07 14575.691075.7136 0.0226 75.6334 0.05 14378.770078.7949 0.0249 77.7052 0.66 15568.791468.8397 0.0473 68.6960 0.09 15370.411770.4587 0.0479 74.2835 0.66 16374.483374.5218 0.0335 74.4216 0.66 17573.882273.9096 0.0274 73.6303 0.65 17373.521073.5523 0.0313 73.4409 0.68 18572.650372.6647 0.0144 72.6261 0.022 18374.570274.5849 0.0147 74.5279 0.04 19576.7476 0.0037 60.6371 0.66 203 80.9394 80.9284 0.0151 80.6329 0.66 21574.756274.7969 0.0237 74.6732 0.05 22577.3587 0.066 79.5535 0.047 0.0237 77.3195 0.06 21374.728974.7269 0.0237 77.3276 0.05 22 $3.80.9137$ 80.9284 0.0251 80.6529 0.05 22577.33	12	5	79.8679	79.8958			0.0603
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18 3 74.5702 74.5849 0.0147 74.5279 0.04 19 5 77.8009 77.8049 0.0040 77.7357 0.06 19 3 80.2304 80.2391 0.0037 80.1611 0.66 20 5 76.7393 76.77476 0.0033 76.67771 0.066 20 3 80.9137 80.9238 0.0151 80.8529 0.066 21 5 74.7562 74.7995 0.0230 74.7216 0.033 21 3 74.7289 74.7569 0.0230 74.6732 0.057 22 5 77.3614 77.3887 0.0273 77.3195 0.044 22 3 79.5998 79.7064 0.1066 79.5535 0.044 23 5 79.1137 79.1375 0.0238 79.0621 0.057 23 3 80.7210 80.7442 0.0232 80.6700 0.057 24 5 73.3326 73.3667 0.0341 73.2939 0.032 24 5 73.9352 74.0111 0.0259 73.9313 0.057 25 3 75.6130 75.6423 0.0293 75.5702 0.044 26 5 72.4090 72.4300 0.0293 75.5702 0.044 26 5 72.4090 72.4300 0.0299 74.5431 0.057 27 5 74.6797 74.6217 0.0238 74.5431 <t< td=""><td>13</td><td>5</td><td>71.4527</td><td></td><td></td><td></td><td>0.0608</td></t<>	13	5	71.4527				0.0608
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$\begin{array}{cccccccccccccccccccccccccccccccccccc$		ī				74.7216	0.0346
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	21	3					0.0557
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	22	5				77 . 3195	0.0419
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	22	3		79.7064			0.0463
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	23	5		79.1375			0.0516
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	23	3	80,7210				0.0510
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	24	5	73.3326	73.3667			0.0387
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	24		75.3824	75.9084			0.0442
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	25	5				73.9313	0.0539
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- 15 -

TA	BLE	4

<u> </u>		Water Film Test Results Corrosion	Tuder	
Test Number	Steel Finish	White Black 24 Hr		3lack
1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 10 11 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 10 11 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 10 11 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 11 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 0 3 1 1 1 2 2 3 3 4 4 5 5 6 6 7 7 8 8 9 9 0 3 1 1 1 1 2 2 3 1 1 1 1 1 1 1 1 1 1 1 1	5353535353535 3535353535353535353535353	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	2442249332229994992922249929222992299229	

RECOMMENDATIONS

The results of the corrosion tests were plotted against composition, but failed to show any strong relationship. It is hoped that when the results are used in a multiple correlation a relationship may be revealed. To this end, therefore, it is recommended that the following relationships be examined by correlation:

- (1) Corrosion index after humidity test against composition
 - of coating.
- (2) Gain in diffusivity (absolute or as per cent of original; the latter seems to show a much stronger trend) against composition.
- (3) Weight gain during humidity test against composition.
- (4) Corrosion index after the water film test against

composition.

It is felt that the weight loss data are less reliable than the weight gain figures, owing to the difficulty of complete removal of corrosion products without removing zinc simultaneously. Accordingly, it is believed that correlation of weight loss figures is unlikely to yield useful results.

<u>APPENDIX III(a)</u>

PRELIMINARY STATISTICAL ANALYSIS OF COATING TEST DATA

by

H. L. Williams

(Consolidated Mining and Smelting Company of Canada, Limited, Trail, B.C.) February 19, 1958

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INTRODUCTION

The data produced as part of the galvanizing project being carried out at the Mines Branch, Ottawa, under the sponsorship of the Canadian Zinc Research and Development Committee have been studied.

The data consisted of "test logs" obtained in a statistically designed experiment. The various properties of interest have been treated separately and the findings to date are set forth below.

Standard regression analyses were applied to each set of data for both steel finishes. In each case, an equation of the type shown below was fitted.

$$X = b_0 + b_1 X_1 + b_2 X_2 + b_3 X_3 + b_4 X_4 + b_{11} X_1^2 + b_{22} X_2^2 + b_{33} X_3^2$$

+ $b_{44} X_4^2 + b_{12} X_1 X_2 + b_{13} X_1 X_3 + b_{14} X_1 X_4 + b_{23} X_2 X_3$
+ $b_{24} X_2 X_4 + b_{34} X_3 X_4$

where: $X_1 = (bath temperature °C - 455)/10$,

 $X_2 = \log (\text{immersion time, min})/0.3010,$

 $X_3 = (aluminium content \% - 0.1)/0.1,$

 $X_{1} = (lead content \% - 0.5)/0.5,$

Y = the dependent variable in question, and

b's = the unknown regression coefficients.

All regression coefficients were then tested for significance, using a "t" test. Most of those which were found to be non-significant were eliminated and the calculations repeated. In the case where quadratic or interaction terms were significant, the corresponding linear terms were left in the equation even though in some cases they were not significant. Also in the case where a term was significant for one steel finish but not for the other, the non-significant term was left in for comparative purposes.

The standard deviation (S_s) of the variation between specimens within baths, the standard deviation (S_B) of the variation between similar baths, and the standard deviation (S_R) of the variation about regression have also been calculated and are shown for each of the first six dependent variables. The "yardsticks" used in measuring the last six dependent variables were too rough to obtain measures of S_B or S_s , therefore only S_R is shown for these variables. The number of degrees of freedom upon which the various standard deviations are based is also shown. The standard error of each regression coefficient has been set forth directly beneath its respective co-efficient. Where the regression coefficient is not significant, it has been marked by an asterisk. All statistical tests were carried out at the 95% confidence level.

It is intended that this study be continued. A canonical analysis will be carried out on the significant variables affecting each dependent variable. Where no significant difference has been found between the relationships for the two types of steel finish, these will be combined and a canonical analysis carried out on the combined data. The results of these analyses will be forwarded as completed.

- 18 -

No. 3 Steel Finish $Y_1 = 1.66 + 0.11X_1 + 0.59X_2 - 0.91X_3 + 0.25X_2^2 - 0.44X_3^2$ 0.06 0.06 0.09 0.06 0.09 SE b's 0.12 $S_R = 0.31$ df = 24 $S_B = 0.03$ df = 5 $S_S = 0.06$ df = 60 No. 5 Steel Finish $Y_1^1 = 1.75 + 0.17X_1 + 0.63X_2 - 0.90X_3 + 0.24X_2^2 - 0.55X_3^2$ SE b's 0,12 0.08 0,08 0.12 0.08 0.15 $S_R = 0.40$ df = 24 $S_B = 0.09$ df = 5 $S_a = 0.09$ df = 60

These equations show that immersion time (X_2) and aluminium content (X_3) are the most important of the factors studied. The effect of bath temperature (X_1) is minor and lead content (X_1) failed to show a significant effect. The relationships found for the two types of steel finish do not differ to a significant degree.

 Y_2 - IRON CONTENT OF COATING (g/m²)

<u>No. 3 Steel Finish</u> $Y_2 = 26.57 + 3.28X_1 + 9.99X_2 - 13.51X_3 + 4.27X_2^2 - 13.21X_3^2$ SE b's - 1.50 1.06 1.06 1.50 0.98 1.95 $S_R = 5.21 \text{ df} = 24$ $S_B = 0.59 \text{ df} = 5$ $S_S = 1.28 \text{ df} = 60$ No. 5 Steel Finish

 $Y_2^{1} = 25.98 + 3.45X_1 + 10.21X_2 - 13.49X_3 + 4.28X_2^{2} - 13.03X_3^{2}$ SE b's = 1.69 1.19 1.19 1.69 1.10 2.19 $S_R = 5.84 \quad df = 24$ $S_B = 0.94 \quad df = 5$ $S_s = 1.30 \quad df = 60$

The factors, bath temperature, immersion time and aluminium content, were found to affect the iron content of the zinc coating to

- 19 -

a significant degree. The latter two are the most important. Variation in lead content failed to show a significant effect. The relationships found for the two types of steel finish did not differ to a significant degree.

Y_3 - STEEL WEIGHT LOSS (g/m²)

No. 3 Steel Finish

 $Y_{3} = 26.83 + 3.52X_{1} + 10.44X_{2} - 13.30X_{3} + 4.36X_{2}^{2} - 10.66X_{3}^{2} - 3.77X_{2}X_{3}$ SE b's = 1.30 0.92 0.92 1.30 0.84 1.69 1.59 $S_{R} = 4.50 \quad df = 23$ $S_{B} = 0.77 \quad df = 5$ $S_{S} = 4.03 \quad df = 60$

No. 5 Steel Finish

 $Y_{3}^{1} = 24.97 + 3.33X_{1} + 9.34X_{2} - 13.56X_{3} + 3.57X_{2}^{2} - 9.36X_{3}^{2} - 3.22X_{2}X_{3}$ SE b's = 1.69 1.20 1.20 1.69 1.10 2.20 2.07 $S_{R} = 5.86 \quad df = 23$ $S_{B} = 0.96 \quad df = 5$ $S_{s}^{B} = 3.16 \quad df = 60$

The steel weight loss during galvanizing and subsequent stripping was found to be a function of the bath temperature, immersion time and aluminium content. The latter two are the most important. Variation in lead content failed to show a significant effect. The relationships found for the two types of steel finish did not differ to a significant degree. These equations bear a close resemblance to those found for iron content of coating, Y_2 .

Y, - COATING THICKNESS (mm)

No. 3 Steel Finish

 $Y_{4} = 0.0378 + 0.0129X_{2} - 0.0207X_{3} + 0.0047X_{2}^{2} - 0.0114X_{3}^{2} - 0.0063X_{2}X_{3}$ SE b's = 0.0021 0.0015 0.0021 0.0014 0.0027 0.0026 $S_{R} = 0.0072 \quad df = 24$ $S_{B} = 0.0032 \quad df = 5$ $S_{S} = 0.0021 \quad df = 60$ No. 5 Steel Finish

 $x_{4}^{1} = 0.0362 + 0.0139x_{2} - 0.0216x_{3} + 0.0053x_{2}^{2} - 0.0089x_{3}^{2} - 0.0060x_{2}x_{3}$ SE b'a = 0.0023 0.0017 0.0023 0.0015 0.0030 0.0029 $S_{R} = 0.0081 \quad df = 24$ $S_{B} = 0.0009 \quad df = 5$ $S_{s} = 0.0021 \quad df = 60$

Coating thickness was found to be a function of immersion time and aluminium content. Variation in lead content and in bath temperature failed to show significant effects. The relationships found for the two steel finishes were not significantly different.

In the case of the high aluminium data, the maximum and minimum figures given were averaged and the averages used in the calculations. A more satisfactory measurement could have been obtained originally if several measurements had been made at random and these averaged for each specimen.

 Y_{5} - ALLOY THICKNESS (mm)

No. 3 Steel Finish

 $X_5 = 0.0236 + 0.0033X_1 + 0.0108X_2 - 0.0132X_3 + 0.0047X_2^2 - 0.0134X_3^2$ SE b's = 0.0021 0.0015 0.0015 0.0021 0.0014 0.0027

 $\begin{array}{rll} {\rm S}_{\rm R} &=& 0.0073 & {\rm df} &=& 24 \\ {\rm S}_{\rm B} &=& 0.0019 & {\rm df} &=& 5 \\ {\rm S}_{\rm S} &=& 0.0013 & {\rm df} &=& 60 \end{array}$

No. 5 Steel Finish

 $x_{5}^{1} = 0.0229 + 0.0031 \dot{x}_{1} + 0.0116 x_{2} - 0.0139 x_{3} + 0.0056 x_{2}^{2} - 0.0126 x_{3}^{2}$ SE b's = 0.0022 0.0016 0.0016 0.0022 0.0014 0.0029 $s_{R} = 0.0076 \quad df = 24$ $s_{B} = 0.0010 \quad df = 5$ $s_{5} = 0.0010 \quad df = 60$

Alloy thickness was found to be a function of bath temperature, immersion time and aluminium content. The latter two were the most

- 21 -

important. Variation in lead content failed to show a significant effect. The relationships found for the two steel finishes were not significantly different.

In the case of the missing thickness measurements for the high aluminium specimens, an estimate of 0.0001 mm was used. Since under microscopic examination the alloy layer was found to be discontinuous and thin, and since some estimate is required for the calculations, this very small figure was chosen.

Y6 - RATIO, ALLOY THICKNESS : COATING THICKNESS (%)

No. 3 Steel Finish

	¥6	8044 977	62.71	+ 4.65X +	• 5.23X ₂ -	25.16X3	$-34.55x_3^2$
SE	b's	H	1.68	1.45	1.45	2.05	2.65
				S _R = S _B = S _S =	7.11 2.15 1.32	df = df = df =	25 5 60
	No		5 Steel	Finish			

No. 5 Steel Finish

 $Y_{6}^{1} = 64.14 + 4.06X_{1} + 4.91X_{2} - 25.50X_{3} - 35.63X_{3}^{2}$ SE b's = 1.55 1.34 1.34 1.89 2.45 $S_{R} = 6.56 \qquad df = 25$ $S_{B} = 1.78 \qquad df = 5$ $S_{g} = 1.68 \qquad df = 60$

The alloy : coating ratio was found to be a function of bath temperature, immersion time and aluminium content. Variation in lead content of the bath failed to show a significant effect. The relationships found for the two steel finishes were not significantly different.

In the case of the high aluminium specimens, even though neither alloy thickness or coating thickness were known with any degree of accuracy, an estimate had to be inserted. Since a very low ratio was indicated, 3% was chosen. CODE 1. Excellent, no cracking

- 2. Good, network of fine cracks
- 3. Fair, general cracking, with coating broken up into small blocks
- 4. Poor, wide separation of medium size blocks
- 5. Very poor, general peeling of coating in large blocks

No. 3 Steel Finish

 $Y_7 = 2.94 + 0.19 X_1 + 0.47 X_2 - 1.39 X_3 - 0.56 X_3^2$ SE b's = 0.11 0.10 0.10 0.14 0.17

 $S_{\rm R} = 0.47$ df = 25

No. 5 Steel Finish

 $Y_7^1 = 3.07 + 0.22X_1 + 0.50X_2 - 1.33X_3 - 0.74X_3^2$ SE b's = 0.12 0.11 0.11 0.15 0.19 $S_R = 0.52$ df = 25

Ductility was found to be a function of bath temperature, immersion time and aluminium content. The latter two factors are the most important of those studied. Variation in lead content failed to show a significant effect. The relationships found for the two steel finishes did not differ to a significant extent.

Yg - ADHERENCE (Bend Test)

<u>CODE</u> - Minimum bend radius causing flaking (90° bend plus a 180° reverse bend).

1.	0.050-in.	5.	0.192-in.
2.	0.070 " 0.100 "	6.	0.252 "
3.	0.100 "	7.	0.320 "
4.	0.144 "	8.	0.400 "

No. 3 Steel Finish

 $Y_8 = 5.43 + 0.32X_1 + 0.65X_2 - 2.54X_3 - 1.88X_3^2$ SE b's = 0.13 0.12 0.12 0.16 0.21

 $S_{R} = 0.57$ df = 25

No. 5 Steel Finish

$$Y_8^1 = 5.45 + 0.30X_1 + 0.73X_2 - 2.53X_3 - 1.93X_3^2$$

SE b's = 0.17 0.14 0.14 0.20 0.26
 $S_p = 0.70$ df = 25

Adherence was found to be a function of bath temperature, immersion time and aluminium content. Variation in lead content failed to show a significant effect. The relationships found for the two steel finishes did not differ to a significant degree.

3. Small

4. No spangle

 $\frac{No. 3 \text{ Steel Finish}}{Y_9} = 2.37 - 0.17 X_3 - 1.03 X_4 + 0.42 X_3^2 + 0.13 X_4^2$ SE b's = 0.16 0.16 0.16 0.21 0.21 $S_R = 0.59 \quad \text{df} = 25$

No. 5 Steel Finish

 $Y_9^1 = 3.66 - 0.11 \overset{*}{X}_3 - 0.69 X_4 + 0.26 \overset{*}{X}_3^2 - 0.44 X_4^2$ SE b's = 0.15 0.15 0.15 0.20 0.20

 $S_{R} = 0.56$ df = 25

Panels made of steel finish No. 3 had significantly larger spangles on the average than panels of steel finish No. 5.

With finish No. 3, spangle sizes was found to be a function of the aluminium and lead contents of the bath. Bath temperature and immersion time failed to show a significant effect.

With finish No. 5, the only factor of significance was lead content.

In both cases, the amount of variation obtained in spangle size was small.

Y10 - SPANGLE CONTRAST

3. Low or no contrast. Spangles outlined only.

4. No contrast (no spangles).

No. 3 Steel Finish

 $Y_{10} = 2.40 + 0.06 \overset{*}{X}_{3} - 0.33 \chi_{4} + 1.26 \chi_{3}^{2} + 0.84 \chi_{4}^{2}$ SE b's = 0.13 0.13 0.13 0.17 0.17

$$S_{\rm R} = 0.45$$
 df = 25

<u>No. 5 Steel Finish</u> $Y_{10}^{1} = 3.64 + 0.03 X_{3}^{*} - 0.31 X_{4} + 0.35 X_{3}^{*2} - 0.06 X_{4}^{*2}$ SE b's = 0.13 0.13 0.13 0.17 0.17 $S_{R} = 0.47$ df = 25

Panels made of steel finish No. 3 showed more spangle contrast on the average than panels of steel finish No. 5.

With finish No. 3, spangle contrast was found to be a function of the lead and aluminium contents of the bath. Bath temperature and immersion time failed to show a significant effect.

With finish No. 5, the only factor of significance was lead content.

In both cases, the amount of spangle contrast obtained was small.

Y11 - BRIGHTNESS (Photometer Reading)

 $\begin{array}{cccc} \underline{\text{CODE}} - 1 & 0 & -1.25 \\ 2 & 1.5 & -2.75 \\ 3 & 3.0 & -4.25 \\ 4 & 4.5 & + \end{array}$

No. 3 Steel Finish

 $Y_{11} = 2.42 + 0.38X_1 - 0.75X_3 - 0.50X_4 + 0.63X_3^2 - 0.24X_4^{*2}$ SE b's = 0.17 0.12 0.17 0.23 0.23 0.23

$$s_{R} = 0.60$$
 df = 24

No. 5 Steel Finish

 $X_{11}^{1} = 2.71 + 0.35X_{1} - 0.83X_{3} - 0.42X_{4} + 0.23X_{3}^{*2} - 0.48X_{4}^{2}$ SE b's = 0.16 0.11 0.16 0.16 0.20 0.20 $S_{R} = 0.55$ df = 24

Brightness was found to be a function of bath temperature, aluminium content and lead content. Immersion time failed to show an effect. The relationships found for the two steel finishes did not differ to a statistically significant degree.

Y12 - ROUGHNESS

CODE - 1. Very smooth

2. Moderately smooth

- 3. Fine to moderately rough, sandpaper texture
- 4. Rough texture

No. 3 Steel Finish

$$Y_{12} = 2.06 + 0.29X_1 + 0.38X_2 - 0.25X_3 + 0.03X_3^2$$

SE b's = 0.13 0.11 0.11 0.16 0.21
 $S_R = 0.55$ df = 25

No. 5 Steel Finish

 $x_{12}^{1} = 2.76 + 0.37x_{1} + 0.26x_{2}^{*} - 0.50x_{3} + 0.41x_{3}^{*2}$ SE b's = 0.16 0.14 0.14 0.19 0.25

 $S_R = 0.67$ df = 25

Panels made of steel finish No. 5 had significantly rougher coatings on the average than panels of steel finish No. 3.

With steel finish No. 3, roughness was found to be a function of bath temperature, immersion time and aluminium content. Variation in lead content failed to show a significant effect.

With steel finish No. 5, roughness was found to be a function of only bath temperature and aluminium content.

- 26 -

APPENDIX III(b)

PRELIMINARY STATISTICAL ANALYSIS OF ACCELERATED CORROSION TEST DATA

by

H. L. Williams

(Consolidated Mining and Smelting Company of Canada, Limited, Trail, B.C.) March 4, 1958

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INTRODUCTION

The corrosion data that were obtained by J. G. Sibakin of the Steel Company of Canada, Limited, on the galvanizing project panels have been studied.

The various properties of interest have been treated separately in a manner similar to that of Appendix III(a).

Standard regression analyses were applied to each set of data for both steel finishes. In each case, an equation of the type shown below was fitted.

 $\begin{aligned} \mathbf{Y} &= \mathbf{b}_0 + \mathbf{b}_1 \mathbf{X}_1 + \mathbf{b}_2 \mathbf{X}_2 + \mathbf{b}_3 \mathbf{X}_3 + \mathbf{b}_4 \mathbf{X}_4 + \mathbf{b}_{11} \mathbf{X}_1^2 + \mathbf{b}_{22} \mathbf{X}_2^2 + \mathbf{b}_{33} \mathbf{X}_3^2 + \mathbf{b}_{44} \mathbf{X}_4^2 \\ &+ \mathbf{b}_{12} \mathbf{X}_1 \mathbf{X}_2 + \mathbf{b}_{13} \mathbf{X}_1 \mathbf{X}_3 + \mathbf{b}_{14} \mathbf{X}_1 \mathbf{X}_4 + \mathbf{b}_{23} \mathbf{X}_2 \mathbf{X}_3 + \mathbf{b}_{24} \mathbf{X}_2 \mathbf{X}_4 + \mathbf{b}_{34} \mathbf{X}_3 \mathbf{X}_4 \\ &\text{where } \mathbf{X}_1 = (\text{bath temperature } \circ \mathbf{C} - 455)/10, \end{aligned}$

 $X_2 = (\log \text{ immersion time, min})/0.3010,$

 $X_3 = (alumintum content \% - 0.1)/0.1,$

 $X_4 = (lead content \% - 0.5)/0.5,$

Y = the dependent variable in question, and

b's = the unknown regression coefficients.

All regression coefficients were then tested for significance, using a "t" test. Most of those which were found to be non-significant were climinated and the calculations repeated. In the case where quadratic terms were significant, the corresponding lipear terms were left in the equation, even though in some cases they were not significant. Also, in the case where a term or terms were significant for one steel finish but not for the other, the corresponding non-significant term or terms were left in for comparative purposes.

The standard deviation of the variation between similar baths (S_B) , and the standard deviation of the variation about regression (S_R) , have been calculated and are shown in each case. The "yardsticks" used in measuring the "corrosion index" (Y_{14}, Y_{17}, Y_{18}) were too rough to obtain a measure of S_B and, therefore, only S_R is shown for these variables. The number of degrees of freedom upon which the various standard deviations are based is also shown. The standard error of each regression coefficient has been set forth directly beneath its respective coefficient. Where the regression coefficient is not significant, it has been marked with an asterlsk. All statistical tests were carried out at the 95% confidence level.

Canonical analysis will be carried out on the significant variables of each dependent variable and our findings reported in due course.

Since black staining only occurred in a few panels, no attempt was made to treat its occurrence statistically.

- 28 -

Y13 - PERCENT GAIN IN AVERAGE DIFFUSIVITY AFTER THE HUMIDITY TEST

 $\frac{No. 3 \text{ Steel Finish}}{Y_{13}} = 7.23 - 6.01^{*}X_{1} + 10.88X_{3}^{*} + 8.07X_{4}^{*} - 7.54X_{1}^{2} + 19.96X_{3}^{2} + 17.78X_{4}^{2}$ SE b's = 6.35 3.75 5.31 5.31 3.48 6.95 6.95 $\frac{S_{R} = 18.4}{S_{B} = 3.2} \quad \text{df} = 23$ $\frac{No. 5 \text{ Steel Finish}}{Y_{13}^{1}} = -7.12 - 5.13X_{1} + 14.68X_{3} + 5.08X_{4}^{*} - 3.69X_{1}^{*2} + 35.77X_{3}^{2} + 20.56X_{4}^{2}$ SE b's = 5.80 3.43 4.85 4.85 3.18 6.35 6.35

 $S_{R} = 16.8$ df = 23 $S_{B} = 6.7$ df = 5

The factors, aluminium and lead content, were found to affect the percent gain in average diffusivity to a significant degree for both finishes. Bath temperature also had a significant effect with No. 3 steel finish. Immersion time was not significant in either case.

 $Y_{1,\ell}$ - CORROSION INDEX - WHITE, AFTER HUMIDITY TEST

There were no significant factors found for either No. 3 or No. 5 steel finishes.

$$S_{R} = 0.55$$

 $S_{B} = 0.56$

CODE for CORROSION INDEX:

INDEX		2 OF SURFACE CORRODED
0	-	0
1		l to 25
2	-	26 to 50
3	-	51. to 75
4		75 to 100
		•

- 29 -

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Y15 - WEIGHT GAIN FROM CORROSION AFTER HUMIDITY TEST

No. 3 Steel Finish

 $Y_{15} = 34.8 + 5.2X_{1} + 0.9X_{2}^{*} + 0.2X_{3}^{*} + 5.8X_{4}^{*} + 3.8X_{1}^{2*} - 1.3X_{2}^{*2} + 15.9X_{3}^{2} - 6.5X_{4}^{*2}$ SE b's = 6.5 3.2 3.2 4.4 4.4 3.0 3.0 6.1 6.1 $S_{R} = 15.9 \qquad df = 21$ $S_{B} = 6.2 \qquad df = 5$

No. 5 Steel Finish

 $x_{15}^{1} = 42.8 + 2.2x_{1}^{*} - 2.8x_{2}^{*} + 6.4x_{3} + 6.5x_{4} - 5.0x_{1}^{2} - 7.3x_{2}^{2} + 9.1x_{3}^{2} - 8.8x_{4}^{2}$ SE b's = 3.1 1.6 1.6 2.2 2.2 1.4 1.4 2.9 2.9 $s_{R} = 7.6 \qquad df = 21$ $s_{B}^{R} = 7.8 \qquad df = 5$

The factor, aluminium content, was found to affect the weight gain from corrosion to a significant degree for both finishes. Bath temperature, immersion time and lead content also had a significant effect with No. 5 steel finish.

 $Y_{16} - \text{WEIGHT LOSS FROM CORROSION AFTER HUMIDITY TEST (mg)}$ No. 3 Steel Finish $Y_{16} = 64.0 + 4.1X_2^* + 24.5X_3 + 8.9X_2^2$ SE b's = 4.2 3.6 5.1 3.3 $S_R = 17.6 \quad df = 26$ $S_B = 4.8 \quad df = 5$ No. 5 Steel Finish $Y_{16}^1 = 65.2 + 3.3X_2^* + 21.8X_3 + 2.6X_2^2$ SE b's = 4.6 4.0 5.7 3.7 $S_R = 19.7 \quad df = 26$ $S_B = 25.6 \quad df = 5$

The factor, aluminium content, was found to affect the weight loss from corrosion to a significant degree for both finishes. Immersion time also had a significant effect with No. 3 finish. Bath temperature and load content failed to show a significant effect. $Y_{17} - \text{CORROSION INDEX} - \text{WHITE } (24 \text{ hr}) - \text{AFTER WATER FILM TEST}$ $\frac{\text{No. 3 Steel Finish}}{Y_{17}} = 2.56 - 0.21X_3^* + 0.15X_3^2$ SE b's = 0.11 0.14 0.18 $S_R = 0.48 \qquad \text{df} = 27$ $\frac{\text{No. 5 Steel Finish}}{Y_{17}} = 2.42 - 0.25X_3 + 0.42X_3^2$ SE b's = 0.08 0.10 0.13 $S_R = 0.34 \qquad \text{df} = 27$

The factor, aluminium content, was found to affect the corrosion index (24 hr) to a significant degree for No. 5 steel finish but not for No. 3. Bath temperature, immersion time and lead content failed to show a significant effect.

 x_{18} - CORROSION INDEX - WHITE (48 hr) - AFTER WATER FILM TEST <u>No. 3 Steel Finish</u> $x_{18} = 2.77 + 0.08x_2^* - 0.21x_3^* - 0.02x_2^2 + 0.20x_3^2$ SE b's = 0.12 0.09 0.12 0.08 0.16 $s_R = 0.43$ df = 25

 $Y_{18}^{1} = 2.63 + 0.02X_{2}^{*} - 0.12X_{3}^{*} - 0.15X_{2}^{2} + 0.43X_{3}^{2}$ SE b's = 0.09 0.06 0.09 0.06 0.12 $S_{p} = 0.31$ df = 25

No. 5 Steel Finish

The factors, immersion time and aluminium content, were found to have a significant effect upon the corrosion index (48 hr) of No. 5 steel finish, but not of No. 3 steel finish. Bath temperature and lead content failed to show a significant effect.

- 31 -

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APPENDIX IV

CANONICAL ANALYSIS OF COATING TESTS AND ACCELERATED-CORROSION DATA

by

H. L. Williams

(Consolidated Mining and Smelting Company of Canada, Limited, Trail, B.C.) May 23, 1958

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INTRODUCTION

A canonical analysis has been carried out on each of the relationships reported in Appendices III(a) and III(b). Where the two steel finishes failed to show a significant difference, the data were combined and the canonical analysis carried out on the combined data. Graphs have been plotted showing a few of these relationships. The nomenclature and symbols used below are the same as those used by O. L. Davies in Chapter 11 of his book, "Design and Analysis of Industrial Experiments" (Hafner Publishing Company, New York, N. Y., 1956), in which a good exposition of this type of analysis is given.

As in the previous appendices,

 $x_1 = (bath temperature °C - 455)/10,$

 $x_2 = \log (\text{Immersion time, min})/0.3010,$

 $x_3 = (aluminium content \% - 0.1)/0.1,$

 $x_1 = (1ead content \% - 0.5)/0.5$, and

Y = the dependent variable in question.

Y1 - COATING WEIGHT (oz/sq ft)

Both steel finishes combined.

The full second degree equation was fitted to the data using standard regression techniques.

 $Y_1 = 1.76 + 0.14x_1 + 0.61x_2 - 0.90x_3 - 0.06x_1^2 + 0.24x_2^2 - 0.51x_3^2$ SE b's = 0.08 0.05 0.05 0.07 0.04 0.04 0.09

+
$$0.05x_1x_2$$
 + $0.01x_1x_3 - 0.18x_2x_3$

0.06 0.08 0.08

X₁ X₂ X₃

The standard error for each term is shown directly below its respective term.

The fitted equation was then reduced to the canonical form.

a) The coordinates x_{1s} , x_{2s} , x_{3s} of the center point S of the system and the predicted value \tilde{Y}_{1s} at this point were found to be as follows.

 $x_{1s} = 0.49$, $x_{2s} = -1.55$, $x_{3s} = -0.61$ and $Y_{1s} = 1.60$ oz/sq ft.

b) The canonical form of the second degree equation is $Y_1-1.60 = -0.06 X_1^2 + 0.25X_2^2 - 0.52X_3^2$

c) The new coordinates $(X_1, X_2 \text{ and } X_3)$ for any point are given in terms of the old coordinates (x_1, x_2, x_3) by equations which may be written as follows.

$(x_1 - 0.49)$	$(x_2 + 1.55)$	$(x_3 + 0.61)$
0.9964 -0.0828	-0.0800 -0.9901	0.0268 0.1138
-0.0159	0.1151	0.9932

The entries in the rows are the coefficients in the equation which expresses the X's in terms of the x's. Thus the first such equation is

 $x_1 = 0.9964$ ($x_1 - 0.49$) - 0.0800 ($x_2 + 1.55$) + 0.0268 ($x_3 + 0.61$)

Inspection of the canonical form (b) shows that the third term is the predominant one. Any movement away from the center point in the direction of the X_2 axis will lead to a rapid drop in coating weight, but changes in both X_1 and X_2 can be made with considerably smaller effects on yield. The X_2 coefficient is positive, therefore, a gain in coating weight can be expected on moving away from the center point S in either direction along the X_2 axis. The standard errors of the coefficients in this equation are of roughly the same magnitude as those of the quadratic effects in the original equation. The coefficient of X_1 is therefore not significant. Future experiments should be made in the direction of the X_2 axis to obtain an increased coating weight.

In Figure 1, this relationship has been plotted, by assuming that the non-significant factor X_1 is zero and by setting x_1 equal to 0.49 (460°C)(860°F). Inspection of this graph will show the promising regions.

- 33 -

 Y_2 - IRON CONTENT OF COATING (g/m²)

Both steel finishes combined.

For this and the following dependent variables, the various equations and their canonical forms will be listed in the order used for Y₁. A few comments will be made on each.

Second degree equation -

 $Y_2 = 26.82 + 3.36x_1 + 10.10x_2 - 13.50x_3 - 0.54x_1^2 + 4.21x_2^2 - 13.26x_3^2$ 0.71 0.42 0.60 0.39 0.39 0.78 SE b's 0.42 + $1.00x_1x_2 - 0.64x_1x_3 - 2.93x_2x_3$ 0.73 0.52 0.73 Canonical form -Center of system - S Ð.) $\begin{array}{rcl} x_{1s} &=& 1.88\\ x_{2s} &=& -1.56\\ x_{3s} &=& -0.38\\ Y_{2s} &=& 24.70 \text{ g/m}^2 \end{array}$. b) Equation $x_2 - 24.70 = -0.59x_1^2 + 4.39x_2^2 - 13.39x_3^2$ c) Axes - $(x_1 - 1.88)$ $(x_2 + 1.56)$ $(x_3 + 0.38)$ 0.1077 X1 X2 X3 -0.9941 -0.1063 0.0130 -0.9905 0.0880 0.0824 0.9964

Inspection of the canonical form shows that the third term is predominant and consists mainly of the effect of aluminium content. This (X_3) term is negative and any movement away from the center point in the direction of the X_3 axis will lead to a drop in iron content of the coating. The second term (X_2) is also significant but smaller and is mainly the effect of immersion time. This term is positive and a gain in iron content can be expected on moving away from the center point S along the X_2 axis. The third term, X_1 , is not significant.

In Figure 2, this relationship has been plotted by assuming that the X_1 term is zero and by setting x_1 equal to 0.49 (460°C) (860°F). A reduction in bath temperature would tend to decrease the iron content of the coating. Inspection of this graph will indicate the most promising fields for future tests.

$$Y_3 - STEEL WEIGHT LOSS (g/m2)$$

Both steel finishes combined.

Second degree equation -

 $Y_{3} = 25.76 + 3.42x_{1} + 9.89x_{2} - 13.43x_{3} + 0.14x_{1}^{2} + 3.98x_{2}^{2} - 9.97x_{3}^{2}$ SE b's = 0.70 0.41 0.41 0.58 0.38 0.38 0.76 + 1.83x_{1}x_{2} - 0.71x_{1}x_{3} - 3.50x_{2}x_{3} 0.51 0.72 0.72

Canonical form:

a) Center of system - S

xls		8.06	
x _{2s}	Π	-3.27	
		-0.39	·
х _{3s} Ү.3s	11	26.02	g/m²

b) Equation -

$$x_3 - 26.02 = -0.07x_1^2 + 4.41x_2^2 - 10.19x_3^2$$

c) Axes -

	(x <u>1</u> - 8.06)	$(x_2 + 3.27)$	$(x_3 + 0.39)$
Xı	0.9757	-0,2189	0.CO46
X2	-0.2181	-0,9681	0.1231
X_2 X_3	0.0234	0,1209	0.9924

The pattern here is the same as that found for Y_1 and Y_2 except that the center of the system is remote from the experimental conditions.

The coefficient of X_1 is not significant and was assumed to be zero for the purpose of plotting Figure 3. In addition, the bath temperature x_1 was set at 0.49 (460°C)(860°F). The effect of immersion time and aluminium content can be readily seen from the graph. Figures 2 and 3 may be compared directly since both have been plotted for a bath temperature of 460°C (860°F). The two plots are very similar. Y, - COATING THICKNESS (mm)

Both steel finishes combined.

Second degree equation -

 $Y_4 = 0.037 + 0.013x_2 - 0.021x_3 + 0.005x_2^2 - 0.010x_3^2 - 0.006x_2x_3$ SE b's = 0.002 0.001 0.002 0.001 0.003 0.003 Canonical form -

a) Center of system - S

$$\begin{array}{rcl} x_{2s} &=& -1.67 \\ x_{3s} &=& -0.53 \\ Y_{4s} &=& 0.03 \ \text{mm} \end{array}$$

b) Equation -

$$Y_4 - 0.031 = 0.006 x_2^2 - 0.011 x_3^2$$

c) Axes -

X2 X3

,	$(x_2 + 1.67)$	$(x_3 + 0.53)$
	0.9815 0.1915	-0.1915 0.9815

This relationship is shown in Figure 4.

Y5 - ALLOY THICKNESS (mm)

Both stoel finishes combined.

Second degree equation -

 $Y_{5} = 0.023 + 0.003x_{1} + 0.011x_{2} - 0.014x_{3} - 0.0002x_{1}^{2} + 0.005x_{2}^{2}$ SE b's = 0.001 0.001 0.001 0.001 0.001 0.001 - 0.013x_{3}^{2} + 0.001x_{1}x_{2} - 0.001x_{1}x_{3} - 0.002x_{2}x_{3} 0.001 0.001 0.001 0.001 Canonical form a) Center of system - S

$$\begin{array}{rcrr} x_{1s} &=& 3.08 \\ x_{2s} &=& -1.64 \\ x_{3s} &=& -0.48 \\ Y_{5s} &=& 0.022 \ \mathrm{mm} \end{array}$$

b) Equation -

$$x_5 - 0.022 = -0.0003x_1^2 + 0.0053x_2^2 - 0.0130x_3^2$$

c) Axes -

	$(x_1 - 3.08)$	$(x_2 + 1.64)$	$(x_3 + 0.48)$
X	0.9905	-0.1356	-0.0227
X1 X2 X3	-0.1339 0.0320	-0,9887 0,0646	0.0675 0.9974

The pattern here is similar to that found for Y_1 to Y_2 with the center of the system remote from the experimental conditions in regard to x_1 .

In plotting Figure 5, since X_1 is not significant, it was assumed to be zero. In addition x_1 , the bath temperature was taken at the zero level or 455°C (851°F).

Y₆ - RATIO, ALLOY THICKNESS: COATING THICKNESS (%)

Both steel finishes combined.

Second degree equation -

 $x_{6} = 61.32 + 4.35x_{1} + 5.07x_{2} - 25.33x_{3} + 1.22x_{1}^{2} + 1.14x_{2}^{2} - 34.56x_{3}^{2}$ SE b's = 0.92 0.54 0.54 0.77 0.50 0.50 1.00 + 0.86x_{1}x_{2} - 1.59x_{1}x_{3} + 1.71x_{2}x_{3}

Canonical form -

a) Center of system - S $x_{1s} = -1.54$ $x_{2s} = -1.37$ $x_{3s} = -0.37$ $Y_{6s} = 59.1\%$

b) Equation -

$$Y_6 - 59.13 = 0.79X_1^2 + 1.61X_2^2 - 34.60X_3^2$$

- 37 -

	$(x_1 + 1.54)$	$(x_2 + 1.37)$	$(x_3 + 0.37)$
Xn	0,6733	-0.7394	-0,0049
X_2^{\perp}	0.7392	0.6735	0,0009
X ₃	0.0226	-0.0242	0.9995

In Figures 6(a), 6(b) and 6(c), this relationship has been plotted. These figures have been plotted setting the aluminium content, the major variable, at 0, 0.1 and 0.2% and showing the effect of the two minor variables at each aluminium level.

These graphs show the relative effects of the three variables and indicate the most promising fields for future tests.

It should be remembered that in the original data, the high aluminium coatings (0,2%) failed to show an alloy layer of measureable thickness. An estimate of 3% was used in each case to facilitate the calculations. Figure 6(c) is therefore, only a rough approximation of the true relationship.

In Figure 6(d), the ratio of alloy thickness: coating thickness has been plotted against aluminium content of the bath for a bath temperature of $455^{\circ}C$ ($851^{\circ}F$) and an immersion time of one minute.

Yry - COATING DUCTILITY (Cupping Test)

Both steel finishes combined.

CODE - 1. Excellent, no cracking 2. Good, network of fine cracks

3. Fair, general cracking, small blocks

4. Poor, wide separation of medium-sized blocks

5. Very poor, general peeling of coating in large blocks.

Second degree equation -

$$X_{7} = 3.01 + 0.21x_{1} + 0.49x_{2} - 1.36x_{3} - 0.11x_{1}^{2} + 0.10x_{2}^{2} - 0.65x_{3}^{2}$$

SE b's = 0.06 0.04 0.04 0.05 0.04 0.04 0.07
+ 0.10x_{1}x_{2} + 0.08x_{1}x_{3} - 0.08x_{2}x_{3}
0.05 0.07 0.07

Canonical form -

a) Center of system - S

$$\begin{array}{rcl} x_{1s} &=& -0.59 \\ x_{2s} &=& -2.44 \\ x_{3s} &=& -0.93 \\ y_{7s} &=& 2.99 \end{array}$$

b) Equation -

$$Y_7 - 2.99 = -0.11X_1^2 + 0.12X_2^2 - 0.65X_3^2$$

c) Axes -

X1 X2 X3

	$(x_1 + 0.59)$	$(x_2 + 2.44)$	$(x_3 + 0.93)$
	-0.9720	0.2168	-0.0903
•	0.2210	0.9745	-0.0401
}	-0.0796	0.0594	0.9951

This relationship has been plotted in a manner similar to that used for Y_6 . Figures 7(a), 7(b), 7(c) have been plotted setting the aluminium content, the major variable, at 0, 0.1 and 0.2% and showing the effect of the two minor variables at each aluminium level.

 Λ comparison of the three graphs will show the relative effects of the three variables.

In Figure 7(d), the coating ductility has been plotted against aluminium content of the bath for a bath temperature of 455° C (851°F) and an immersion time of one minute.

 Y_{c} - COATING ADHERENCE (Bend Test)

Both steel finishes combined.

<u>CODE</u> - Minimum bend radius causing flaking (90° bend plus a 180° reverse bend).

1.	0.050-in.	5.	0.192-in.
2.	0.070 "	6.	0.252 "
3.	0.100 "	7.	0.320 "
4.	0.144 "	8.	0.400 "

Second degree equation -

 $Y_8 = 5.38 + 0.31x_1 + 0.69x_2 - 2.54x_3 - 0.09x_1^2 + 0.16x_2^2 - 1.89x_3^2$ SE b's = 0.08 0.05 0.05 0.07 0.05 0.05 0.09 - 0.07x_1x_2 - 0.02x_1x_3 - 0.04x_2x_3 0.06 0.09 0.09 Canonical form -

a) Center of system - S

$$\begin{array}{rcl} x_{1s} &=& 2.36 \\ x_{2s} &=& -1.79 \\ x_{3s} &=& -0.66 \\ y_{8s} &=& 5.97 \end{array}$$

b) Equation -

$$Y_8 - 5.97 = -0.10X_1^2 + 0.16X_2^2 - 1.89X_3^2$$

c) Axes -

X1 X2 X3

$(x_1 - 2.36)$	$(x_2 + 1.79)$	$(x_3 + 0.66)$
-0.9913	-0.1315	0.0073
-0.1316	0,991.3	-0.0095
0.0059	0.01.03	0.9999

Examination of the canonical form shows the third term (X_3) to be predominant and this term is mainly the effect of changes in aluminium content. The first term (X_1) which is mainly the effect of bath temperature is not significant. The second term (X_2) although significant is small and consists mainly of the effect of immersion time.

By assuming that X_1 is zero and by setting bath temperature at the zero level (455°C)(851°F), Figure 8 has been plotted showing the effects of immersion time and aluminium content changes on adherence.

Yo - SPANGLE SIZE

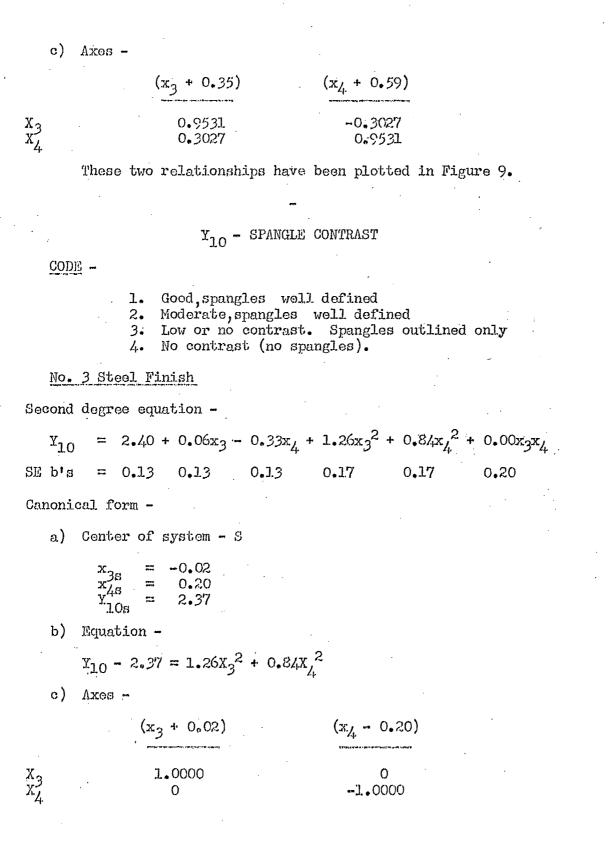
CODE -

- L. Large
- 2. Medium
- 3. Small.
- 4. No spangle

No. 3 Steol Finish

Second dogree equation -

 $Y_9 = 2.37 - 0.17x_3 - 1.03x_4 + 1.42x_3^2 + 0.13x_4^2 - 0.50x_3x_4$ SE b's= 0.16 0,16 0.16 0.21 0.21 0.28 Canonical form a) Center of system - S $x_{3s} = 1.13$ $x_{4s}^{3s} = 6.11$ $Y_{0s}^{3s} = -0.86$ b) Equation - $Y_9 + 0.86 = 1.47 X_3^2 + 0.08 X_1^2$ c) Axes -(x₃ - 1.13) (x₄ - 6.11) Хз ХД 0.9830 -0.1837 0.9830 No. 5 Steel Finish Second degree equation - $Y_9^1 = 3.66 - 0.11x_3 - 0.69x_4 + 0.27x_3^2 - 0.44x_4^2 - 0.50x_3x_4$ SE b's = 0.15 0.15 0.15 0.20 0.20 0.27 Canonical form a) Center of system - S $\begin{array}{rcl} x_{38} & = & -0.35 \\ x_{48} & = & -0.59 \\ y_{98}^{1} & = & 3.88 \end{array}$ b) Equation - $Y_9^1 - 3.88 = 0.34 X_3^2 - 0.52 X_4^2$



No. 5 Steel Finish

Second degree equation -

 $Y_{10}^1 = 3.64 + 0.03x_3 - 0.31x_4 + 0.35x_3^2 - 0.06x_4^2 + 0.00x_3x_4$ SE b's = 0.13 0.13 0.13 0.17 0.17 0.20

Canonical form -

a) Center of system - S

$$x_{3s} = -0.08$$

 $x_{4s}^2 = -2.42$
 $y_{10s}^2 = 4.01$

b) Equation -

$$Y_{10}^1 - 4.01 = 0.35 X_3^2 - 0.06 X_4^2$$

c) Axes -

Х3 Х4

$(x_3 + 0.08)$	(x ₄ + 2.42)
1.0000	0
0	-1,0000

These two relationships have been plotted in Figure 10. For the No. 5 steel finish, very little variation in spangle contrast was experienced. As a result, only one contour line falls within the experimental region. An improvement in contrast is indicated in the direction of the arrow marked 3.

Y₁₁ - BRIGHTNESS (Photometer Reading)

Both steel finishes combined.

CODE -

1. 0 - 1.25
2. 1.5 - 2.75
3.
$$3.0 - 4.25$$

4. 4.5^+

Second degree equation -

$$Y_{11} = 2.40 + 0.36x_1 - 0.79x_3 - 0.46x_4 + 0.17x_1^2 + 0.47x_3^2 - 0.32x_4^2$$

SE b's = 0.13 0.07 0.11 0.11 0.07 0.14 0.14
+ 0.00x_1x_3 + 0.00x_1x_4 - 0.46x_3x_4
0.13 0.13 0.18

Canonical form -

a) Center of system - S

$$\begin{array}{rcrr} x_{1s} &=& -1.05 \\ x_{3s} &=& 0.37 \\ x_{4s} &=& -0.98 \\ y_{11s} &=& 2.29 \end{array}$$

b) Equation -

$$x_{11} - 2.29 = 0.19 x_1^2 + 0.53 x_3^2 - 0.38 x_4^2$$

c) Axes -

X1 X3 X4

$(x_1 + 1.05)$	(x ₃ - 0.37)	$(x_4 + 0.98)$
1.0000	0	0
-0.0008	-0.9657	0.2596
0.0002	-0.2595	-0.9657

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Examination of the canonical form shows that the effect of factor X_1 is small and is almost entirely due to x_1 . In plotting Figure 11, it was therefore assumed that X_1 is zero, and x_1 was set at the zero level (455°C) (851°F). Only the effects of x_3 and x_4 upon brightness are shown.

An increase in bath temperature within the range of experimental conditions will tend to increase the photometer readings.

Y₁₂ - ROUGHNESS

CODE -

1. Very smooth

- 2. Moderately smooth
- 3. Fine to moderately rough, sandpaper texture
- 4. Rough texture

No. 3 Steel Finish

Second degree equation -

 $Y_{12} = 1.95 + 0.29x_1 + 0.38x_2 - 0.25x_3 + 0.06x_1^2 + 0.06x_2^2 + 1.05x_3^2$ SE b's = 0.14 0.09 0.09 0.12 0.08 0.08 0.16 + 0.06x_1x_2 + 0.25x_1x_3 + 0.00x_2x_3 0.10 0.15 0.15 Canonical form a) Center of system - S Equation b) $Y_{12} - 1.20 = 0.08 X_1^2 + 0.02 X_2^2 + 1.07 X_3^2$ c) Axes - $(x_1 + 2.27)$ $(x_2 + 2.01)$ (x₃ - 0.39) 0.6110 0.7820 0.1228 0.7890 -0.0653 -0.6144 0.0038 -0.1043 0.9924 No. 5 Steel Finish Second degree equation - $Y_{12}^{1} = 2.84 + 0.37x_{1} + 0.26x_{2} - 0.50x_{3} - 0.11x_{1}^{2} + 0.02x_{2}^{2} + 0.39x_{3}^{2}$ SE b's = 0.17 0.10 0.10 0.14 0.09 0.09 0.18

$$- 0.15x_1x_2 + 0.42x_1x_3 + 0.00x_2x_3$$

0.12 0.17 0.17

Canonical form -

. ·

a) Center of system - S

$$\begin{array}{rcl} x_{1s} &=& 1.67\\ x_{2s} &=& -0.64\\ x_{3s} &=& -0.25\\ y_{1} &=& 3.13 \end{array}$$

b) Equation -

 $x_{12}^{1} - 3.13 = 0.21x_{1}^{2} + 0.04x_{2}^{2} + 0.46x_{1}^{2}$

c) Axos -

X1 X2 X3

(x ₁ - 1.67)	$(x_2 + 0.64)$	(x ₃ + 0.25)
	genagenen i selon articritike a energintike	and the second and a life of a life of the second of the
-0.9012	-0.2951	0.3173
0.2582	-0.9538	-0.1541
0.3479	-0.0565	0.9358

For both steel finishes X_2 is not significant and has been assumed to be zero for plotting purposes. In addition, immersion time (x_2) , although it does have an effect upon roughness, has been set at the zero level (1 min) to facilitate the plotting of Figure 12.

In the canonical form X_3 is the most important variable in both cases and its effect is related mainly to the aluminium content of the bath.

Y13 - PERCENT GAIN IN AVERAGE DIFFUSIVITY AFTER THE HUMIDITY TEST

No. 3 Steel Finish

Second degree equation -

 $Y_{13} = 7.23 - 6.01x_1 + 10.88x_3 + 8.07x_4 - 7.54x_1^2 + 19.96x_3^2 + 17.78x_4^2$ SE b's = 6.66 3.94 5.57 5.57 2.58 5.16 5.16

$$-1.11x_1x_3 + 4.60x_1x_4 + 6.10x_3x_4$$

Canonical form -

a) Conter of system - S

$$\begin{array}{rcl} x_{1s} &=& -0.42 \\ x_{3s} &=& -0.26 \\ x_{4s} &=& -0.13 \\ y_{12s} &=& 6.53\% \end{array}$$

b) Equation -

$$x_{13} - 6.53 = -7.77x_1^2 + 22.14x_3^2 + 15.84x_4^2$$

c) Axes -

	$(x_1 + 0.42)$	$(x_3 + 0.26)$	$(x_4 + 0.13)$
xı	-0,9952	-0.0301	0.0932
x ₃	0,0300	0.8115	0.5836
x ₄	-0,0933	0.5836	-0.8067

No. 5 Steel Finish

Second degree equation -

$$Y_{13}^{1} = -7.12 - 5.13x_{1} + 14.68x_{3} + 5.08x_{4} - 3.69x_{1}^{2} + 35.77x_{3}^{2}$$

SE b's = 6.21 3.68 5.20 5.20 3.40 6.81
+ 20.56x_{4}^{2} + 0.66x_{1}x_{3} - 0.62x_{1}x_{4} + 0.92x_{3}x_{4}
6.81 6.37 6.37 9.01

Canonical form -

a) Center of system - S

$$\begin{array}{rcl} x_{1s} &=& -0.70 \\ x_{3s} &=& -0.20 \\ x_{4s} &=& -0.13 \\ y_{13s} &=& -7.09\% \end{array}$$

b) Equation -

$$x_{13}^{1} + 7.09 = -3.69 x_{1}^{2} + 35.79 x_{3}^{2} + 20.55 x_{4}^{2}$$

c) Axes -

X₁ X₃ X₄

(x <u>1</u> + 0.70)	$(x_3 + 0.20)$	$(x_4 + 0.13)$
-0,9999	0.0085	-0.0128
0.0082 0.0133	0.9995 0.0301	0.0302 -0.9995

Examination of the canonical forms for both steel finishes shows that X_1 is small and not significant and its effect is related mainly to x_1 the bath temperature. To facilitate plotting of the two graphs in Figure 13 the variable X_1 was assumed to be zero and the bath temperature x_1 taken as -0.7 (448°C)(838°F). Y₁₄ ~ CORROSION INDEX - WHITE, AFTER HUMIDITY TEST

There were no significant factors found for either of the two steel finishes. There is, therefore, no Figure 14.

Y₁₅ - WEIGHT GAIN FROM CORROSION AFTER HUMIDITY TEST (mg) Both steel finishes combined.

Second degree equation -

 $Y_{15} = 38.8 + 3.7x_1 - 1.0x_2 + 3.3x_3 + 6.2x_4 - 4.4x_1^2 - 4.3x_2^2 + 12.5x_3^2$ 1.6 1.6 2.5 2.5 3.3 1.7 SE b's = 3.51.7 $-7.7x_{4}^{2} + 1.9x_{1}x_{2} - 2.3x_{1}x_{3} - 1.5x_{1}x_{4} - 6.9x_{2}x_{3} + 2.2x_{2}x_{4} - 0.6x_{3}x_{4}$ 4.3 3.3 2.1 3.0 3.0

Canonical form -

a) Center of system - S

$$\begin{array}{rcl} x_{1s} &=& 0.39 \\ x_{2s} &=& 0.12 \\ x_{3s} &=& -0.05 \\ x_{4s} &=& 0.38 \\ y_{15s} &=& 40.6 \\ \end{array}$$

b) Equation -

$$Y_{15} - 40.6 = -3.99 X_1^2 - 4.88 X_2^2 + 13.28 X_3^2 - 8.29 X_4^2$$

c) Axes -

X12 X2 X3 X4

(x ₁ - 0.39)	(x ₂ - 0,12)	(x ₃ + 0.05)	(x ₄ - 0.38)
and and a second property of the second s	4-444	alle stara and a star a	The state of the s
0.8156	0.5521	0.1730	-0.0192
0.5233	-0.7338	-0.1175	-0.4171
0.0742	0.1966	-0.9774	0.0225
0.2449	-0.3373	-0.0284	0.9085

ţ.

With a four-factor system, as with a three-factor system, some assumptions have to be made and some conditions set, in order to plot the important factors in two dimensions.

Canonical factors X_1 and X_2 are small and of doubtful significance compared to X_3 and X_4 ; therefore, by assuming that X_1 and X_2 are not different from zero and by setting x_1 and x_2 both at the zero level (i.e., a bath temperature of 455° C (851° F) and a one-minute immersion time), a two-dimensional system in x_3 and x_4 may be plotted. This plot only roughly represents the true.relationship but is sufficient for our purpose. Examination of the canonical form will give a better idea of the whole system.

Y16 - WEIGHT LOSS FROM CORROSION AFTER HUMIDITY TEST (mg)

Both steel finishes combined.

Second degree equation -

 $Y_{16} = 60.4 + 3.7x_2 + 23.2x_3 + 6.2x_2^2 + 9.6x_3^2 - 7.8x_2x_3$ SE b's = 3.6 2.5 3.6 2.3 4.7 4.4 Canonical form -

a) Center of system - S

x _{2s}	==	-1.42
	=	-1.80
x3s Y16s	=	36.9 mgs

b) Equation -

$$x_{16} - 36.9 = 3.65 x_2^2 + 12.14 x_3^2$$

c) Axes -

X X3

$$(x_2 + 1.42) \qquad (x_3 + 1.80) \\ \hline 0.8334 \qquad 0.5527 \\ \hline 0.5527 \qquad -0.8334$$

This relationship is shown in Figure 16.

 Y_{17} - CORROSION INDEX - WHITE (24 hr) - AFTER WATER FILM TEST $Y_{17} = 2.49 - 0.23x_3 + 0.28x_3^2$

Here only one factor was significant and it has been plotted directly in Figure 17.

Y18 - CORROSION INDEX - WHITE (48 hr) - AFTER WATER FILM TEST

Both steel finishes combined. Second degree equation - $Y_{18} = 2.70 + 0.05x_2 + 0.17x_3 - 0.08x_2^2 + 0.31x_3^2 - 0.09x_2x_3$

 $SE b's = 0.08 \quad 0.06 \quad 0.08 \quad 0.05 \quad 0.11 \quad 0.10$ Canonical form -

a) Center of system - S

x2s X33 Yigs

b) Equation -

 $Y_{18} - 2.69 = -0.09 X_2^2 + 0.32 X_3^2$

c) Axes -

X2 X3

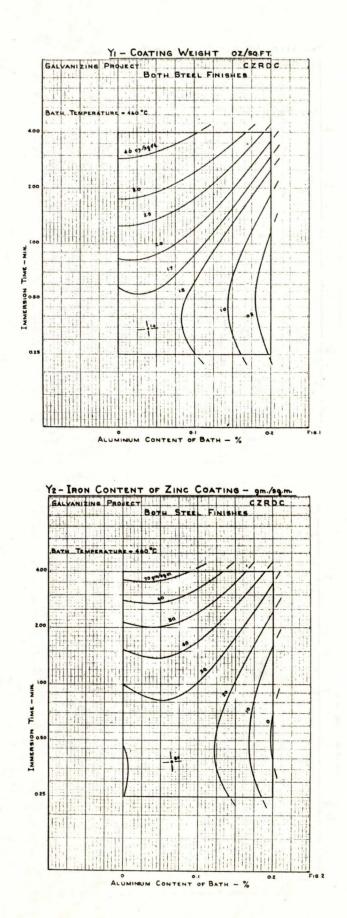
$(x_2 - 0.42)$	$(x_3 + 0.20)$
0.9934	0 , 1150
0.1150	-0,9934

4. 4.

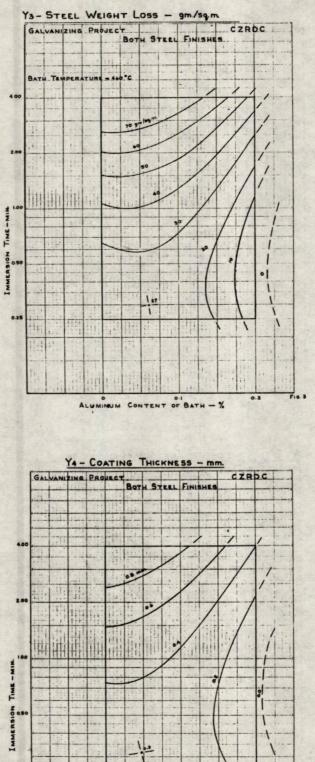
This relationship is shown in Figure 18.

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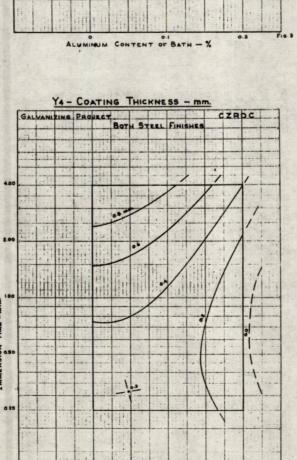
(Figures 1-18 follow, (on pages 51 to 62



- 51 -



- 52 -

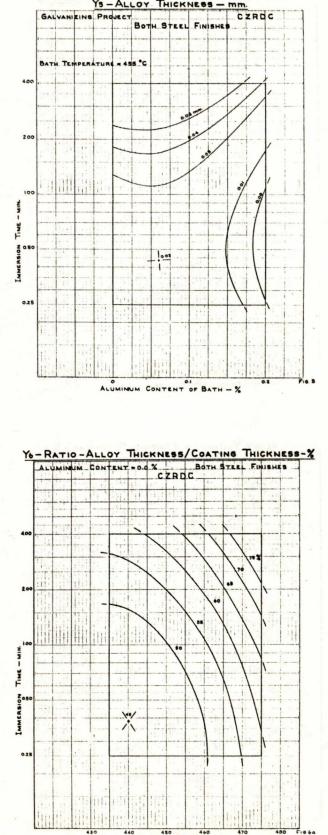


ALUMINUM CONTENT OF BATH - %

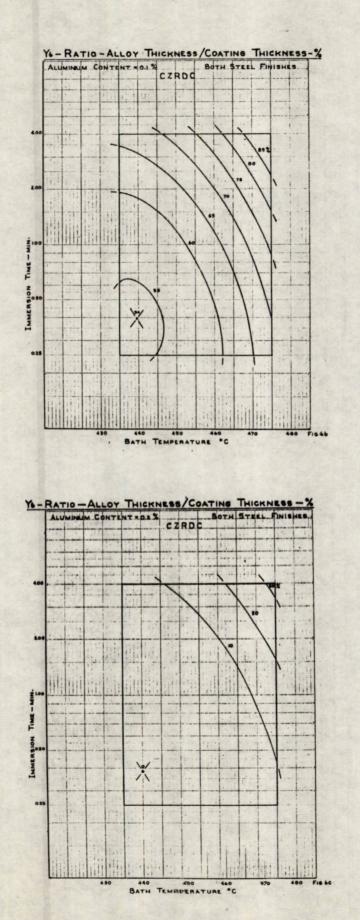
FIG.4

0.2

YS-ALLOY THICKNESS - MM.

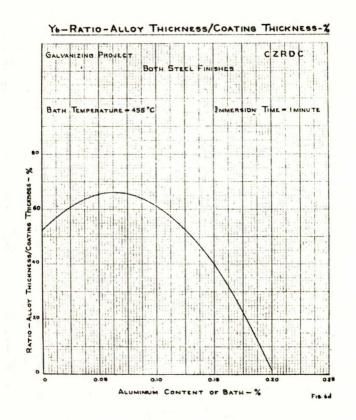


440 450 460 BATH TEMPERATURE °C

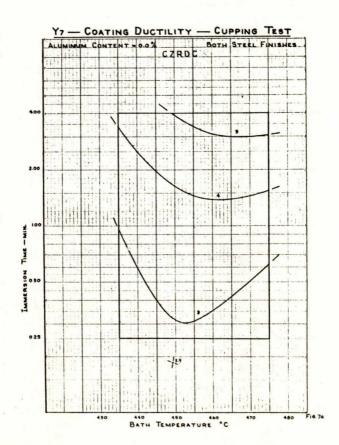


- 54 -

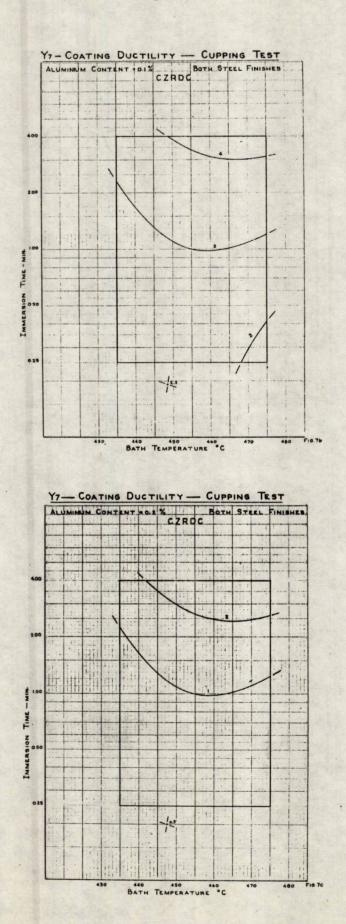
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- 55 -



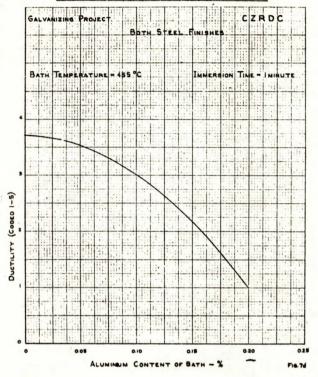
- 56 -

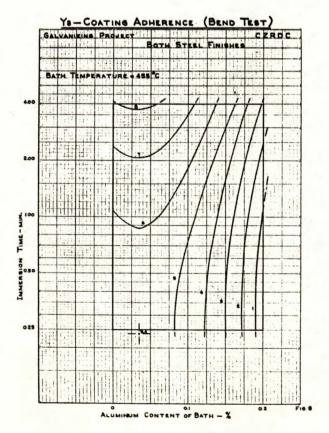
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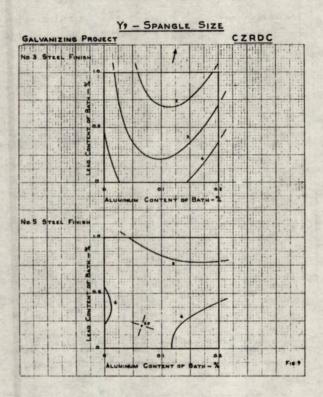
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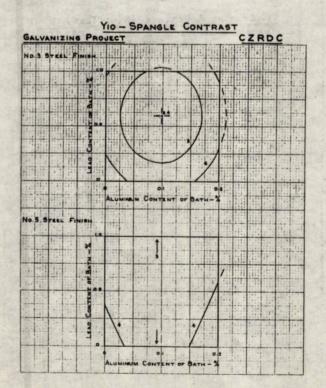
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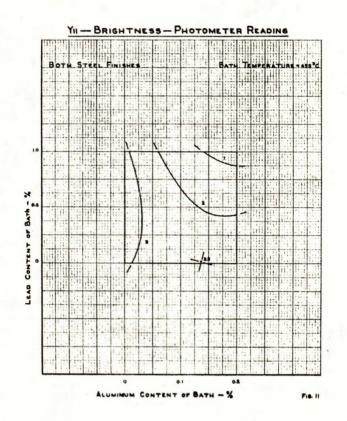


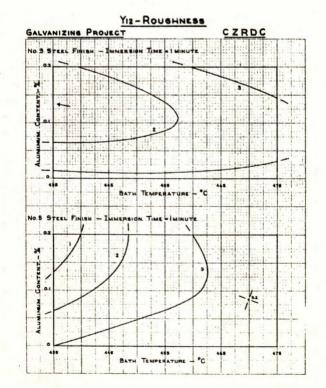
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F18 10

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Fiel2

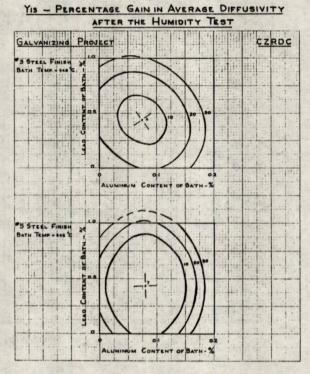


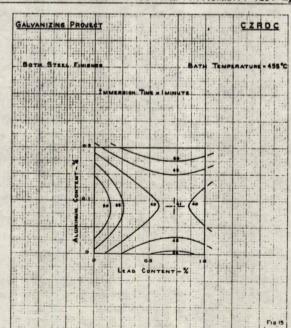
FIG 13

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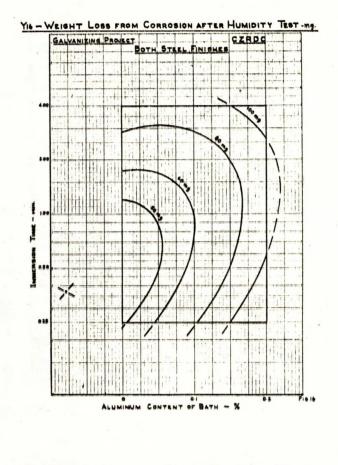
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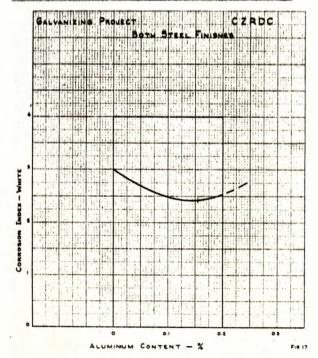
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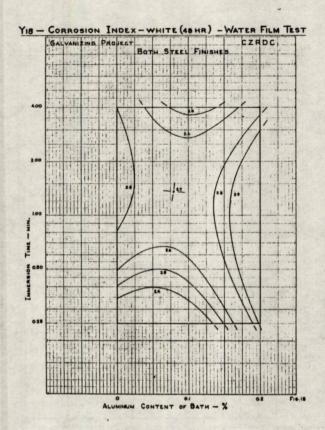
YIS - WEIGHT GAIN FROM CORROSION AFTER HUMIDITY TEST-mg.







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