OTTAWA

July 14th, 1941.

<u>R E P O R T</u>

of the

ORE DRESSING AND METALLURGICAL LABORATORIES.

Investigation No. 1049.

Flotation Concentration of Copper-Lead-Zinc-Silver Ore from the Shefford Gold Mines Company Limited, Lake Megantic, Frontenac County, Quebec.

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Shipment:

A bag of ore, weighing 117 pounds, was received on May 20th, 1941, from the Shefford Gold Mines Company Limited, Lake Megantic, Frontenac county, Quebec. The shipment was submitted by W. A. Marois, Box 153, 105A Laval Street, Lake Megantic, Quebec.

Sample:

The sample was taken from a vein on Lot No. 10, Range 6, in the township of Ditchfield, Frontenac county, Quebec.

Characteristics of the Ore:

Six polished sections were prepared and examined microscopically for the purpose of determining the character of the ore.

July 16th. 1941.

Gangue -

The gangue is composed of slightly fractured white quartz, which bears a few local, light-brown stains of iron oxides, and encloses narrow streaks and small patches of dark greenish-grey material; the latter probably represent remnants of country rock.

Metallic Minerals -

Metallic mineralization is moderately abundant and consists of a rather coarse-grained admixture of pyrite, chalcopyrite, galena and sphelerite in which pyrite preponderates as disseminated grains with the other sulphides to a large extent filling the interstices between them. In places the iron sulphide is fractured and the fractures filled with gangue, more rarely with galena and chalcopyrite. It also contains numerous inclusions of gangue and occasional small grains of the other sulphides.

Chalcopyrite, galena, and sphalerite occur in gangue as small irregular masses and disseminated grains, which, as already mentioned, are frequently interstitial to grains of pyrite. These three sulphides are often intimately associated and each encloses inclusions of the others. Those of chalcopyrite in sphalerite usually take the form of tiny

Megantic, Cushes . (Continued on next page)

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(Characteristics of the Ore, cont'd) -

dots which are scattered at random in some places and arranged in rows in other places. Although each mineral is locally abundant, its total quantity is relatively small.

Sampling and Assaying:

After crushing and grinding to minus 14 mesh, a sample for assay was cut by the standard method. The sample assayed as follows:

of iron oxides, and encloses nigrow streaks and small patches

	WELL LE	Gold	-	0.01	ounce per ton of ore.
t - t	The c	Silver		0.41	11 htly fractured
Sette W		Copper		0.50	per cent.
white, qua	rtz. B	Lead .	10 How	0.36	. 11 mht brown stains.
1		Zinc	-	0.30	

of dark grantal <u>EXPERIMENTAL TESTS</u>. Les probably represent reanants of the gold and silver values in the ore received were respectively 0.01 and 0.41 ounce per ton of ore. The recovery of these metals by amalgamation on an ore as low and as this would not be economical. The test work consisted of flotation concentration. A bulk concentrate was obtained which assayed as follows; silver, 1.78 ouncesper ton; copper, 3.09 per cent; lead, the 1.48 per cent; zinc, 2.94 per cent. The per cent recoveries were: silver, 72.9; copper, 93.1; lead, 81.3; zinc, 87.4.0 (Test No. 1).

A copper-lead concentrate was obtained which assayed 6.80 ounces of silver per ton; 14.55 per cent copper, 8.06 per cent lead and 5.61 per cent zinc. The Cu-Pb concentrate to contained 45.7 per cent of the silver, 75.2 per cent of the ly copper and 77.5 per cent of the lead in the ore. The Those copper-lead ratio of concentration was 36.4 into 1. The zinc concentrate obtained assayed 19.0 per cent zinc, 3.98 per

(Continued on next page)

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(Experimental Tests, cont'd) -

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cent copper and 0.42 per cent lead. The zinc concentrate contained 54.7 per cent of the zinc in the ore. The ratio lativoly mall. of concentration was 66.7 into 1.

Details of the tests follow:

Test No. 1. - Bulk Flotation.

awan, a

A sample of minus 14 mesh product was ground to 86 per cent minus 200 mesh with 2.0 pounds of soda ash and 0.10 pound of Aerofloat No. 25 per ton of ore. The pulp was transferred to a flotation cell and 1.0 pound of copper sulphate per ton of ore was added and conditioned 5 minutes. Then 0.15 pound of Reagent 301 and 0.12 pound of pine oil per ton of ore were added and the froth skimmed off for 10 minutes pectively 0.82 and 0.41 dutos per teh of one. The The concentrate was cleaned by refloating. No.

additional reagents were used.

	Weight,: Assays, :						Distribution,					
Product :		: <u>0z./</u> : Au			Per cent Pb : Zn		and an end of the second se	cent Cu : P	b :	Zn		
Conc. :	100.00	0.009	0.39	0.53	0.29.0.54	100.0 51.3	100.0	100.0:10 93.1: 8	0.0:	100.0 87.4		
Cleaner tailing: Tailing	2.59 81.29	0.02	1.62	1.11	0.87:0.76	5.5	10.6:	5.4: 1.5: 1	7.7:	3.6		

Ratio of concentration = 6.2:1. which assared 6.60 pH of flotation tailing solution = 7.7.

a llotation was made.

cent land the Blat part odes what. Tas, 6d-20 componizato contained as . There are allies, W. 3 por cont of the appper and W7. Lappart char the last in the ove. The

concentrate obtained he's had le. of per contained, 5,98 per

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(Experimental Tests, cont'd) -

<u>Test No. 2. - Selective Flotation</u>. Contained In this test a sample of ore was ground to 86°100 per cent minus 200 mesh with the following reagents:

Details of the berts follows Lb./ton of ore

10

	Soda ash		S	4.0	
1 E .	Sodium cy	anide	Flatssion	0.15	
	Zinc sulp	hate	• A	0.30	
A gam	Aerofloat	No. 25	shiproduct	0.10	sound.

The pulp was transferred to a flotation cell and copper and lead minerals were floated for 5 minutes. Then the pulp was conditioned for 10 minutes with 7.0 pounds of lime per ton. The sphalerite was floated by adding 0.20 pound of copper sulphate and 0.02 pound of Aerofloat No. 213 per ton of ore. The copper-lead concentrate was cleaned by refloating. No additional reagents were used.

	:Weight,	Neight,: Assays : Distribu						ution,					
Product	: per	: Oz.	Oz./ton		: Per cent :				per cent				
	: cent	: Au	: Ag	: Cu	: F	b :	Zn	Ag	1.1	Bu	L(n)	Pb	Zn
Feed	100.00		0.41	. 0.5	3:0.	29:	0.52	100.	:	100.	0:1	00.00	100.
Cu-Pb conc.		0.03	:6.80	:14.5	5:8.	06:	5.61	:45.	7:	75.	2:	77.5	29.
Cu-Pb	0.00.0.9	00:0.	0.01	1:0.2	Stok.	64.	00.00	100.	C 1.]	.00.	0.10	20,01	9.00%
cleaner	CILDIO.	12 12 -1	101.2.01	1.4.1	128.14	02.	all a set	122	3.	· alia à .	1.8.1	37 . 01	1.1.1.0
tailing	: 0.55	:)	:3.90*									3.5	
Zn'eone.	: 1.50		fil the other	: 3.9	8:0.	42:1	19.00	po.	0.	11.	2:		54.
Tailing :	95.20	:0.005	:0.15	: 0.0	6:0.	.05:	0.07:	34.	8:	10.	7:	16.7	12.

additional reagents were used. Results of Selective Flotation.

Calculated value. PH of the Cu-Pb flotation solution =

Ratios of concentration:

Copper-lead = 36.4:1.Zinc = 66.7:1.

9.3.

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- Page 6 -

(Experimental Tests, contid) -

Conclusions:

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It would not be economical to treat an ore by amalgamation for the recovery of gold and silver which is as low in these metals as the ore received.

Bulk concentration of the ore by flotation gave a concentrate which assayed 1.78 ounces of silver, 3.09 per cent copper, 1.48 per cent lead, and 2.94 per cent zinc. This grade of concentrate is too low for economical smelting treatment. Then the pulp was Selective flotation gave a copper-lead concentrate which is a commercial grade. The zinc concentrate was low grade, assaying 19.00 per cent zinc.

	1		Results of	C Solad	time 15th	tatten	19 18	34 3	19.	
	1	coopent,	1	A L So P	and and a second		· · 1	and C.C. i. D.	uticon,	,
·	Product	1. 2962	: Va. 1991	La come conserve	141.00M	Summinister ?	an and a second second	SCP.	dont.	
		cent	I AN I A	E CIL	20:	212 :	hills in	GIL :	E D C	LAID AND AND
	Food Cu-Pb cloumpr tall JFK:G 2n conc.	• 2,73 • 0,55		0000	000000	5.61: 8.69:	45.7:	2.91	77.5: 3.5:	29.7
	Tailing									
	ana ang sa	Gelculat	od value.	nu ta mita ain.	Second Second Second	2 - MARIN & CRIME 23	(BB) (A) AND (A) AND (A)	PR 2. deta 2010 1. 193	felle angle, o sportige of sol	A PLATE AN ADALOGY OF

pH of the Cu-Pb flotation solution = 9.3.

Copacy-lead = 36.4:1.

. Ratios of Abbentration:

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