## Report No. 242

The concentration of the copper ore ("F" orebody) of the Noranda Mines Ltd. Rayn, Quebec by C. S. Parsons

<u>Shipments:</u> Two shipments of ore were received from Noranda Mines Ltd. Qouyn, Quebed. Lot No. 1, weight 118 lbs. was received September 9, 1925, and Lot No. 2, weight 1500 lbs. January 27, 1926 Characteristics and Analyses: containing chalcopyrite, pyrite, pyrrhotite and

gold. The chalcopyrite is disseminated, and fine grinding is necessary to free it. Analyses of the two lots were as follows:

Copper Gold	1.55% 0.25 oz/1	Arsenic	0.03	
Silver	0.28 " trace	Sulphur Silica	39.86	
Zino	0.28 %	Market State		

Lot No. 2 Copper 1.6% Gold 0.3 oz/ton Purpose of The samples were submitted for the purpose of <u>Experimental work:</u> determining whether this type of ore could be

concentrated. A concentrate was desired which would contain the gold and copper; the greater portion of the pyrite and pyrrhotite to be eliminated in a tailing product. Also, if possible, to determine what sulphides the gold is associated with.

Flotation Tests Lot No. 1



Lot No. 1

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Product	Weight		Assay Per cent of values				values	
		Au oz	Ag oz	Cu 7	re %	Au	Ag	Ou
Magnetic	45.4	0.07	0.13	0.62	57.73	12.0	22.0	19.1
Concentrate	7.2	1.66	0.86	12.04	40.28	44.5	23.1	138.9
Middling	10.2	0.41	0.46	1.83	44.72	15.4	17.2	12.4
Tailing	37.4	0.20	0.27	0.38	46.45	28.1	37.7	9.6
	roducts	0.27	0.27	1.47	50.95			

Flotation tests Lot No. 2

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12	Concentrate Tailing	40.6 59.4	3.56 0.14	0.57	94.5 5.5	83.1 16.9
13	Concentrate Tailing	26.8	5.13	0.68	90.0 10.0	86.4 13.6
14	Concentrate Tailing	49.4 50.6	3.0	0.40 0.12	97.5	76.3 23.7
15	Concentrate Tailing	71.8 78.2	6.12	0.78 0.15	87.8 12.2	59.4 40.6
16	Concentrate Tailing	24.6 75.4	5.45	0.59 0.10	63.1 36.9	65.7 34.3
17	Concentrate Tailing	21.3 78.7	5.53	0.67 0.08	55.4 44.6	72.1 27.9
18	Concentrate Tailing	23.9	5.76	0.64 0.08	64.7 35.3	71.5 28.5
19	Concentrate Tailing	25.3 76.1	6.70 0.63	0.61 0.14	78.23	52.08 37.92
20	Concentrate Tailing	29.2 70.8	5.58	0.56 0.08	77.35 22.65	74.09 25.91
- 21	Concentrate Tailing	32.3	6.38 0.19	0.54 0.06	94.0 6.0	81.0
23	Concentrate Tailing	46.5 53.5	4.50 0.14	0.46 0.11	96.5 3.5	78.5

No, 12	1000 grams ore 45 Soda ash P.T.&T. #400 Xanthate	minutes in mill 3.0 grams 6 drops 0.1 gram added to cells
No. 13	1000 grams ore 45 Soda ash P.T.& T #400 Xanthate	minutes in mill 3.0 grams 6 drops 0.1 gram added to cells
	Cyanide Copper sulphate	0.5 " added to ball mill 2 c.c.

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No. 14	1000 grams ore 45 minutes in mill Soda ash 3 grams AuPonts crecsote 6 drops Cresylic acid 8 drops
No. 15	1000 grams ore 45 minutes in millSoda ash3 gramsDuponts cresote8 dropsCresylic acid8 dropsCyanide0.5 gramsXanthate0.1 " added to cells
No. 16	1000 grams ore 60 minutes in mill (-20 mesh) Barretts #4 5 drops Cresylic acid 5 drops Soda ash 3 grams
No. 17	1000 grams ore 60 minutes in mill (-20 mesh) Soda ash Duponts crecsote 5 drops Cresylic acid 5 drops
No. 18	1000 grams ore 60 minutes in mill (-20 mesh) Water gas tar 60) Cresylic acid 20) 10 drops Coal tar creo 20)
No. 19	1000 grams ore -20 mesh 60 minutes in mill Mixture as in 18 10 drops Soda ash 2 grams Cyanide .15 "
No. 20	1000 grams ore -20 mesh 60 minutes in mill Mixture as in 18 10 drops Soda ash 3 grams
No. 21	1000 grams ore -20 mesh 60 minutes in mill Mixture as in 18 10 drops Soda ash 3 grams Cyanide .15 " After running few minutes added 1 c.c. copper sulphate - copper coloured froth came
No. 23	1000 grams ore -20 mesh 20 minutes in mill Soda ash 3 grams Cyanide .2 " Mixture as in 18 10 drops Xanthate .1 gram added to cell Copper sulphate 1.5 cc " "

<u>Conclusions:</u> From the results of the magnetic separation test it is evident that the pyrrhotite does not carry the gold. Attention is drawn to test Nol 12 on Lot 1, where 78% of the gold was recovered but only 64% of the copper, and to the fact that when **x** lime is used a high copper concentrate can be obtained with an excellent recovery of the copper, but without a corresponding recovery of gold. These results prove that the gold is not entirely associated with the copper.

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One very important point was ascertained, namely that the tests conducted on the ore immediately after it was received gave very good

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results - reference to tests Nos. 1-5 Lot 2 - but as the ore became older the results fell off. This characteristic was very noticeable.

Special attention is drawn to the use of cyanide with soda ash. Thests Nos. 4, 5, 13, 15, 19 and 23 were run using cyanide. In the last three of these tests the ore was showing decided signs of oxidation. The results in test nos. 4, 5, and 13, are very encouraging, and warrant the conducting of a tonnage scale test using these reagents.

The experience with this ore shows that the samples used in flotation tests must be sxlected from freshly mined ore and guarded in every possible against oxidigation.

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