

Ser

3

622(21)

212 te

Energy, Mines and Énergie, Mines et Resources Canada Resources Canada

CANMET

Canada Centre Centre canadien for Mineral de la technologie and Energy des minéraux Technology et de l'énergie

CANIMET LIBRARY JUN 25 1918 HEQUE H STREET **BIBLIO** CANADA 555 BOO OTTAWA C ONT. OG1

CERTIFICATION OF REFERENCE IRON ORE SCH-1 FOR SODIUM AND POTASSIUM

R. Sutarno, D.J. Charette, W.S. Bowman and G.H.Faye



MINERALS RESEARCH PROGRAM MINERAL SCIENCES LABORATORIES CANMET REPORT 78-5

Minister of Supply and Services Canada 1978 O Ministre des Approvisionnements et Services Canada 1978

Available by mail from: En vente par la poste:

Printing and Publishing Supply and Services Canada, Ottawa, Canada K1A 0S9

CANMET Energy, Mines and Resources Canada. 555 Booth St., Ottawa, Canada K1A 0G1

Approvisionnements et Services Canada, Ottawa, Canada K1A 089 CANMET Énergie, Mines et Ressources Canada, 555, rue Booth Ottawa, Canada K1A 0G1

Imprimerie et Édition

or through your bookseller. ou chez votre libraire.

Catalogue No. M38-13/78-5 ISBN 0-660-01808-X Price: Canada: \$1.00 Other countries: \$1.20

Price subject to change without notice.

N^o de catalogue M38-13/78-5 ISBN 0-660-01808-X

Prix: Canada: \$1.00 Autres pays: \$1.20

Prix sujet à changement sans avis préalable.

CANMET REPORT 78-5

CERTIFICATION OF REFERENCE IRON ORE SCH-1 FOR SODIUM AND POTASSIUM

Ъy

R. Sutarno*, D.J. Charette**, W.S. Bowman*** and G.H. Faye*

SYNOPSIS

Reference iron ore SCH-1, previously certified for nine constituents, has been certified for sodium and potassium. Analytical results were obtained from 53 contributors in Canada and abroad, most of whom used an atomic absorption procedure that is a candidate method of the International Organization for Standardization. A statistical treatment of the data yielded recommended values which are 0.019% and 0.026% for sodium and potassium respectively.

*Research Scientists, **Chemist, ***Technologist, Mineral Sciences Laboratories, Canada Centre for Mineral and Energy Technology, Department of Energy, Mines and Resources, Ottawa, Canada.

RAPPORT DE CANMET 78-5

HOMOLOGATION DU MINERAI DE FER DE REFERENCE SCH-1 POUR LE SODIUM ET LE POTASSIUM

par

R. Sutarno*, D.J. Charette**, W.S. Bowman***, et G.H. Faye*

SYNOPSIS

Le minerai de fer de référence SCH-1 ayant été homologué auparavant pour neuf composants est maintenant homologué pour le sodium et le potassium. Des résultats d'analyse ont été obtenus de 53 contribuants au Canada et à l'étranger, dont la plupart emploie un procédé d'absorption atomique reconnu par l'Organisation internationnale de normalisation. Après le traitement statistique des données, on a obtenu des valeurs recommandées de 0.019% pour le sodium et de 0.026% pour le potassium.

*Chercheurs scientifiques, **Chimiste, ***Technologue, Laboratoires des sciences minérales, Centre canadien de la technologie des minéraux et de l'énergie, Ministère de l'Energie, des Mines et des Ressources, Ottawa, Canada.

INTRODUCTION

In 1975, reference iron ore SCH-1 was certified for nine constituents¹. However, only provisional values for sodium and potassium were assigned because of the lack of consensus in the results obtained by atomic absorption and flame emission methods in a "free-choice" interlaboratory analytical program. The certification factors for these elements were 3.9 and 4.1 for sodium and potassium respectively, both close to 4, the maximum value for certifiability².

Because of the importance of even low concentrations of sodium and potassium in iron ore metallurgy, it was decided to pursue the certification of SCH-1 for these metals. Between 1972 and 1976, a working group of Sub-Committee 2 (Chemical Analysis) of ISO Technical Committee 102, Iron Ores, established that a particular procedure for the determination of sodium and potassium by atomic absorption spectrophotometry was reliable when applied to iron ores. Therefore, in 1977, a Canadian interlaboratory program was arranged in which 10 independent analysts were requested to analyze SCH-1 for sodium and potassium by the a.a. method, for certification purposes. Fortuitously, at the same time, Sub-Committee 2 of ISO Technical Committee 102 organized an extensive international trial involving up to 48 laboratories which applied the candidate a.a. method to a number of reference iron ores, including SCH-1. Because CCRMP (Canadian Certified Reference Materials Project) personnel were involved in the international program, it was possible to obtain the raw analytical results for SCH-1 and combine them with data from the concurrent Canadian program as well as with the original data from 1975¹ ultimately for the computation of recommended values for sodium and potassium.

This report presents details of the interlaboratory programs and the statistical treatment of all analytical results from 1975 through 1977. It should be noted that for continuity and clarity, laboratories (contributors) participating in the two 1977 programs have been assigned code numbers that follow consecutively those used in the original report of certification of SCH-1 in 1975¹.

Original interlaboratory program, 1975¹

Contributors were requested to analyze two randomly-selected bottles of SCH-1 in quintuplicate for 13 constituents, including sodium and potassium, by methods of their choice. The names of the participating laboratories were given previously¹ and their sodium and potassium results appear in Tables 1a and 2a respectively.

Canadian interlaboratory program, 1977

As in the 1975 program, contributors were requested to analyze in quintuplicate two randomlyselected bottles of SCH-1 for sodium and potassium. They were asked to follow in detail the procedure of the ISO-evaluated a.a, method as described in the appendix. As will be seen, the results of this program shown in Tables 1b and 2b were used to confirm homogeneity of SCH-1.

Laboratories participating in the Canadianorganized program for 1977 were:

- Broken Hill Proprietary Limited, Australia
- Dominion Foundries and Steel Limited, Hamilton, Ontario
- Falconbridge Nickel Mines Limited, Metallurgical Laboratories, Thornhill, Ontario
- Geological Survey of Canada, Ottawa, Ontario
- Hudson Bay Mining and Smelting Company Limited, Flin Flon, Manitoba
- Iron Ore Company of Canada, Sept-Iles, Quebec
- Lakefield Research of Canada Limited, Lakefield, Ontario
- Lerch Brothers, U.S.A.
- Ontario Ministry of Natural Resources, Toronto, Ontario
- Sherritt Gordon Mines Limited, Mining and Milling Division, Lynn Lake, Manitoba

ISO interlaboratory program, 1977

As mentioned, the international program was designed by ISO/TC 102/SC 2 to determine the precision and relative accuracy of a candidate a.a. method for the determination of sodium and potassium in iron ores. Each participating laboratory analyzed in duplicate two randomly-selected bottles of a number of reference samples including SCH-1³. The results for the latter were made available to the CCRMP by special arrangement with ISO. Although 48 laboratories accepted samples and presumably would contribute results to ISO/TC 102/SC 2, it was considered sufficient for the certification of SCH-1 to utilize results from only the first 22 laboratories reporting as given in Tables 1c and 2c. The countries and the number of laboratories that contributed were: Canada 2, France 5, India 3, Japan 6, Sweden 3 and U.K. 3.

STATISTICAL TREATMENT OF ANALYTICAL RESULTS

As in the case of the 1975 interlaboratory program¹, the analytical data were evaluated to confirm the homogeneity of SCH-1 with respect to sodium and potassium and to estimate consensus values and their confidence intervals for use as the recommended values.

Confirmation of homogeneity

In addition to the lack of consensus among analytical results obtained for sodium and potassium in 1975, the between-bottle homogeneity of the potassium content was questionable. Table 4a, reproduced from the previous report¹, shows that 4 out of 18 laboratories (excluding those with zero within-laboratory variance) rejected the null hypothesis. It was therefore considered necessary to verify the between-bottle homogeneity. Table 4b shows there is no rejection of the null hypothesis inherent in the analytical results from the 1977 interlaboratory program. Because the ISO interlaboratory program produced analytical results only in duplicate, the same t-test would not be meaningful. Therefore, a two-way analysis of variance was performed, i.e. between-laboratories, betweenbottles within-laboratories, and within-bottles. After excluding those data with zero withinlaboratory variance and those with more than two duplicates per bottle³, 17 sets were available for this purpose. The results confirmed that the sample is homogeneous.

This was also the case with sodium results; Table 3a shows that three sets were rejected, while 3b shows no rejection. The two-way analysis of variance with 19 sets of results from the ISO international test confirms homogeneity of the sodium content of SCH-1.

Estimation of consensus values and related statistical parameters

The analytical results comprising a total of 53 sets for both sodium and potassium are given in Table 1 and illustrated in Figure 1. The improvement in precision of the 1977 results over those of 1975 is clearly evident.

As is normal practice in avoiding bias, sets of results whose means differ from the overall mean by more than twice the standard deviation of the individual results were excluded from the one-way analysis variance used to estimate the consensus values and their confidence intervals (Table 5).

It is evident that the 1977 means for sodium and potassium given in Table 5 are essentially the same as the provisional values assigned after the original interlaboratory program of 1975¹. However, the confidence intervals and average withinlaboratory coefficients of variation are appreciably smaller than the corresponding ones of 1975. The improved quality of results is reflected in the new certification factors which are 2.7 and 2.6 respectively for sodium and potassium. Because these factors are much lower than the critical value of 4 2 , the consensus means are accepted as recommended values for the alkali metals (Table 6). Success of the 1977 program was due to the exclusive use of the "standardized" ISO method (see Appendix) which is in contrast to the "free-choice" scheme of 1975.

Discussion of analytical method

In view of the newly assigned recommended values for sodium and potassium, it is of interest to reconsider the original 1975 values for methodological significance. Out of 21 laboratories reporting, six and four used flame emission spectroscopy for sodium and potassium respectively, and their means ranged from 0.01% to 0.06% for sodium and 0.02% to 0.04% for potassium. The emission method is evidently much less satisfactory than the atomic absorption method for a material such as SCH-1 and this is in line with theoretical expectations⁴.

Although details of procedures used by the various contributors are not available, it is probable that the greater overall range and poorer within-laboratory precision of the 1975 atomic absorption results were due to incomplete decomposition, giving low results on the one hand, and on the other hand contamination from use of glassware rather than the recommended plastic-ware, giving high results.

REFERENCES

- Steger, H.F., Bowman, W.S., Sutarno, R. and Faye, G.H., "Iron Ore, SCH-1: its characterization and preparation for use as a certified reference material"; Mineral Sciences Laboratories Report MRP/MSL 75-168 (TR), August 1975.
- Sutarno, R. and Faye, G.H., "A means for assessing certified reference ores and related materials"; Talanta v. 22, pp 676-681; 1975.
- Sutarno, R., "Procedure for statistical evaluation of analytical data resulting from international tests", Mineral Sciences Laboratories Report MRP/MSL 77-393 (IR).
- Lvov, B.V., "Atomic absorption spectrochemical analysis"; American Elsevier Publishing Co. Inc., New York; p 35; 1970.

APPENDIX

OUTLINE FOR ISO-PROPOSED ATOMIC ABSORPTION METHOD FOR SODIUM IN IRON ORES

Subsamples, weighing approximately 0.5 g, of ore previously dried at 105°C for two hours, are decomposed by repeated treatment with a 1:1 mixture of concentrated hydrochloric and hydrofluoric acids. The test solution, essentially free of insoluble matter, is diluted to an appropriate volume in a plastic volumetric flask with dilute hydrochloric acid. Plastic pipettes are used if aliquoting is necessary.

The absorbance of the test solution is measured together with a blank solution having essentially the same acid and iron concentrations as the test solution. Depending upon the concentrations of the alkali metals, a wavelength of 589.0 or 589.6 nm is used for sodium, and 766.5 or 769.9 nm is used for potassium.







3

l	1	l		
	1	F		

TABLE	Ł
-------	---

Sodium results	of	three	Interlahoust	_ ·		
			incertaboratory	Programs:	(wt	Z)

(a) Original Contdet							(WL /)		
(a) of ginar Gertifica	ation Prop	gram, 197	5					•		
LAB- 1 (A.A.)	010									
LA8- 3 (A.A.)	•019	•0192	.0196	5 . 0188	. 1100					
LAB- 5 (FLAME)	• 013	• • • • • • • • • • • • • • • • • • • •	•017(0510.	.0240	•0194	•0187	.0183	0104	
LAB- 6 (A.A.)	- 032	0 0320	•0340	.0340	-0340	•0150	.0140	.0230	.0200	.0186
	.017	0.1010	.0190	-0180	-0170	•0320	•0320	.0310	0200	•0150
LAB= 7 (FLAME)	.010	0100				•0100	-0190	.0190	-0190	+0300
	.0260	0 0.250	.0100	•0100	.0100	0100				•0160
	.0190	0230	•0250	•0260	.0250	+0100	0.75.			
LAB-12 (FLAME)	.0600	.0600	*0190	•0190	.0180	.0190	•0250	.0560	.0250	.0250
LAB=16 (A A L	.0193	.0206	•0000	•0600	.0600	.0600	•0:80	•0190	.0190	-0190
1A8 = 16 (51 AVE)	•0190	.0200	0212	.0198	•0204	•0197	.0204	.0600	.0600	.0600
LAB-18 (A.A.)	•0140	.0130	-0150	-0210	•0200	.0200	-0210	.0208	.0194	•0198
LAB-20 (A.A.)	.0058	.0058	.0048	.0160	.0170	.0150	.0160	•U200	.0200	.0190
LAB-21 (A.A.)	•0260	•0560	.0240	.0230	•0058	.0058	.0058	.0140	<i>0</i> 150	.0160
LAB-23 (FLAME)	•0020	.0580	.0610	.0570	•0230	•0230	.0250	.0230	.0058	.0058
LAB-24 (A.A.)	.0710	•0130	.0150	.0120	-0300	•0540	.0520	.0540	.0200	.0210
LAB-26 (A.A.)	•0310	.0330	.0270	.0330	.0220	•0120	.0150	.0110	.0120	•0550
LAB-28 (A.A.)	+0300	•0194	.0193	.0194	.0107	•0190	.0150	.0190	0200	+0130
LAB~30 (A.A.)	.0203	•0300	.0200	.0300	••••	•0191	.0193	.0196	.0192	0100
	.0200	+0190	.0198	.0194	+0196	.0102				•0193
CAD-39 (A.A.)	.0170	-0170	•0200	•0200	.0200	+0300	.0192	.0198	.0194	. 02 90
			.0100	•0200	.0170	.0170	.0200	.0200	.0300	.0200
(b) Construction							• • 180	•0170	.0170	.0170
(b) Canadian Interlabora	atory Pro	era., 107	7							
LAB-61 (0 0)		52 din, 197	/							
LAB-42 (A A)	.0185	•0179	.0186	.0179	4101					
LAB=43 (A A)	.0140	.0190	.0180	.0170	.0181	.0180	.0177	.0193	0177	0.01
LAB-44 (A.A.)	.0210	.0210	.0220	.0220	.0180	•0180	.0180	.0180	0180	•0101
LAB-45 (A.A.)	•0193	.0195	.0191	.0193	.0220	.0200	.0230	.0200	0200	+0180
LAB-46 (A.A.)	.0200	.0210	.0200	.0190	0200	•0195	.0190	.0191	.0195	.0210
LAB-47 (A.A.)	•0196	.0198	.0191	.0205	.0199	.0200	.0200	.0200	0200	- 0230
LAB-48 (A.A.)	*0220	·0200	.0210	.0200	.0200	•0195	.0196	.0195	.0206	.0196
	0130	.0110	.0110	.0170	.0120	0120	0120			
LAB-49 (A.A.)	.0160	•0120				• • • • • • •	.0120	.0130	.0110	.0130
LAB-50 (A.A.)	-0100	+0100	.0180	.0190						
		*0110	.0110	•0110	.0140	.0100	0110			
(c) TSO T							•0110	.0100	.0100	.0110
(c) 150 Incernational Tes	st									
LAB-51	0104									
LAB-52	0104	•0180	•0174	.0182	.0178	0170	0124			
LAB-53	.0166	•0180	.0172	.0174		+01/6	.0176	.0169		
LA8-54	.0177	+0180	.0170	.0183						
LAB-55	.0182	•0100	.0172	•0163						
LAB-56	.0189	•0193	•0189	•0196						
	0196	+0100	.0188	.0185						
	.0180	.0160	•0186	•0174						
	.0160	.0160	•0150	•0160						
	•0200	.0170	.0220	•0150						
LAB-40	.0190	.0200	.0190	.0180						
LAB-63	•0193	•017B	.0181	0105						
LAB~64	•0194	.0182	.0198	-0195						
LAB-65	•0186	•0203	.0186	.0202						
LAB-66	.0192	.0182	.0190	.0180						
LAB-67	+0194	.0195	.0202	.0204						
LAB-68	.0200	•0184	.0186	.0184						
LAB-69	-0207	.0203	.0203	.0200						
LAB-70	.0189	00180	.0170	.0170						
	.0196	+010F	.0192	.0185						
LAB-72	.0181	*0193	•0196	.0204						
		-0102	.01/8	.0181						

.

TABLE 2

Potassium results of three Interlaboratory Programs; (wt %)

(a) Original Certification Program, 1975

LAB-1 (A.A.) LAB-3 (A.A.) LAB-5 (A.A.) LAB-6 (A.A.) LAB-7 (FLAME) LAB-10 (A.A.) LAB-10 (A.A.) LAB-11 (FLAME) LAB-14 (A.A.) LAB-16 (FLAME) LAB-16 (A.A.) LAB-20 (A.A.) LAB-21 (A.A.) LAB-23 (FLAME) LAB-23 (A.A.) LAB-23 (A.A.) LAB-23 (A.A.) LAB-23 (A.A.) LAB-37 (A.A.) LAB-39 (A.A.)	.0226 .030 .031 .016 .020 .021 .0221 .0241 .0240 .0410 .0360 .0210 .0500 .0400 .0301 .0170 .0250	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3 .027 0 .025 0 .0240 0 .0210 0 .0210 0 .0220 0 .0220 0 .0220 0 .0220 0 .0220 0 .0220 0 .0240 0 .0280 .0450 .0240 .0240 .0250 .0450 .0280	5 .02 $A60$.0310 0 .0170 0 .0210 0 .0210 0 .0210 0 .0210 0 .0210 0 .0210 0 .0240 0 .0240 0 .0240 0 .0240 0 .0240 0 .0290 0 .0270	.0283 .0270 .0310 .0200 .0230 .0210 .0240 .0240 .0240 .0340 .0340 .0390 .0390 .0390 .0300 .0208 .0600 .0295 .0180 .0280	.0274 .0260 .0340 .0170 .0210 .0210 .0210 .0240 .0280 .0280 .0280 .0280 .0280 .0280 .0280 .0280 .0280 .0240 .0250 .0213 .0400 .0309 .0180 .0240	.0274 .0260 .0320 .0160 .0220 .0260 .0260 .0260 .0260 .0240 .0450 .0450 .0450 .0450 .0420 .0220 .0220 .0450 .0220 .0450 .0220 .0450 .0280	.0264 .0270 .0270 .0160 .0210 .0210 .0280 .0280 .0280 .0280 .0230 .0400 .0230 .0400 .0230 .0210 .0300 .0299 .0180 .0240	.0271 .0270 .0360 .0160 .0200 .0200 .0350 .0280 .0150 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .04500 .045000 .045000 .045000 .045000 .045000 .0450000000000	.0255 .0280 .0220 .0270 .0200 .0200 .0200 .0340 .0340 .0210 .0440 .0390 .0440 .0390 .0440 .0310 .0210 .0210 .0240
incertatora	cory Prog	gram, 197	7							
LAB-41 (A.A.) LAB-42 (A.A.) LAB-43 (A.A.) LAB-44 (A.A.) LAB-45 (A.A.) LAB-45 (A.A.) LAB-46 (A.A.) LAB-47 (A.A.) LAB-49 (A.A.) LAB-49 (A.A.)	.0261 .0250 .0270 .0208 .0255 .0255 .0255 .0260 .0280 .0280 .0280 .0310	. 9272 . 9250 . 0250 . 0250 . 0272 . 9260 . 0290 . 0310 . 0270 . 0310	.0269 .0250 .0230 .0210 .0250 .0294 .0250 .0280 .0280 .0300 .0310	.0262 .0250 .0270 .0210 .0240 .0268 .0260 .0310 .0320 .0310	.0263 .0250 .0270 .0211 .0250 .0250 .0250 .0280	.0267 .0240 .0250 .0250 .0250 .0267 .0280	.0255 .0260 .0270 .0210 .0250 .0270 .0270	.0265 .0240 .0250 .0211 .0250 .0292 .0260	.0259 .0250 .0270 .0210 .0250 .0268 .0280	.0266 .0250 .0270 .0211 .0250 .0267 .0310
(c) ISO International m						.0300	.0310	.0310	.0320	.0330
LAB-51 LAB-52 LAB-53 LAB-55 LAB-55 LAB-55 LAB-56 LAB-57 LAB-56 LAB-61 LAB-62 LAB-63 LAB-63 LAB-63 LAB-64 LAB-65 LAB-66 LAB-67 LAB-67 LAB-71 LAB-72	. 0264 . 0248 . 0257 . 0257 . 0254 . 0259 . 0259 . 0250 . 0250 . 0250 . 0254 . 0254 . 0254 . 0254 . 0251 . 0254 . 0251 . 0254 . 0257 . 0250 . 0264 . 0257	.0268 .0257 .0260 .0230 .0235 .0257 .0250 .0250 .0240 .0250 .0260 .0270 .0262 .0265 .0266 .0264 .0260 .0264 .0264 .0264 .0264 .0267 .0251	.0273 .0250 .0254 .0223 .0288 .0257 .0265 .0240 .0240 .0240 .0240 .0245 .0258 .0256 .0264 .0261 .0260 .0260 .0257 .0230 .0226 .0260 .0226 .0260	.0269 .0245 .0260 .0238 .0254 .0267 .0230 .0250 .0250 .0250 .0260 .0266 .0266 .0266 .0258 .0258 .0258 .0252 .0230 .0252 .0230 .0271 .0249	.0249	•0250	•0247	•0249		

Results of t-test on between-bottle means of analytical results for sodium in SCH-1

TABLE 3

(a) Original Certification Program, 1975

	BOTTLE 1			BOTTLE 2								
	N	MEAN	ST.DEV.	 N			NULL HYPOTH.			OVERALL		
LAB-1 (A.A.) LAB-3 (A.A.) LAB-5 (A.A.) LAB-6 (A.A.) LAB-7 (FLAME) LAB-10 (A.A.) LAB-11 (FLAME) LAB-11 (FLAME) LAB-14 (A.A.) LAB-14 (A.A.) LAB-20 (A.A.) LAB-21 (A.A.) LAB-21 (A.A.) LAB-23 (FLAME) LAB-24 (A.A.) LAB-26 (A.A.) LAB-26 (A.A.) LAB-26 (A.A.) LAB-26 (A.A.) LAB-37 (A.A.) LAB-39 (A.A.)	* 5556355555555554552555	MEAN •0283 •0276 •0300 •0168 •0200 •0218 •0240 •0342 •0288 •0142 •0246 •0442 •0442 •0444 •0315 •0208 •05500 •0400 •0300 •0172 •0264	ST.DEV. .0005 .0028 .0014 .0017 0.0000 .0016 .0007 .0089 .0004 .0004 .0004 .0004 .0004 .0004 .0005 .0009 .0009 .0004 .0011 0.0008 .0008 .0008 .0004 .0001	ក ភេទទេ ទំ	MEAN .0268 .0268 .0302 .0168 .0200 .0220 .0204 .0200 .0344 .0282 .0450 .0410 .0222 .0446 .0410 .0220 .0420 .0420 .0420 .0420 .0420 .0420 .0420 .0420 .0420 .0446 .0410 .0220 .0446 .0410 .0220 .0302 .0304 .0446 .0410 .0446 .0410 .0446 .0410 .0446 .0410 .0446 .0410 .0446 .0446 .0410 .0446 .0410 .0446 .0446 .0410 .0446 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0446 .0420 .0420 .0446 .0420 .05000 .05000 .05000 .05000 .05000 .05000 .05000 .05000 .0	ST.DEV. .0008 .0057 .0010 .0010 .0012 .0005 .0005 .0004 .0005 .0014 .0038 .0010 .0014 .0038 .0010 .0014 .0038 .0010 .0004 .0013 .0013 .0017	NULL HYPOTH. REJECT A A A ***R** A A A A A A REJECT A REJECT A REJECT A REJECT A A REJECT A	N 10 10 10 12 6 10 10 10 10 10 10 10 10 10 10	MEAN .0275 .0272 .0301 .0168 .0200 .0219 .0207 .0220 .0343 .0285 .0146 .0234 .0244 .0444 .04407 .0273 .0209 .04460 .0450 .0303 .0180 .0257	ST.DEV. .0010 .0020 .0039 .0013 0.0000 .0014 .0007 .0063 .0005 .0005 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0005 .0018 .0019 .0019 .0014 .0005 .0019 .0014 .0005 .0015 .0015 .0015 .0015 .0005 .0015 .0015 .0015 .0005 .0015 .0015 .0015 .0005 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0015 .0005 .0015 .0005 .0015 .0005 .0015 .0005	C.V.(%) 3.70 7.31 12.96 7.94 0.00 6.26 3.26 28.75 1.41 1.85 3.54 7.59 1.16 3.08 19.01 1.15 21.00 12.83 2.28 6.93	

TOTAL 201 .0274 .0094 9

34.30

(b)	Canadian	Interlaboratory	Program.	1977
		· - J	- robrant,	1211

LAB-41 LAB-42 LAB-43 LAB-44 LAB-45 LAB-46 LAB-46 LAB-47 LAB-48 LAB-49 LAB-50	(A.A.) (A.A.) (A.A.) (A.A.) (A.A.) (A.A.) (A.A.) (A.A.) (A.A.) (A.A.) (A.A.)	5 5 5 5 7HERE 6 7HERE 5	.0265 .0250 .0268 .0209 .0248 .0272 IS ONLY .0287 IS ONLY .0310	.0005 .0000 .0011 .0001 .0004 .0014 1 BOTTLE .0012 1 BOTTLE .0000	55555 655	.0262 .0248 .0262 .0210 .0250 .0273 .0285 .0314	.0005 .0008 .0011 .0002 .0000 .0011 .0021 .0021	A A A A A A A	10 10 10 10 10 10 10 10 10 10 12 4	.0264 .0249 .0265 .0210 .0249 .0272 .0256 .0286 .0300 .0312	.0005 .0006 .0011 .0001 .0003 .0012 .0005 .0016 .0022 .0008	1.89 2.28 4.08 1.27 4.36 2.14 5.67 7.20 2.53
								TOTAL	91	.0265	.0029	11.03

(-)		Gereitt	BOTTLE	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		BOTTLE	2						
								NULL HYPOTH.	OVERALL				
		N	MEAN	ST.DEV.	N	MEAN	ST.DEV.		N 100	MEAN	ST.DEV.	C.V.(%)	
LAB- 3 LAB- 5 LAB- 6 LAB- 7	(A.A.) (FLAME) (A.A.) (FLAME)	5 5 6 3	.0160 .0332 .0180	•0029 •0011 •0013 0-0000	5 5 6	.0174 .0310 .0177	.0039 .0010 .0015	A REJECT	10 10 10 12	•0190 •0167 •0321 •0178	•0004 •0033 •0015 •0013	2.32 19.97 4.75 7.50	
LAB- 9 LAB-10 LAB-11 LAB-12	(A.A.) (A.A.) (FLAME) (FLAME)	555555	.0254 .018B .0600	•0005 •0004 •0000	3 5 5 5 5 5 5 5 5	.0254 .0188 .0600	.0005 .0004 .0000	***R** A A A	6 10 10 10	.0100 .0254 .0188 .0600	0.0000 .0005 .0004 .0000	0.00 2.03 2.24	
LAB-14 LAB-16 LAB-18	(A.A.) (FLAME) (A.A.)	ភភភ	.0200 .0150 .0055	•0007 •0016 •0004	555	•0200 •0200 •0152 •0058	.0008 .0008 .0000	A A A A	10 10 10	•0201 •0200 •0151	•0006 •0007 •0012	3.13 3.33 7.93	
LAB-21 LAB-23 LAB-24	(A.A.) (FLAME) (A.A.)	5554	•0588 •0122 •0310	•0015 •0026 •0004 •0028	5 5 5 5	•0224 •0532 •0120 •0190	•0019 •0016 •0007 •0025	A REJECT A REJECT	10 10 10	•0234 •0560 •0121	.0020 .0036 .0006	8.35 6.41 4.69	
LAB-26 LAB-28 LAB-30 LAB-37	(A+A+) (A+A+) (A+A+)	5 2 5 5	•0193 •0300 •0196 •0200	•0001 0•0000 •0005 0•0000	525	•0193 •0250 •0194	.0002 .0071 .0004	A A A	10 4 10	•0243 •0193 •0275 •0195	•0001 •0050 •0004	27.95 .77 18.18 2.10	
LAB-39	(A.A.)	5	•0174	•0015	5	•0172	•0004	A A TOTAL	10 10	.0220 .0173	•0042 •0011	19.17 6.12	

TABLE 4		
Results of t-test on between-bottle means of analytical results for potassium	in	SCH-1

(b) Canadian Interlaboratory Program, 1977

LAB-41 LAB-42 LAB-43 LAB-44 LAB-45 LAB-45 LAB-47 LAB-47 LAB-48 LAB-49 LAB-50	$(A \cdot A \cdot)$ $(A \cdot A \cdot)$	5 5 5 5 5 5 5 7 4 6 7 1 4 5 5	.0182 .0180 .0216 .0193 .0200 .0198 I5 ONLY .0127 IS ONLY .0114	.0003 .0007 .0005 .0001 .0005 1 BOTTLE .0023 1 BOTTLE .0015	5555 555 6 5	.0182 .0180 .0208 .0193 .0202 .0198 .0122 .0104	.0007 .0000 .0013 .0002 .0004 .0005 .0008	А А А А А А		10 10 10 10 10 5 12 4 10	.0182 .0180 .0212 .0193 .0201 .0198 .0206 .0124 .0178 .0109	.0005 .0005 .0010 .0002 .0006 .0005 .0009 .0016 .0013 .0012	2.72 2.62 4.87 .95 2.82 2.34 4.34 13.06 7.09 10.98
									TOTAL	91	.0176	.0036	20.43

7

TABLE 5

Estimation of statistical parameters for sodium and potassium in SCH-1 from the results of 1975 and 1977 interlaboratory programs.

Statistical	Na	L .		К
parameters	1975	1977	1975	1977
Results, no. of sets Total Used for ANOVA	21 19	53 51	21 21	53 50
No. of results Total used for ANOVA	201 181	384 364	201 201	384 360
Median, %	0.0192	0.0190	0.0260	0.0258
Mean, %	0.0192	0.0186	0.0274	0.0256
95% Confidence limits of the means, % Low High	0.0163 0.0221	0.0172 0.0200	0.0232 0.0316	0.0241 0.0271
Average within- laboratory cv,%	7.71	5.48	7.53	4.52
Certification factor	3.9	2.7	4.1	2.6

TABLE 6

Recommended values for sodium and potassium in SCH-1:

Constituent	Recommended value	95% confidence limits
Sodium	0.019	0.017 - 0.020
Potassium	0.026	0.024 - 0.027