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## DEPARTMENT OF ENERGY, MINES AND RESOURCES MINES BRANCH OTTAWA

# THE FLOATABILITY OF TWENTY-ONE NON-METALLIC MINERALS

R. A. WYMAN

MINERAL PROCESSING DIVISION

JANUARY 1969

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THE FLOATABILITY OF TWENTY-ONE NON-METALLIC MINERALS

by

R.A. Wyman\*

#### ABSTRACT

The floatabilities of twenty-one non-metallic minerals as affected by pH regulators, collectors and modifiers have been determined. Symbols have been used in presentation of results to afford ease of comparison.

The work herein reported extends that contained in Technical Bulletin TB 70, "The Floatability of Eleven Common Non-Metallic Minerals" (April 1965). Ten additional minerals are included and fourteen additional collectors.

This project is intended to assist those working on the floatation of non-metallic minerals.

<sup>\*</sup>Head, Industrial Minerals Milling Section, Mineral Processing Division, Mines Branch, Department of Energy, Mines and Resources, Ottawa, Canada.

#### Direction des mines

#### Bulletin technique TB-108

## Étude n° MP-IM-R-1

## LA FLOTTABILITÉ DE VINGT ET UN MINÉRAUX NON MÉTALLIQUES

par

R.A. Wyman\*

#### RÉSUMÉ

La présente étude fait état de l'action des régulateurs de pH, des agents collecteurs et des agents modificateurs sur la flottabilité de vingt et un minéraux non métalliques. Afin de simplifier la comparaison, on a utilisé des symboles pour la présentation des résultats.

Ces données représentent une extension des études contenues dans le Bulletin technique TB 70, intitulé "La flottabilité de onze minéraux communs non métalliques", publié en avril 1965. Dix nouveaux minéraux ont été inclus, ainsi que quatorze agents collecteurs supplémentaires.

Ces travaux ont été entrepris en vue d'aider les chercheurs dans le domaine de la flottation des minéraux non métalliques.

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Carlon

#### INTRODUCTION

For a number of years the Industrial Minerals Milling Section employed a small-scale test to aid in solving floatation problems. The test method was simple and direct. It was intended to be indicative rather than definitive, a guide rather than a solution.

Because the information obtained has proved helpful in solving a number of floatation problems, it was decided to publish the results. Technical Bulletin TB 70 was issued in 1965 and presented the results of approximately 4000 of the small-scale experiments, designed to indicate the floatability of eleven non-metallic minerals under a variety of conditions. Its purpose was to point out possible new approaches for the separation of non-metallic minerals by floatation, particularly where separation is now difficult, and to suggest possible methods of separation for minerals for which no effective system is presently known.

The work has since been extended to approximately 14,000 of the small-scale experiments. These have indicated the floatabilities of twenty-one non-metallic minerals under a variety of conditions. It is the purpose of this report to present these data in a readily usable form.

#### SCOPE

The responses of 21 non-metallic minerals to 31 different collectors under acid, neutral and basic conditions are indicated. The effects of six common modifiers on these responses are shown.

#### EXPERIMENTAL PROCEDURE

Minerals to be investigated are obtained in the purest form available, as crystals where possible. Each mineral is pulverized, and the 28 to 325-mesh fraction is isolated. This fraction is selected because most floatation is done within this size range and a rough visual estimation of the number of particles picked up by a bubble can be made.

These individual fractions are rinsed lightly with dilute HCl and then washed with distilled water to neutral pH. The intent is to cleanse the surfaces without etching or altering them significantly. Each mineral is stored in a clean bottle in distilled water.

The apparatus used consists of a small turntable, a beaker, a stirring rod, and a glass tube with a rubber eyedropper bulb and screw clamp at its top. Before an experiment, the beaker, stirring rod and glass tube are rinsed with chromic-acid cleaning solution and washed thoroughly with distilled water. Measuring devices such as graduates and pipettes are similarly cleaned once or twice per day.

A single experiment consists of the following steps:

- 1) About 100 ml of distilled water is put into the beaker.
- 2) A small quantity (about 0.01 g) of the particular mineral to be examined is added.
- 3) The pH regulator, if used, is added and mixed by stirring rod.
- 4) Modifier (to give 0.05 gpl concentration), if used, is added and mixed by stirring rod.
- 5) Collector to give 0.1 gpl concentration is added and mixed by stirring rod.
- 6) The beaker is placed on the turntable. The bubble tube is lowered vertically until its tip is about a half-inch above the mineral.
- 7) The screw clamp on the rubber bulb is turned to produce a bubble at the lower end of the glass tube.
- 8) The beaker and turntable are raised until the bubble makes contact with the mineral.
- 9) The beaker is lowered until the bubble and adhering particles are surrounded by the liquid.
- 10) Particles adhering to the bubble are estimated (none, one, two, ten, fifteen, etc.).
- 11) The glass tube is lightly tapped and the tenacity with which the particles cling to the bubble (i.e. poor, fair or good) is estimated.
- 12) These estimations are recorded on a standard form.

The experiments progress from simple to more complex. The first step for each mineral is to run a blank, i.e., the bubble is placed in contact with the mineral in distilled water only. With many minerals, several particles will be picked up but may drop off as soon as the tube

is tapped lightly. Each mineral is then tested for response to each collector in turn, under acidic, neutral and basic conditions. In the neutral case, the system consists only of the mineral and the collector in distilled water. For acidic conditions, 1 ml of  $H_2SO_4$  of a predetermined concentration is added from a 1-ml dispensing bottle. The acid is of such a concentration that when 1 ml is added to 100 ml of distilled water a pH of 3.0 will result. Similarly, for basic conditions, 1 ml of  $Na_2CO_3$  of predetermined concentration is added from a 1-ml dispensing bottle, producing a pH of 10.0 in 100 ml of distilled water. For both the acidic and basic experiments the regulator is mixed-in by stirring and the collector is added and also mixed-in by stirring, before bubble contact is made.

The pH will vary slightly during the experiment, owing to the presence of the mineral and the other reagents. However, the quantity of both mineral and reagents is comparatively small and the operations are conducted rapidly. It has been found that the pH varies little from that of the acid or base in distilled water. The maximum change in pH is observed when ionizing modifiers are used. Since the general acidic or basic nature of the system is not altered, however, and the total variation is rarely beyond one pH point, it is considered preferable not to readjust the pH to the specified points. This would add greatly to the time required to perform an individual acidic or basic experiment. Moreover, since the observations are not absolute, the over-all results must be considered as relative only. The one operation with any significant bias is that for "neutral" when HF is used as a modifier. The pH in this case is always slightly on the acid side. However, this would also be the case in larger-scale floatation practice.

In experiments with modifiers, each system - mineral, collector, and pH control - is repeated with each modifier. The modifier is added before the collector and mixed by stirring.

#### MINERALS INVESTIGATED

The condition in selecting minerals for this study was that they come from natural occurrences and be nearly pure. The geographical origin of some of the minerals used could not be ascertained. With the exception of talc, all are crystalline. Minerals herein reported upon are as follows:

	Mineral	Colour	Origin	
1.	Albite	White	Ontario ?	
2.	Apatite	Green	Quebec ?	
3.	Barite	Buff	B.C.	
4.	Beryl	Pale green	Manitoba	
5.	Biotite	Black	Ontario	
6.	Calcite	White	Ontario ?	
7.	Celestite	White	N.S. ?	
8.	Corundum	Purple-grey	Madagascar	
9.	Fluorite	Purple, green	Newfoundland?	
10.	Garnet	Deep red	Ontario	
11.	Gypsum	Transparent	N.S.	
12.	Hornblende	Black	Ontario	
13.	Kyanite	Greenish blue	Ontario	
14.	Magnesite	Off white	Ontario	
15.	Muscovite	Transparent	Ontario ?	
16.	Orthoclase	White	Ontario ?	
17.	Quartz	Transparent	Ontario ?	
18.	Spodumene	Light greenish	Manitoba	
19.	Talc	Greenish (schistose)	Quebec	
20.	Topaz	Transparent, tan	Brazil	
21.	Vermiculite	Dark grey	Ontario	

#### COLLECTORS INVESTIGATED

The principle of collection in floatation is simple - to render the surfaces of the mineral to be floated hydrophobic, thus allowing attachment to an air bubble and elevation to the surface. In theory this should be possible with any mineral. In practice there are many complicating factors, and finding a chemical that will be selective for one specific mineral in a mixture of minerals is by no means easy.

Since the mineral surface is the site of the action, the chemicals involved are the so-called surface-active agents (or surfactants). Broadly, these agents fall into three classes according to their activity - non-ionic, anionic, and cationic. Non-ionic agents do not ionize in solution. When anionic agents are dissolved, the negative ions, or anions, are active, and when cationic agents are dissolved the positive ions, or cations, are active. Although fatty acids are anionic, they are widely used in non-metallic mineral floatation and, for the purpose of this report, have been considered as a fourth class.

There are many thousands of surfactants available, and the list is continually increasing and changing. The older, stable types are rapidly being replaced by biodegradables. For the purpose of floatation these can be narrowed into a comparatively small number of general types, with an even narrower grouping applicable to non-metallics. Reagents that are in more or less common use as floatation collectors have been included as well as newcomers.

The collectors used are listed below according to their activity, type, trade name, and general chemical name.

NON-IONIC, Ethanol

Igepal CTA Rexol 25J

Alkyl phenoxypoly- (ethyleneoxy) ethanol

Nonyl phenol polyethoxy ethanol

ANIONIC, Petroleum sulphonate

Reagent 801 Petroflote 462 Leonal SA "Green acid"-type petroleum sulphonate Sodium alkyl-aryl petroleum sulphonic acid Naphthalene sulphonic acid derivative

#### ANIONIC, Fatty acid sulphonate

Emcol 4510

Fatty acid aliphatic sulphonate

Sulphonated castor oil Sulphonated castor oil (60% fats)

Prestabit oil V

Sulphonated fatty acids

#### ANIONIC, Alkyl sulphate

Duponol 80

Sodium octyl sulphate

Nutrapon WAC

Sodium lauryl sulphate

Duponol OS

Diethyl cyclohexylamine lauryl sulphate

#### ANIONIC, Sodium taurate

Igepon TE 42

Sodium-N-methyl-N-tallow acid taurate

Igepon T 33

Sodium-N-methyl-N-oleoyl taurate

Igepon TN 74

Sodium-N-methyl-N-palmitoyl taurate

#### CATIONIC, Primary amine

Adogen 170A

Technical tallow amine acetate

Armac C

Coco amine acetate

Armeen L-9

Primary beta amine

#### CATIONIC, Diamine

Duomac T

Tallow diamine di-acetate

Duomac C

Coco diamine di-acetate

Duomeen L-11

Beta diamine

#### CATIONIC, Tertiary amine

Nalcimine G 11

Hydroxyethyl alkyl imidazoline (glyoxalidine)

Armeen 12D

Lauryl amine

Armeen DLM 11

Beta tertiary amine

#### CATIONIC, Quaternary

Emcol E 607-40

N(laur oyl-colamino-formyl-methyl) pyridinium

chloride

ARQUAD 12

n-alkyl trimethyl ammonium chloride

Retarder LA

Cetyl trimethyl ammonium bromide

Ammonyx T

Cetyl trimethyl benzyl ammonium chloride

FATTY ACID, Tall oil

Neofat 42-12 Tall oil base fatty acid

Actinol FA 2 Tall oil fatty acid

FATTY ACID, Oleic

Harfat 231 Oleic acid blend

Distilled oleic acid Oleic acid

#### MODIFIERS INVESTIGATED

As pointed out in the previous section, floatation is effected by the action of a collector, causing the surface of a mineral to become hydrophobic. In many cases, however, it is necessary to use, as well, modifying agents to achieve this. Such agents simply aid in getting the collector onto the surface of the mineral to be floated. This action is called activation. On the other hand, a modifying agent may prevent a collector from getting onto the surface of unwanted minerals. This action is called depression. These are probably the most important functions of modifiers.

The regulation of pH is itself a form of modifying. As indicated earlier, all the experiments take place at three pH levels. The cleaning of surfaces, dispersion of ultra-fine solids, precipitation of dissolved salts, and other functions, are handled by modifying agents. Some agents serve more than one purpose - e.g., sodium carbonate may act as a dispersant, activator, depressant, or means of pH control. Metal ions are believed to alter the mineral surfaces by being adsorbed to these surfaces. Organic colloids are thought to blank off surfaces. Organic acids may aid in activation, depression, or cleaning surfaces.

The pH-controlling agents employed have been described under EXPERIMENTAL PROCEDURE. Other modifiers used in the experiments are: 1) three to provide ion resurfacing - FeSO<sub>4</sub> for iron, Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> for aluminum, and HF for fluorine; 2) two organic colloids to blank mineral surfaces - starch and dextrin; and 3) one organic acid - citric acid.

#### RESULTS

The object in the method of presenting the results is usability. Ready comparison of the results for individual minerals is necessary in order to facilitate selection of the liquid systems in which separations would be likely to take place. In this report, the estimations for each mineral have been condensed so that they will fit onto the front and back of of a single sheet. Each sheet may be removed from the report so that it may be laid beside another, or so that two or more may be overlapped, for easier comparison.

For each mineral, floatation is estimated according to collectors on the front of a page, and according to modifiers on the back of a page. To make comparison as easy as possible, the floatability is depicted by four symbols - black for excellent, cross-hatched for good, stippled for fair, and blank for poor. Thus the darker areas, black and cross-hatched, generally indicate acceptable floatation, whereas the lighter areas, stippled and white, indicate unsatisfactory flotation at best.

In the initial recording of observations as described under EXPERIMENTAL PROCEDURE, the number of particles adhering to the bubble and the strength of the bond were indicated. Thus, 15G stood for about 15 particles picked up and firmly held by the bubble (good bond). If the record was 15F, then some or most of the particles could be dislodged by a light tapping on the bubble tube (fair bond). A record such as 3P would indicate 3 particles picked up and either dropping from the bubble after a moment or two or dropping after a light tap on the bubble tube (poor bond).

In assessing the results for this method of presentation, one has to use a form of weighted average. This consists of arbitrarily allotting the number 1 to poor bond, 2 to fair bond and 3 to good bond, then multiplying the number of particles by these numbers. The aggregate of all cases divided by the number of cases is then used to determine the average floatability. An average below 10 is considered to be poor between 10 and 25 fair, 25 to 40 good, and above 40 excellent. In the occasional doubtful case between poor and fair, or between fair and good, floatability has been indicated by using both symbols. Results which differ greatly from the average have not been used in the calculation but are pointed out as exceptions.

Results are presented for the various minerals in alphabetical order, and each sheet is dated so that, with subsequent extensions of the work, the latest and most up-to-date sheets will be retained.

[	lent Flotation			r Flotation		
CLA S	S No. Tes	ted	Acid	Neutral	Basic	
Non-io	mic 2					
Anioni	с 12					
Cation						
Fatty	acid 4					
Class	TYPE	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3				
"	Fatty acid sulphonate	3				
n	Alkyl sulphate	3				
. "	Sodium taurate	3				
Cationic	Primary amine	3				
11	Diamine	3				
и	Tertiary amine	3				
tt.	Quaternary	4				
Fatty acid	Talloil	2				
11	Oleic	2				

	No Modifier	Fe	Al	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON IONI		Ethano		A		
Acid Neutral Basic	ANIONIC	<u> </u>	Petrol	eum Sulpho	onate		
Acid Neutral Basic	ANIONIC	E	Fatty /	Acid Sulph	onate		
Acid Neutral Basic	ANIONIC	F	Alkyl	Sulphate	П	B	
Acid Neutral Basic	ANIONIC		: • :	Taurate	H		
Acid Neutral Basic	CATIONIC		Primar	y Amine			
Acid Neutral Basic	CATIONIC		Diamin		1 1	1	
Acid Neutral Basic	CATIONIC	**		ry Amine	<b>1</b> 2.88		
Acid Neutral Basic	CATIONIC		Quater		<b>**</b>		:::
Acid Neutral Basic	FATTY AC		Talloi	:::			
Acid Neutral Basic	FATTY AC		Oleic	B			

1. Poor flotation with Duomeen L-11

Fair Flotation Excellent Flotation KEY: 羉 Poor Flotation Good Flotation Basic CLASS No. Tested Acid Neutral 2 Non-ionic ::: 12 Anionic 13 Cationic ::: 🕸 ::: Fatty acid 4 No. Tested TYPE Acid Neutral Basic Class 2 Non-ionic Ethanol 3 Petroleum Anionic sulphonate **888** Fatty acid 3 sulphonate Alkyl 3 **333** sulphate **33** 3 Sodium ::: taurate ::: 88 3 Cationic Primary amine 533 Diamine  $\Box$ :: Tertiary 3 amine Quaternary ::: Fatty acid Talloil \*\* 2 ::: Oleic **883** 

	•					•	
	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION	Е	Kthan	ol			E H
Acid Neutral Basic	ANIONIC			leum Sulpho	1 1 1	::: :::	
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Acid Neutral Basic	ANIONIC 2	<b>※</b>	Alkyl :::	<u>}.:</u>	:: <b>33</b>	:: <u>:</u>	
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Acid Neutral Basic	CATIONIO			ry Amine			5
Acid Neutral Basic	CATIONIC CATIONIC		Diamin				
Acid Neutral Basic		::: :::	6	ary Amine		: : : : : :	7
Acid Neutral Basic	CATIONIC	:-:	Quater	8	: · : : · :	: · : : · :	::: :::
Acid Neutral Basic			Talloi	:::		**	∷: ::: <b>:::</b>
Acid Neutral Basic	FATTY AC		Oleic	:: <b>: ※</b>	::: ::: <b>※</b>		:::
1 Poor flot: 2 Excellent 3 Poor 4 Good 5 Poor 6 Good 7 Poor 8 Excellent	flotation	with Nutrap " Igepor " Armeer " Nalcom	TE 42 TE 42 L-9 nine G 11 G 11		Pata	October 1	
8 Excellent		Emco1	E 607-40		Date	oo toner, t	700,,,,,,

	ent Flotation lotation	<b>3</b>		r Flotation		
CLASS	No. Tes	ted	Acid	Neutral	Basic	
Non-ion:	ic 2					
Anionic	12					
Cationi	c 13		:::	$\Box$		
Fatty a	cid 4					
Class	TYPE	No. Tested	Acid	Neutral	Basic	
With the same of t	*******		- ACIU	Nouclai	DASIC	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3	$\Box$			
***	Fatty acid sulphonate	3				
**	Alkyl sulphate	3				
"	Sodium taurate	3				
Cationic	Primary amine	3	:::	:::	:::	
***	Diamine	3			:-;	
11	Tertiary amine	3				
11	Quaternary	4	:::			
Fatty acid	Talloil	2				
11	Oleic	2				

7.

## EFFECT OF MODIFIERS

	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION:	ic	Kthano			E	
Acid Neutral Basic	ANIONIC	 	— ₩	eum Sulpho		 ⊗ ⊗	
Acid Neutral Basic	ANIONIC		<b>₩</b>	Acid Sulph	onate		<b>**</b>
Acid Neutral Basic	ANIONIC	1 1 1	1 1	Sulphate	<b>※</b> ∷:		1 ※  ••
Acid Neutral Basic	ANIONIC			Taurate			
Acid Neutral Basic	CATIONIC		2 2 2	y Amine 2 2 2 2	2 2 2 2	2 2 3	2
Acid Neutral Basic	CATIONIC :: CATIONIC		Diamin  We Tertia	e  :::::::::::::::::::::::::::::::::::	<u>:</u> .:: 4	<b>₩</b> ::: :::	<b>**</b>
Acid Neutral Basic	CATIONIC		Quater	.·;			5
Acid Neutral Basic	FATTY AC		Talloi			6 7	7
Acid Neutral Basic	FATTY AC	:•:	Oleic		<b>**</b>	<b>**</b>	
Acid Neutral Basic		:::		<b>₩</b>	· · · · · · · · · · · · · · · · · · ·		
1. Poor flot 2. " 3. Excellent 4. Poor 5. Excellent 6. "	11 11	Armeen L-9 Adogen 170 Duomeen L- Nalcamine Ammonyx T	) ) A -11	•			

October 1968

Emcol E 607-40

	ent Flotation lotation			ir Flotation		
CLASS		sted	Acid	Neutral	Basic	
Non-i on						
Anionic	12					
Cationi			<b>888</b>		<b>883</b>	
Fatty a	cid 4					•
Class	TYPE	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3	:: <b>:</b> ₩			
***	Fatty acid sulphonate	3	:::			
ii	Alkyl sulphate	3	:::			
11	Sodium taurate	3	<b>∷</b>			
Cationic	Primary amine	3	· ·			
11	Diamine	3				
n .	Tertiary amine	3			<b>∷</b>	
H .	Quaternary	4	$\overline{\cdots}$			
Fatty acid	Talloil	2				
***	Oleic	2				

	y			T			
	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION	c 📙	R'thano				
Acid Neutral Basic	ANIONIC		Petrol	eum Sulpho	nate	::: 	<u>:::</u>
Acid Neutral Basic	ANIONIC		Fatty 1	Acid Sulph	onate	<b>∷</b> ⊗	<u>: : :</u>
Acid Neutral Basic	ANIONIC	3	3	Sulphate	<u>::</u>	4	
Acid Neutral Basic	ANIONIC			Taurate			:35
Acid Neutral Basic	CATIONIC	: ·: : ·:		y Amine 5			
Acid Neutral Basic	CATIONIC 6 CATIONIC	6 6 6	Diamin 6 7	·	<u>::</u>	:3 <b>※</b> :3	
Acid Neutral Basic	[388		<u>∷:</u> 8	ry Amine	∷ <b>※</b>	9	
Acid Neutral Basic	CATIONIC FATTY AC		Quater		10	10 10 10	
Acid Neutral Basic	FATTY AC		Talloi				
Acid Neutral Basic	FATTY AC		Oleic				
6. " ' 7. " ' 8. Good ' 9. Poor '		Prestabit (Emcol 4510) Duponol 80 "" Armeen L-9 Duomac T Duomeen L- Valcamine ("" Emcol E 60	l1 G-11		Date ,, Ç	October, 196	8,,,,,

KEY: Excelle	nt Flotation otation	<b>***</b>		r Flotation r Flotation		
CLASS Non-ioni Anionic Cationic Fatty ac	12 13	sted_	Acid	Neutral	Basic	
Class	TYPE	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3				
tt	Fatty acid sulphonate	3				
. 11	Alkyl sulphate	3				
tt.	Sodium taurate	3		<i>1</i> 25		
Cationic	Primary amine	3				
***	Diamine	3				
tt	Tertiary amine	3				
11	Quaternary	4				
Fatty acid	Talloil	2	<u>:::</u>			
11	Oleic	2	<u>::</u>			

			<u> </u>				·
	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON IONI		Kthano	1 <sub>.</sub> 日	E	A	
Acid Neutral Basic	ANIONIC	H	1	eum Sulphor			且
Acid Neutral Basic	ANIONIC		Fatty	Acid Sulpho	onate		且
Acid Neutral Basic	ANIONIC			Sulphate			B
Acid Neutral Basic	ANIONIC	B	Sodium 2 2 2	Taurate		A	
Acid Neutral Basic	CATIONIC		Primar 3	y Amine			
Acid Neutral Basic	CATIONIC		Diamin <b>₩</b>		8 <b>2.</b> €		
Acid Neutral Basic	CATIONIC			ry Amine			
Acid Neutral Basic	CA TION IC		Quater 4 4	4	4	4	
Acid Neutral Basic	FATTY AC		Talloi				
Acid Neutral Basic	FATTY AC		Oleic				

1	Poor	flotation	with	Reagent	201
1.	TOOT	IIO LA LI OII	M T CII	neagent	COL

- Igepon TE 42 2. Good " 3. Poor " 4. " "
- Armeen L-9
- Retarder LA

. KEY: Excelle	E3	-				
CLASS  Non-ion: Anionic Cationic Fatty ac	12 c 13	ted_	Acid	Neutral	Basic	
Class	TYPE	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3	:-:			
	Fatty acid sulphonate	3				
	Alkyl sulphate	3		:::	:::	
tt	Sodium taurate	3				
Cationic	Primary amine	3				
***	Diamine	3	::	::	::: <b>:</b>	
11	Tertiary amine	3				
***	Quaternary	4				
Fatty acid	Talloil	2				٠
tt	Oleic	2				

<del></del>	·						
	· No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON IONIC	e H	R'thano		H	目	
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Acid Neutral Basic	ANIONIC		Fatty 3 3	Acid Sulph	onate	: ·:	
Acid Neutral Basic	ANIONIC		Alkyl	Sulphate 		<u>::</u>	<b>∴</b> ₩
Acid Neutral Basic	ANIONIC	∴ 🕸	Sodium	Taurate	:: <del>**</del>		
Acid Neutral Basic	CATIONIC 4 4 4	4 4 4	::;	y Amine 4 4	<b>∴</b>	<b>33</b> 4 4 4 4 4	<b>3</b> 4 <b>4 4 4</b>
Acid Neutral Basic	CATIONIC  5  CATIONIC  CATIONIC	::: :::	Diamin	e 5 ry Amine	<b>※</b> 5 5 <b>※</b> 5 5	5 \$\$\frac{5}{5}\$	
Acid Neutral Basic			B				:.: :::
Acid Neutral Basic	CATIONIC 6	6	Quater	6 6 6	6	6 6	6 6
Acid Neutral Basic		<u></u>	Talloi				
Acid Neutral Basic	FATTY AC	ID <b>***</b>	Oleic				
1. Poor 2. Good 3 " 4. Poor	flotation w	" Petrof " Dupono " Armeer	1ote 462 o1 80 o L-9				
5 " 6 Excellent		Duomee	en L-11 E 607-40		Date Oc	tober 1968	

	ent Flotation			r Flotation	
CLASS	No. Tes	ted	Acid	Neutral	Basic
Non-ion:	ic 2				
Anionic	12				
Cationio	13	[	∷:⊗	$\overline{\cdot \cdot \cdot}$	
Fatty a	eid 4				
Class	TYPE	No. Tested	Acid	Neutral	Basic
Non-ionic	Ethanol	2			
Anionic	Petroleum sulphonate	3			
tı	Fatty acid sulphonate	3			
II.	Alkyl sulphate	3			
***	Sodium taurate	3			
Cationic	Primary amine	3		::: <b>:</b>	
11	Diamine	3		:::	
11	Tertiary amine	3	:: <b>**</b>	<b>::</b> :	<b>∷</b> :
**	Quaternary	4			:·:
Fatty acid	Talloil	2			
	Oleic	2			

				·····			<del></del>
	No Modifier	Fe	Al	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON IONI		K'thano				
Acid Neutral Basic	ANIONIC		1	eum Sulpho		<b>₩</b>	
Acid Neutral Basic	ANIONIC		Fatty	Acid Sulph	ona te		
Acid Neutral Basic	ANIONIC 2 2	<b>₩</b> 2	∷	Sulphate		2 2	<b>×</b>
Acid Neutral Basic	ANIONIC			Taurate			
Acid Neutral Basic	CATIONIC			y Amine	;·;	i: ×	×
Acid Neutral Basic	CATIONIC CATIONIC		Diamin W			3 :∙:	
Acid Neutral Basic	CATTONIC		···	ry Amine	: :: : : :	5 :·:	: : <b>※</b> : : : :
Acid Neutral Basic	CATIONIC		Quater		:·; 6 :::	::: 	: :: <b>※</b>
Acid Neutral Basic	FATTY AC	***	Talloi	:::			:::
Acid Neutral Basic	FATTY AC	1D	Oleic	::	<b>**</b>		
1, Poor 2, " 3, " 4, " 5, Excellen 6, "	flotation "" t " "	" Dupc " Duon " Arme " Arme	al SA onol 80 neen L-11 een DLM-11 een 12D ol E 607-40				

	KEY: Excellent Flotation Fair Flotation  Good Flotation Poor Flotation							
CLASS Non-ioni Anionic Cationic Fatty ac	12 2 13	ted	Acid	Neutral	Basic			
Class	TYPE	No. Tested	Acid	Neutral	Basic			
Non-ionic	Ethanol	2						
Anionic	Petroleum sulphonate	3			:::			
tt	Fatty acid sulphonate	3			:::			
tt	Alkyl sulphate	3	:: 🐯					
**	Sodium taurate	3						
Cationic	Primary amine	3		:::	:::			
tt	Diamine	3	:::		:::			
11	Tertiary amine	3		:::				
*11	Quaternary	4	:::		::3			
Fatty acid	Talloil	2						
и	Oleic	2	:::					

***************************************	·			·			
:	No Modifier	Fe	A1	F	Starch .	Dextrin	Citric Acid
Acid Neutral Basic	NON ION	ic	H'thano		E .	A	
Acid Neutral Basic	ANIONIC	<b>**</b>	Petrol	eum Sulphon	nate	<b>※</b>	<b>**</b>
Acid Neutral Basic	ANIONIC :·:	:::	Fatty	Acid Sulpho	onate	<b></b>	<b>₩</b>
Acid Neutral Basic	ANIONIC		Alkyl	Sulphate		⊡₩	1 1
Acid Neutral Basic	ANIONIC 2	2 :•:	Sodium 3	Taurate	2 5		
Acid Neutral Basic	CATIONIC		Primar	y Amine		: · : : · : : · :	<b>XX</b>
Acid Neutral Basic	CATIONIC :::	:::	Diamin				
Acid Neutral Basic	CATIONIC		Tertia	ry Amine			
Acid Neutral Basic	CATIONIC	<u> </u>	Quater	: : : : : :			:::
Acid Neutral Basic	FATTY AC	4 <b>₩</b>	Talloi	1 .::4 .::4	4	4 4	∴ 4 ∴ 4
Acid Neutral Basic	FATTY AC	ЭТЬ	Oleic 4	4	::: 4 ::: 4 ::: 4	4 ••••4	4

<sup>1.</sup> Poor 2. " flotation with Duponol OS

Date ... Oc.tober, 1968 ......

<sup>2,</sup> Igepon TN-74

<sup>3.</sup> Excellent Igepon TE-42

<sup>4.</sup> Response inconsistent (poor to excellent)

	ent Flotation		•	ir Flotation	
CLASS	No. Tes	ted	Acid.	Neutral	Basic
Non-ion	ic 2				
Anionic	12				<b>∷:</b>
Cationi	e 13			:::	:::
Fatty a	cid 4		:::		
Class	туре	No. Tested	Acid	Neutral	Basic
Non-ionic	Ethanol	2			
Anionic	Petroleum sulphonate	3			
11	Fatty acid sulphonate	3	-		:-: 1
11	Alkyl sulphate	3			:::
11	Sodium taurate	3			
Cationic	Primary amine	3			<b>∷:</b> ‱
11	Diamine	3		:::	::
***	Tertiary amine	3	:::	:::	:::
tt	Quaternary	4		:::	:::
Fatty acid	Talloil	2	:::	<b>1000</b>	<b>****</b>
11	Oleic	2			

,	No , Modifier	Fe	A1	F	Starch	Dextrin	Citrie Acid
Acid Neutral Basic	NON ION	rc	Kthanol			H	
Acid Neutral Basic	ANIONIC	∷ <b>※</b> ∷	Petrole	eum Sulpho	nate 	₩ ₩	<b>₩</b>
Acid Neutral Basic	ANIONIC	:•:	Fatty A	Acid Sulph	onate	··;	<b>※</b>
Acid Neutral Basic	ANIONIC		Alkyl S	Sulphate 2	$\begin{bmatrix} 1\\1\\1\\1 \end{bmatrix}$	:::	
Acid Neutral Basic	ANIONIC	<b>※</b> 3	Sodium ::: :::	Taurate			
Acid Neutral Basic	CATIONI	C	Primary 4	y Amine	  ⊗	4	[∷.
Acid Neutral Basic	CATIONI	::	Diamine 	<b></b>		<b>**</b>	W:
Acid Neutral Basic	CATIONI	C <b>※</b> ∴ 5	Tertian	ry Amine	<b>⊠ ⊗</b>	<b>※</b>	<u>:::</u>
Acid Neutral Basic	CATIONI		Quateri  				
Acid Neutral Basic	FATTY A		Talloil	:::			::: :::
Acid Neutral Basic	FATTY A	CID	Oleic :::		∷ <b>⊗</b> ∷	<b>™</b> ⊗	:: ::

<sup>1.</sup> Excellent flotation with Nutrapon WAC 2. Poor "Duponol 80

11

11

<sup>2.</sup> Poor

Duponol 80

<sup>3.</sup> Excellent

<sup>11</sup> Igepon TE-42

<sup>4.</sup> Poor

<sup>11</sup> Armeen L-9

<sup>5.</sup> Good

<sup>11</sup> Nalcamine G-11

	ent Flotation lotation		Fair Flotation ::  Poor Flotation					
<del></del>	CLASS No. Tes		Acid	Neutral	Basic			
Anionic			<b></b>	<b>\times</b>				
Cationi Fatty a			::: :::					
Class	ТҮРЕ	No. Tested	Acid	Neutral	Basic	1.4 Table 1.4 Ta		
Non-ionic	Ethanol	2						
Anionic	Petroleum sulphonate	3						
tt	Fatty acid sulphonate	3						
"	Alkyl sulphate	3	:::					
11	Sodium taurate	3	:·:	:::				
Cationic	Primary amine	3	<u>:::</u>	$\odot$	:::			
u ·	Diamine	3	<b>∷</b> :	:::	:·:			
11	Tertiary amine	3	:::	:: <b>:</b>				
tt	Quaternary	4	:::					
Fatty acid	Talloil	2		::3				
ti .	Oleic	2	<b>∷</b> :		<b>∷</b> :			

	No	, Fe	A1	F	Starch	Dextrin	Citric
	Modifier						Acid
Acid Neutral Basic	NON IONI		Ethano				E
Acid Neutral Basic	ANIONIC			eum Sulpho	:::		
Acid Neutral Basic	ANIONIC		Fatty :::	Acid Sulph	onate 2	<b>*</b>	
Acid Neutral Basic	ANIONIC	: : : : : :	A1ky1	Sulphate		<b>:·:</b>	<u>::</u>
Acid Neutral Basic	ANIONIC			Taurate	<b>88</b>	3	::: :::
Acid Neutral Basic	CATIONIC		Prima	Amine	· · · · · · · · · · · · · · · · · · ·		<b></b>
Acid Neutral Basic	CATIONIC 	::: <b>※</b> : : : : : :	Diamir	: · : : · :	4	<b>∷ ⊗</b> <b>∶∶:</b> 5	
Acid Neutral Basic	CATIONIC		: : : · :	ary Amine	:::	::: :::	
Acid Neutral Basic	CATIONIC	[::] [:::]	Quater	 ₩	∷ ∷:	<u>; . ;</u> ; . ;	:::
Acid Neutral Basic	FATTY AC	6 6	Tallo	i1 6	6		<b>∷</b> : 6
Acid Neutral Basic	FATTY AC		Oleic	6 6			

- 1. Excellent flotation with Prestabit Oil V
- 2. " Emcol 4510
- 3. " " Igepon TE-42
- Tagopon 23
- 5 Poor " Duomeen L-11
  - 6. Response Inconsistent (Poor to Excellent)

	KEY: Excellent Flotation Fair Flotation  Good Flotation Poor Flotation								
CLASS Non-ioni Anionic Cationic Fatty ac	12	ted_	Acid	Neutral  :::  :::	Basic  ::: :::				
Class	ТУРЕ	No. Tested	Acid	Neutral	Basic				
Non-ionic	Ethanol	2							
Anionic	Petroleum sulphonate	3		:::	::: <u>:</u> :				
n	Fatty acid sulphonate	3		:::₩					
u	Alkyl sulphate	3							
***	Sodium taurate	3	∷⊠	∷₩	:::				
Cationic	Primary amine	3	:::	:::	<b>∷</b> ₩				
11	Diamine	3	: :: ₩						
11	Tertiary amine	3							
11	Quaternary	4							
Fatty acid	Talloil	2			:: <b>:</b>				
11	Oleic	2		:::					

	<b>V</b>		· · · · · · · · · · · · · · · · · · ·				
	No Modifier	Fe	Al	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION		Kthano		A	E	
Acid Neutral Basic	ANIONIC		Petrol 1	eum Sulphon	nate	:.: :::	<u>: : :</u> : : :
Acid Neutral Basic	ANIONIC	1. t 1. t 1. t		Acid Sulpho	onate 3	::: ∷:: <b>※</b>	2
Acid Neutral Basic	ANIONIC		4	Sulphate :::			F
Acid Neutral Basic	ANIONIC	:: :::	Sodium :::	Taurate  	:: :::	: · : : · :	
Acid Neutral Basic	CATIONI	c XX	Primar 5	y Amine 5 5 5	:: <b>:</b> :::		[:::
Acid Neutral Basic	CATIONI	6	Diamin		7		8 7
Acid Neutral Basic	CATIONIO			ry Amine			
Acid Neutral Basic	CATION IO	10 10 10	Quater 10	10 10	10	10	
Acid Neutral Basic	FATTY AC	∷: <b>※</b>	Talloi :::			<b>**</b>	:::
Acid Neutral Basic	FATTY AC	: · : : · :	Oleic :::		<b>&amp;</b> :∵:		···;
1. Excellent 2. " 3. Poor 4. Excellent 5. Poor 6. " 7. Excellent 8. " 9. " 10. "	11 11 11 11	" Emcol " Presta " Dupona " Armeei " Duomaa " Duomaa " Duomaa " Nalcan	4510 abil Oil V ol 80 n L-9 c C		Date	October 19	968

KEY: Excelle Good F1	nt Flotation otation			r Flotation		
CLASS	No. Tes	ted	Acid	Neutral	Basic	
Non-ioni	c 2					
Anionic	12	_	<b>88</b>	:::		
Cationic	13	Ĺ	<u>∷</u> 888			
Fatty ac	id 4		:::	<b>::</b>		
Class	ТУРЕ	No. Tested	Acid	Neutra <b>l</b>	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3				
l ; tı !	Fatty acid sulphonate	3				
11	Alkyl sulphate	3				
11	Sodium taurate	3	:::			
Cationic	Primary amine	3		∷፠		
tt	Diamine	3	:::88			
tt	Tertiary amine	3	:::			
11	Quaternary	4	<b>∷</b>			
Fatty acid	Talloil	2	:-:	$\Box$		
11	Oleic	2	:::	:•:		

	· · · · · · · · · · · · · · · · · · ·	·		<del> </del>		<del></del>	<del></del>
	No Modifier	Fe	Al	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION	ic	R'than	ol	H		
Acid Neutral Basic	ANIONIC		Petro 1	leum Sulpho	onate 2	2	
Acid Neutral Basic	ANIONIC	. 🕸	12.1	Acid Sulph	onate 3	3	:·: <b>:</b>
Acid Neutral Basic	ANIONIC	<b>₩</b>		Sulphate			<b>※</b>
Acid Neutral Basic	ANIONIC		4	m Taurate	5		<b>≅</b>
Acid Neutral Basic	CATIONI	с <b>Ж</b>	Prima	ry Amine	::: ::: ₩ :::		
Acid Neutral Basic	CATIONI ::: 888 6 7		Diami		<b>∷</b> 6 6 6	6 :::	<b>:</b> :::::::::::::::::::::::::::::::::::
Acid Neutral Basic	CATIONI		Terti	ary Amine	::: <b>88</b>	<b>∷</b> :	
Acid Neutral Basic	CATIONI		Quate	**	  		: · ; : · ;
Acid Neutral Basic	FATTY A		Tallo			9	
Acid Neutral Basic	FATTY A	CID 9	Oleic		9	9 9	:::\bigs
1 Poor flot 2 " 3 Excellent 4 Poor	" "	Reagent 80 Leonal SA Emcol 4510 Igepon TN-					

Date . December . 1968. . . . . . .

<sup>&</sup>quot; Igepon TE-42 5 Excellent " " Duomeen L-11 6 Poor

<sup>11</sup> 7 Duomac T

Excellent " 8 Nalcamine G-11

Response inconsistent (Poor to excellent)

	ent Flotation			r Flotation		
CLASS  Non-ioni  Anionic  Cationic  Fatty ac	12	<u>ted</u> [	Acid	Neutral	Basic	
<u>Class</u> Non-ionic	<u>TYPE</u> Ethanol	No. Tested	Acid	Neutral	Basic	
Anionic	Petroleum sulphonate	3	<u> </u>			
11	Fatty acid sulphonate	3				
11	Alkyl sulphate	3	:::	<b>::</b> :		
ti .	Sodium taurate	3				
Cationic	Primary amine	3	:::			
11	Diamine	3		∷₩		
11	Tertiary amine	3		:::		
11	Quaternary	4	<u>:::</u>	:::	$\mathbb{R}^{\otimes}$	
Fatty acid	Talloil	2				
tt	Oleic	2	<b>∷</b> :			

	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON IONI		<b>K</b> thano			E	
Acid Neutral Basic	ANIONIC		Petrol	eum Sulpho	nate		
Acid Neutral Basic	ANIONIC			Acid Sulph	onate	1	
Acid Neutral Basic	ANIONIC	E	A	Sulphate	F		F
Acid Neutral Basic	ANIONIC	B	Sodium	Taurate	F	A	F
Acid Neutral Basic	CATION IO		Primar	y Amine	· · · · · · · · · · · · · · · · · · ·		:: <b>33</b>
Acid Neutral Basic	CATIONIC 2 2	2 ∴:	Diamin :::			::: 3	
Acid Neutral Basic	CATIONIC	:::	Tertia	ry Amine	:::	:::	: ·: : · :
Acid Neutral Basic	CATION IO	4	Quater :		; · ; ; · ;	:.; :::	<u> </u>
Acid Neutral Basic	FATTY AC		Talloi				
Acid Neutral Basic	FATTY AC	CID CID	Oleic				

1.	Excellent	flotation	with	Adogen	170A

2. " " Duomac T

3. Poor "Duomeen L-11

4. Excellent " ARQUAD 12

Date ... December 1968

	ellent Flotation d Flotation			ir Flotation		
<u>C</u> L	ASS No. T	'ested	Acid	Neutral	Basic	
Non-	ionic 2	}				
Anio	nic 12	1				
Cati	onic 13	}	::!	:::	:::	
Fatt	y acid 4		:-:			
		`				
Class	TYPE	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3				
n	Fatty acid sulphonate	3				
11	Alkyl sulphate	3			∷⊗	•
11	Sodium taurate	3				
Cationic	Primary amine	3		<b>:::</b>		
11	Diamine	3	∷ඎ			
***	Tertiary amine	3				
***	Quaternary	4				i
Fatty acie	d Talloil	2	···			
11	Oleic	2	:::			

	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION		R'thano				
Acid Neutral Basic	ANIONIC	**		eum Sulpho			
Acid Neutral Basic	ANIONIC		Fatty	Acid Sulph	onate		
Acid Neutral Basic	ANIONIC		A1kyl	Sulphate	<b>₩</b>		
Acid Neutral Basic	ANIONIC	::: ::: 1	Sodium	Taurate	**		
Acid Neutral Basic	CATION IO		Primar	y Amine	::: ::: 2	∷ 2 2	.∵ 
Acid Neutral Basic	CATIONIO	3	Diamin 4 3	::: ::: ::::	∷: <b>⊗</b> ∷:	  3	∴: <b>※</b> ∴: ∴:
Acid Neutral Basic	CATIONIO			ry Amine			:.: :::
Acid Neutral Basic	CATIONIO 5 5 5	5 5 5	Quater 5 5 5	5 5 5	5 5 5	5 5 5	5 5 5
Acid Neutral Basic	FATTY AC	∷ <b>※</b>	Talloi		<b>**</b>		
Acid Neutral Basic	FATTY AC		Oleic		<b>※</b>		
1 Poor 2, " 3. " 4 Excellent 5. "	11	" Duome	en L-9 en L-11				

	ent Flotation	<b></b>		r Flotation		····
CLASS  Non-ioni  Anionic  Cationic  Fatty ac	12 **	ed_	Acid	Neutral	Basic  Compared to the compare	
Class	ТҰРЕ	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3	:::			
11	Fatty acid / sulphonate	3				
11	$^{\rm Alkyl}_{\rm sulphate}  V$	3				
11	Sodium / taurate	3				
Cationic	Primary amine	3	∷⊠	:::	:::	
11	Diamine	3	:::	::;		
11	Tertiary amine	3	::		:::	
11	Quaternary	4				
Fatty acid	Talloil $ u$	2				
11	Oleic V	2				

	<del></del>						
	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION	ic 	Ethano			A	
Acid Neutral Basic	ANIONIC	::: 	Petrol	eum Sulpho	nate		
Acid Neutral Basic	ANIONIC		Fatty /	Acid Sulph	onate		
Acid Neutral Basic	ANIONIC		Alkyl	Sulphate			
Acid Neutral Basic	ANIONIC		:::	Taurate			
Acid Neutral Basic	CATIONIO		Primary 2	y Amine	::: :::	**	;·:
Acid Neutral Basic	CATIONIC	4 4	Diamine		::: :::	<b>₩</b>	<b>₩</b>
Acid Neutral Basic	CATIONIC .:		: · : <b>※</b>	ry Amine	<b>※</b> : : :	<b>※</b>	Not gas
Acid Neutral Basic	CATIONIC		Quaterr		∷ ∷: ⊗		
Acid Neutral Basic	FATTY AC	::;	Talloi	B		H	
Acid Neutral Basic	FATTY AC	ETD ETD	Oleic				H
1, Good flot	ation with	Duponol 80					

1. Good flotation	with	Duponol	80	)
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December 1968

<sup>2.</sup> Poor Adogen 170 A 3.

Duomeen L-11

<sup>4,</sup> Duomac T

	KEY: Exceller	nt Flotation			ir Flotation		
	CLASS Non-ionio	No. Tes	sted	Acid	Neutral	Basic	
i	Anionic Cationic Fatty ac	12 13 id 4					
-	<u>Class</u> Non-ionic	TYPE Ethanol	No. Tested	Acid	Neutral	Basic	
	Anionic	Petroleum sulphonate	3				
	***	Fatty acid sulphonate	3				
	11	Alkyl sulphate	3				
	п	Sodium taurate	3				
	Cationic	Primary amine	3	:::	∷₩		
	и.,	Diamine	3	:::			
	11	Tertiary amine	3				
	TT.	Quaternary	4	$\square \bigotimes$			
	Fatty acid	Talloil	2				
		Oleic	2				

				· · · · · · · · · · · · · · · · · · ·	·	
No. Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
NON ION		Ethano	01		1	
ANIONIC		Petro	Leum Sulpho	nate		
ANIONIC		Fatty 2	Acid Sulph	onate		
ANIONIC 3		A1ky1	Sulphate			F
ANIONIC	A					
CATION I	5 5	Primar 6 5	ry Amine	:::	[:: [:::₩	
<b>Fas</b> 7	7 7 <b>※</b> 7		<u>∵</u> <b>⊗</b>	∷ ∴ <b>⊗</b>	; · ; 7 ; · ;	<b>**</b>
<u>⊡</u> 88	1-1	: :: 		::: ::: 8	<b>846</b> <b>₩</b>	<b>**</b>
	9			10	10	□ <u>: : :</u> : : : : : : : : : : : : : : : :
flotation	" Sulpho " Dupono " Dupono	onated Cas ol 80 ol OS ol 170A	tor Oil			
	Modifier  NON ION  ANIONIC  ANIONIC  ANIONIC  CATIONIC  CATIONIC  CATIONIC  CATIONIC  TATTY ACTIONIC  FATTY ACTIONIC  FATTY ACTIONIC  III	Modifier  NON IONIC  ANIONIC  ANIONIC  ANIONIC  CATIONIC  CATIONIC  TO T	Modifier  NON IONIC  ANIONIC  ANIONIC  ANIONIC  ANIONIC  ANIONIC  CATIONIC  CATIONIC  CATIONIC  TOTAL  TOTA	Modifier  NON IONIC  ANIONIC  CATIONIC  TO T	Modifier  NON IONIC  Rethanol  ANIONIC  Sodium Taurate  CATIONIC  To a company  CATIONIC  Tertiary Amine  CATIONIC  Tertiary Amine  Tertiary	Modifier  NON IONIC  Rithanol  ANIONIC  Petroleum Sulphonate  ANIONIC  Fatty Acid Sulphonate  ANIONIC  ANIONIC  Alkyl Sulphate  ANIONIC  Sodium Taurate  CATIONIC  Formary Amine  CATIONIC  Tertiary Amine  CATIONIC  Tertiary Amine  CATIONIC  FATTY ACID  ACID Castor Oil  Bupponol 80  " Sulphonated Castor Oil  Duponol 80 " Adogen 170A

1	ent Flotation		Fair Flotation :			
CLASS Non-ion Anionic Cationi Fatty a	ie 2 12 c 13	ted	Acid	Neutral  Comparison of the com	Basic	
Class	TYPE	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
 	Petroleum sulphonate	3		. 🗀		
 	Fatty acid sulphonate	3				
1 	Alkyl sulphate	3				
i   	Sodium taurate	3				
Cationic	Primary amine	3		:::		
i 1 - n	Diamine	3	<b>:::</b>			
"	Tertiary amine	3	:::			
 	Quaternary	4				
Fatty acid	Talloil	2				
11	Oleic	2	<i>ι</i>			

	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid X Neutral Basic	NON IONI		Ethanol	A	H	A	B'
Acid \( \) Neutral Basic	ANIONIC	Ħ	Petrole	um Sulpho	onate	A	F
Acid Neutral	ANIONIC	且	Patty A	cid Sulph	nonate	H	A
Acid ( Neutral Basic	ANIONIC	A	Alkyl S	A	F	目	H
Acid Neutral Basic	ANIONIC	A	Sodium	Taurate	A	A	F
Acid Neutral Basic	CATIONIC	₩ / ·	Primary	Amine		₩ ₩	
Acid Neutral Basic	CATIONIC E::	1 1	Diamine	:::::::::::::::::::::::::::::::::::::::	:: :::		 
Acid Neutral Basic		::88	Tertiary 2	y Amine	BB (1)		**
Acid Neutral Basic	CATIONIC 3 3	::: ::::	Quaterna	ary 	 	::: <b>8</b> 88	
Acid Neutral Basic	FATTY ACT	A	Talloil	H	A	A	F
Acid Neutral Basic	FATTY ACI		Oleic	H	A	B	H
THE RESERVE AND THE PARTY OF TH						MET MAN OF THE REAL PROPERTY.	

<sup>1</sup> Poor flotation with Duomac T
2. Good " " Nalcamine G-11
3. Poor " ARQUAD 12

KEY: Excellent Flotation Fair Flotation  Good Flotation Poor Flotation						
CLASS Non-ion Anionic Cationic Fatty ac	ic 2 12 c 13	sted	Acid	Neutral	Basic	
Class	ТҮРЕ	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3				
11	Fatty acid sulphonate	3				
н	Alkyl sulphate	3				
11	Sodium taurate	3				
Cationic	Primary amine	3				
H	Diamine	3				
11	Tertiary amine	3				
11	Quaternary	4				
Fatty acid	Talloil	2				
11	Oleic	2				

<del></del>	<del>,</del>						
<b>P</b>	No Modifier	Fe	Al	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION		Ethano				
Acid Neutral Basic	ANIONIC		Petrol	eum Sulpho	onate		
Acid Neutral Basic	ANIONIC		Fatty	Acid Sulph	onate 2 2 2 2		2
Acid Neutral Basic	ANIONIC 3 3	3	A	Sulphate			
Acid Neutral Basic	ANIONIC		Sodium :::	Taurate			
Acid Neutral Basic	CATIONIO	: · : \$\$\$	Primar	y Amine	:::		
Acid Neutral Basic	CATIONIO 5	5 5 5	Diamin 5	**	::: ::: <b>※</b> :::		
Acid Neutral Basic	CATIONIC	**	Tertia	ry Amine	<b>※</b>	:∙: :•:	
Acid Neutral Basic	CATIONIO		Quater	6 6	∷: <b>※</b>	6	
Acid Neutral Basic	FATTY AC		Talloi				
Acid Neutral Basic	FATTY AC		Oleic				
1. Good 2. Fair 3. Good 4. Excellent 5. Poor 6. "	flotation w " " " " " "	" Emcol " Dupono " Duomac " Duomac	4150 1 80 C		Date ., .	December 19	68

	ent Flotation lotation	<b></b>		r Flotation or Flotation	
CLASS  Non-ioni  Anionic  Cationic  Fatty ac	12 c 13	sted	Acid	Neutral	Basic
Class Non-ionic	TYPE Ethanol	No. Tested	Acid	Neutral	Basic
Anionic	Petroleum sulphonate	3			
***	Fatty acid sulphonate	3			
n	Alkyl sulphate	3			
11	Sodium taurate	3			
Cationic	Primary amine	3			
11	Diamine	3			:::₩
н	Tertiary amine	3			∷⊠
11	Quaternary	4			
Fatty acid	Talloil	2			
tt.	Oleic	2			

	·		<del></del>	·		·	
	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON IONI	ic	Ethano	1		: : : <b>※</b>	
Acid Neutral Basic	ANIONIC			eum Sulpho			<u> </u>
Acid Neutral Basic	ANIONIC			Acid Sulph	onate		
Acid Neutral Basic	ANIONIC			Sulphate	<b></b>		
Acid Neutral Basic	ANIONIC	:::	: : : : : :	Taurate	1		
Acid Neutral Basic	CATIONIC			y Amine	2		
Acid Neutral Basic	CATIONIC CATIONIC	3 3	Diamin 3 3	ry Amine	3 4		<b>⊗</b> ! • ;
Acid Neutral Basic					<b>*</b>		::: ***
Acid Neutral Basic	CATIONIC		Quater		**		**
Acid Neutral Basic	FATTY AC		Talloi	1			<b>∷</b> <b>※</b>
Acid Neutral Basic	FATTY AC	CID	Oleic ::			<b>:::</b>	∷ ∷: <b>※</b>
1. Poor 2. "	flotation v	vith Igepon '' Armac					

2, " " Armac C

3. " Duomeen L-11

4. Excellent " Duomac T

	KEY: Excelle	nt Flotation otation			ir Flotation or Flotation	
	CLASS Non-ioni	No. Tes	ted	Acid	Neutral	Basic
	Anionic	12				
	Cationic	13		<u> </u>	<u> </u>	<u>.</u>
	Fatty ac	id 4		:::		
<del></del>	Class	TYPE	No. Tested	Acid	Neutral	Basic
	Non-ionic	Ethanol	2			
	Anionic	Petroleum sulphonate	3			
	**	Fatty acid sulphonate	3			:::
	***	Alkyl sulphate	3			
	u.	Sodium taurate	3			
	Cationic	Primary amine	3			
	11	Diamine	3	:::	:::	:::
	***	Tertiary amine	3			
	**	Quaternary	4	:::		:::
	Fatty acid	Talloil	2	:::		:::
	11	Oleic	2	$\Box$		:::

	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION	IC	Ethano	51		B	
Acid Neutral Basic	ANIONIC		Petrol	eum Sulpho	onate ::;	<b>:</b> -1	<b>:</b> ::
Acid Neutral Basic	ANIONIC			Acid Sulph	nonate		153
Acid Neutral Basic	ANIONIC	<b>₩</b>	Alkyl	Sulphate		2	
Acid Neutral Basic	ANIONIC			Taurate	3	3	
Acid Neutral Basic	CATIONI 5 5	5 5 	5 :·: \$ 5	y Amine 5 5	6	5 5 5	5 :::\times 5
Acid Neutral Basic	CATIONI CATIONI	<u>; : ;</u> . : : <b>⊗</b>	Diamin 7 8	ary Amine	<u>: · :</u> : · :	<u>: · :</u> : · :	::: :::: <b>※</b>
Acid Neutral Basic				:::			:: :: ::;
Acid Neutral Basic	CATIONI  9  FATTY A		Quater 9 Talloi	111		[:::] [:::]	: · :
Acid Neutral Basic	30.1 30.2	<b>₩</b>	: · : : · :	:::	::: <u>**</u>		: ·; <b>※</b>
Acid Neutral Basic	FATTY A	<b>∷ ⊗</b>	Oleic :::	::;	<b>∷ ⊗</b>		
2. Poor 3. " 4. " 5. "	11 11 11 11 11 11 11 11 11 11 11 11 11	Leonal SA Duponol 80 Igepon TN-7 Igepon TE-4 Armeen L-9	.2				

Date .... December .1968.....

Adogen 170 A

Duomac T Emcol E 607-40

Duomac C

6. Good 7. Poor

8, Good 9, "

	ent Flotation			r Flotation		
CLASS Non-ioni Anionic Cationic Fatty ac	12	ted_	Acid	Neutral	Basic  C C C C C C C C C C C C C C C C C C	
Class	ТҮРЕ	No. Tested	Acid	Neutral	Basic	
Non-ionic	Ethanol	2				
Anionic	Petroleum sulphonate	3				
п	Fatty acid sulphonate	3				
***	Alkyl sulphate	3				
"	Sodium taurate	3				
Cationic	Primary amine	3				
U	Diamine	3				
***	Tertiary amine	3				
ri .	Quaternary	4				
Fatty acid	Talloil	2				
11	Oleic	2				

	y		· · · · · · · · · · · · · · · · · · ·		·	<del></del>	
	No Modifier	Fe	A1	F	Starch	Dextrin	Citric Acid
Acid Neutral Basic	NON ION		Kthano		目		
Acid Neutral Basic	ANIONIC	E_	Petrol	eum Sulpho	onate		
Acid Neutral Basic	ANIONIC		Fatty	Acid Sulph	ionate		
Acid Neutral Basic	ANIONIC	A	Alkyl	Sulphate	П		
Acid Neutral Basic	ANIONIC	E	Sodium	Taurate			F
Acid Neutral Basic	CATIONIO	c <b>×</b>	Primar	y Amine		<b>**</b>	
Acid Neutral Basic	CATIONIO 1 2		Diamin	3	1 1	1	: · : : · :
Acid Neutral Basic	CATIONIO		Tertia	ry Amine	⊠፠	**	
Acid Neutral Basic	CATIONIO	<b>□</b>	Quater			,	
Acid Neutral Basic	FATTY AC		Talloi		A		E
Acid Neutral Basic	FATTY AC		Oleic				

1. Poor flotation with Duomeen L-11

2. Excellent flotation with Duomac T

3. Poor flotation with Duomac T

Date ... December 1968

#### CONCLUSION

The method of presenting results provides a ready comparison of the effect of similar liquid systems on different minerals, and indicates areas for possible selective floatation.

#### ACKNOWLEDGEMENT

All of the individual experiments (approximately 14,000) and observations were made by J.H. Colborne of the Industrial Minerals Milling Section. The success of the project is, in large measure, due to his careful and thorough workmanship.

