

Gold mineralization in the Cantung W-skarn deposit, Northwest Territories: An examination of distribution, mineralogy, and petrogenesis

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Abstract: The Cantung mine is a world-class W-skarn deposit; it is located just east of the Yukon border in the Selwyn Mountain Range of the Northwest Territories. The deposit area is within the southern extent of the polymetallic Tintina Gold Belt, which has many notable intrusion-related Au deposits. The extensive W skarns at Cantung were developed by hydrothermal fluids that, based on earlier research, were determined to be predominately supercritical magmatic brines with homogenization temperatures ranging from 270-500°C. Mineralization is composed of calcic exoskarn replacement of a clean limestone and lower grade replacements in a calc- silicate/chert unit; these occur in both the operating open pit and underground mine (the E Zone). The main sulphide identified petrographically is pyrrhotite, which is abundant in all skarn facies. Scheelite and chalcopyrite are dominant and there is locally abundant sphalerite. Native Bi exhibits textures indicative of forming later than the silicate assemblage in the paragenetic sequence, and it is decorated by bismuthinite, Bi tellurides, Ag tellurides, and Bi selenides. Tungsten and Cu are the main mine products, but the Au potential of the deposit merits further investigation.

This study characterized the distribution, mineralogy, and petrogenesis of Au mineralization by examining five skarn samples with bulk rock Au assay values >0.5 ppm taken from the E Zone. No free gold or electrum were identified petrographically or by SEM and FEG- SEM analyses. A positive correlation (Spearman's Rank, r^*) of Au with Bi (0.76), Ag (0.70), Fe (0.64), Cu (0.64), and Mo (0.60) was identified using the bulk rock geochemical data ($n = 48$). The strong correlation between Bi and Au is suggestive of a liquid bismuth collector mechanism for Au enrichment. However, LA ICP-MS analysis of native Bi and Bi alloys failed to reveal significant Au predicted by the liquid bismuth collector model. In contrast, the highest Au concentration was encountered in hessite (Ag_2Te) and other tellurides. Nano-inclusions within chalcopyrite and silicate minerals were also investigated using FEG-SEM for their Au content, but their composition consisted of native Bi. The decoration of native Bi by bismuthinite, Bi tellurides, Ag tellurides, and Bi selenides provides evidence for a late stage S-, Ag-, and Te-rich fluid. This fluid is thought to have remobilized the Au and deposited it as lattice bound invisible Au within the tellurides. This new data constrains Au exploration targets at Cantung to areas of altered skarn or where there is a presence of telluride minerals

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Gold mineralization in the Cantung W-skarn deposit, NWT:

AN EXAMINATION OF DISTRIBUTION, MINERALOGY, AND PETROGENESIS

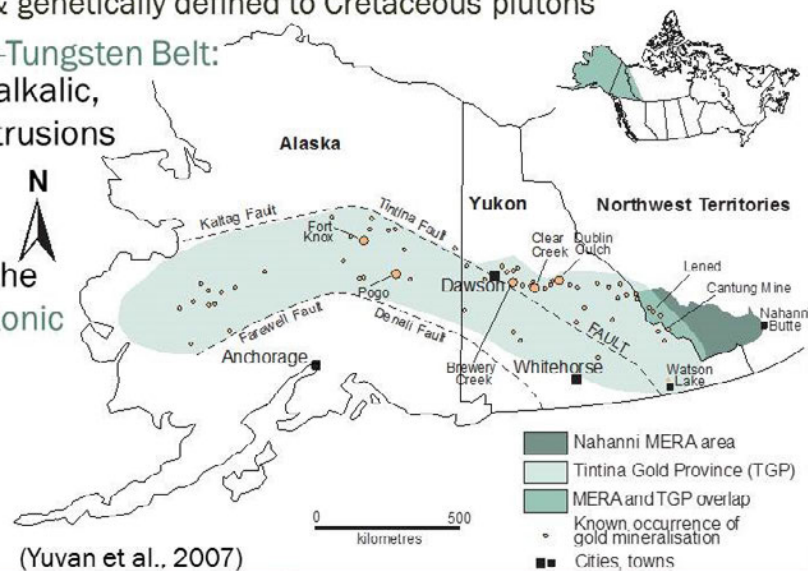
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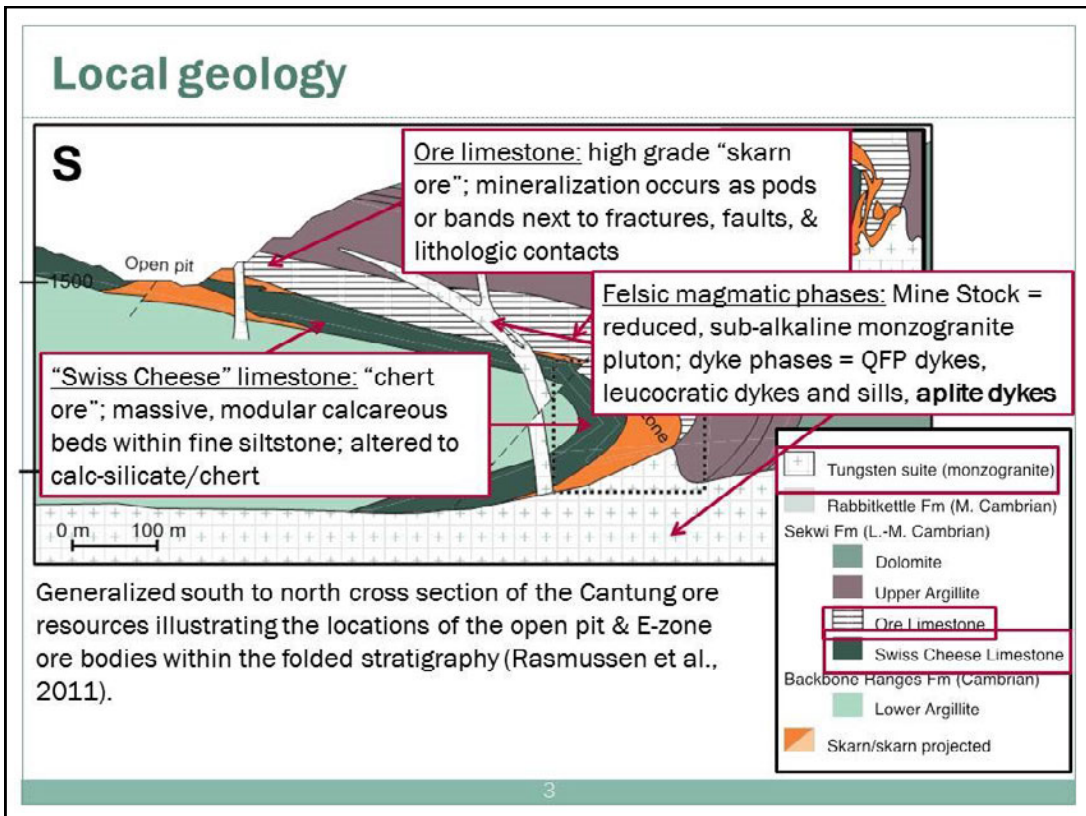
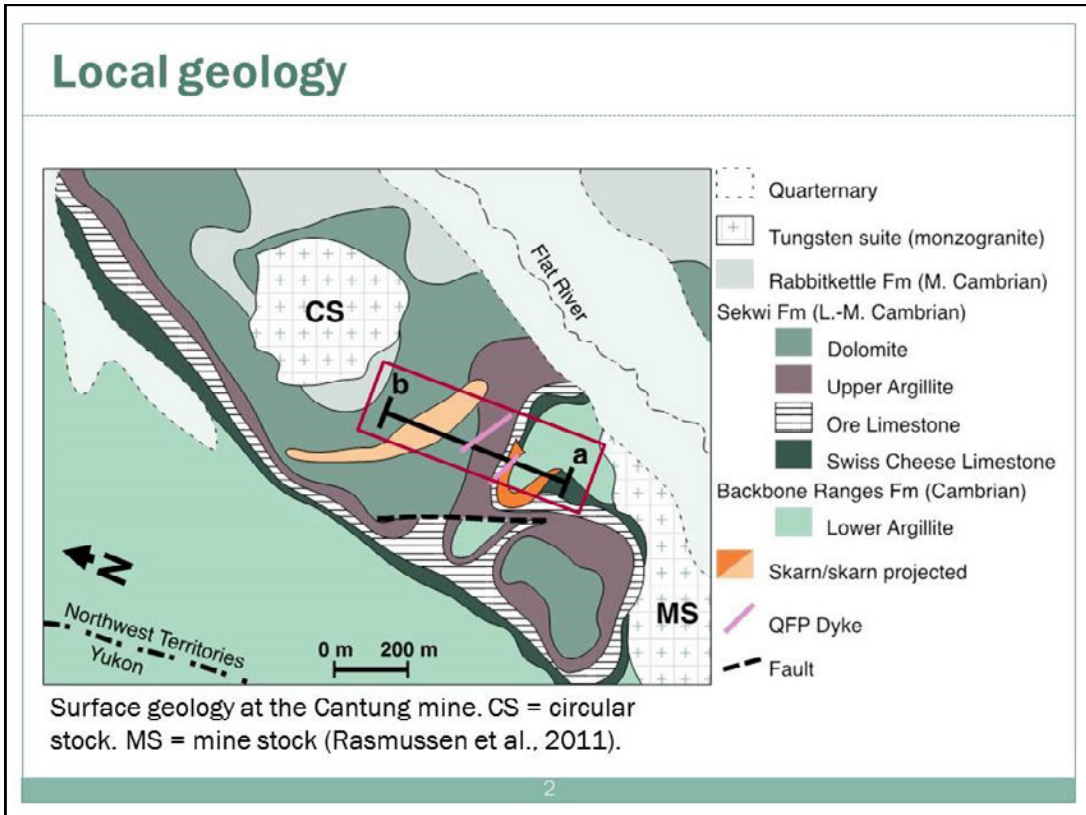
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Regional geology

- Tintina Gold Province (TGP): precious & base metals
 - Spatially & genetically defined to Cretaceous plutons
- Tombstone-Tungsten Belt: felsic, sub-alkalic, reduced intrusions
- S- to I-type granitic plutons of the Selwyn plutonic suite

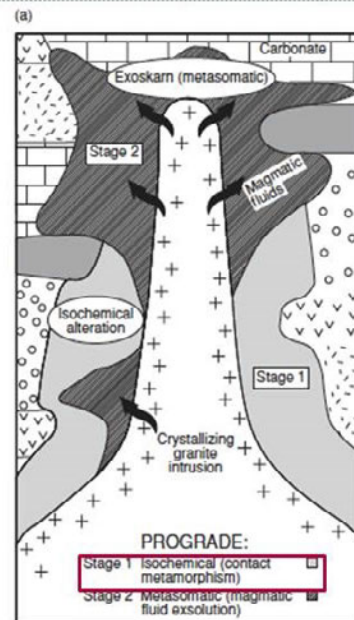




Skarn formation

1. Prograde: contact metamorphism

- Pluton intrudes country rock
- Thermal maximum of 600° - 650°C (Mathieson and Clark, 1984)



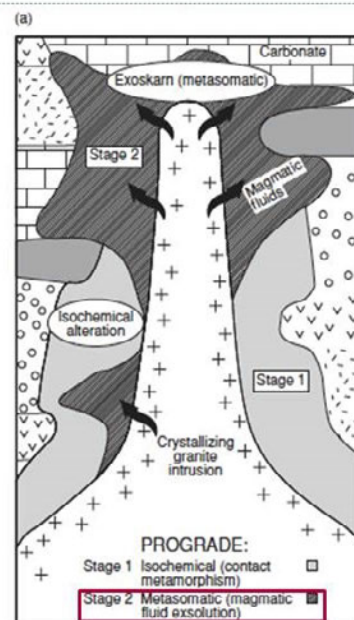
(Robb, 2005)

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Skarn formation

2. Prograde: metasomatism

- Magmatic fluid exsolution
 - ✦ Enrichment of H_2S , $FeCl_2$, $MgCl_2$, & KCl in main mineralizing fluid
 - ✦ Incompatible element enrichment in latest exsolved fluids
- Egress of fluid phase into contact metamorphosed rock
- Periodic trapping of fluid at crystallization front = over-pressuring
- Pervasive metasomatism of aplite dykes (Rasmussen, 2011) = ongoing influx of magmatic fluid

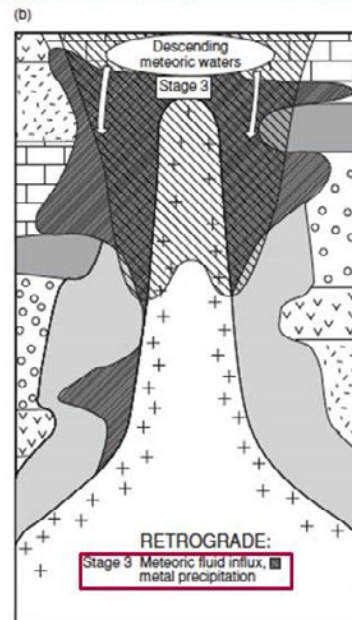


(Robb, 2005)

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Skarn formation

3. Retrograde: meteoric fluid influx & main metal precipitation
 - Magmatic fluid neutralized by carbonates = increased pH
 - ✦ Enhances permeability & fluid flow
 - Decreased temperatures
 - Promotes precipitation of metals + sulphides
 - ✦ Pyrrhotite, scheelite, chalcopyrite, native Bi and Bi-minerals



(Robb, 2005)

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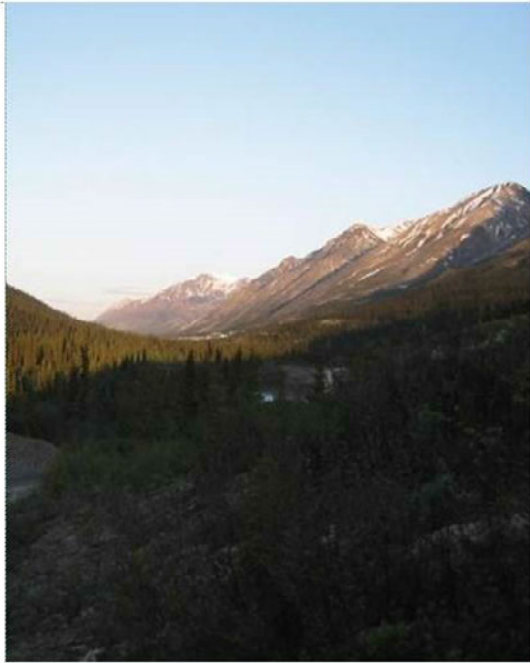
Objectives and purpose

- To characterize:
 - Distribution of Au
 - Petrogenesis of Au
 - Paragenetic sequence
- Purpose: are there economic concentrations of Au in the E-Zone?



Analytical techniques

- Sample E-Zone
 - 46 samples
- Bulk rock geochemistry
 - Acid digestion, ICP-MS
- Petrography
 - Optical microscopy
 - SEM analysis
- FEG-SEM analysis
- LA ICP-MS analysis



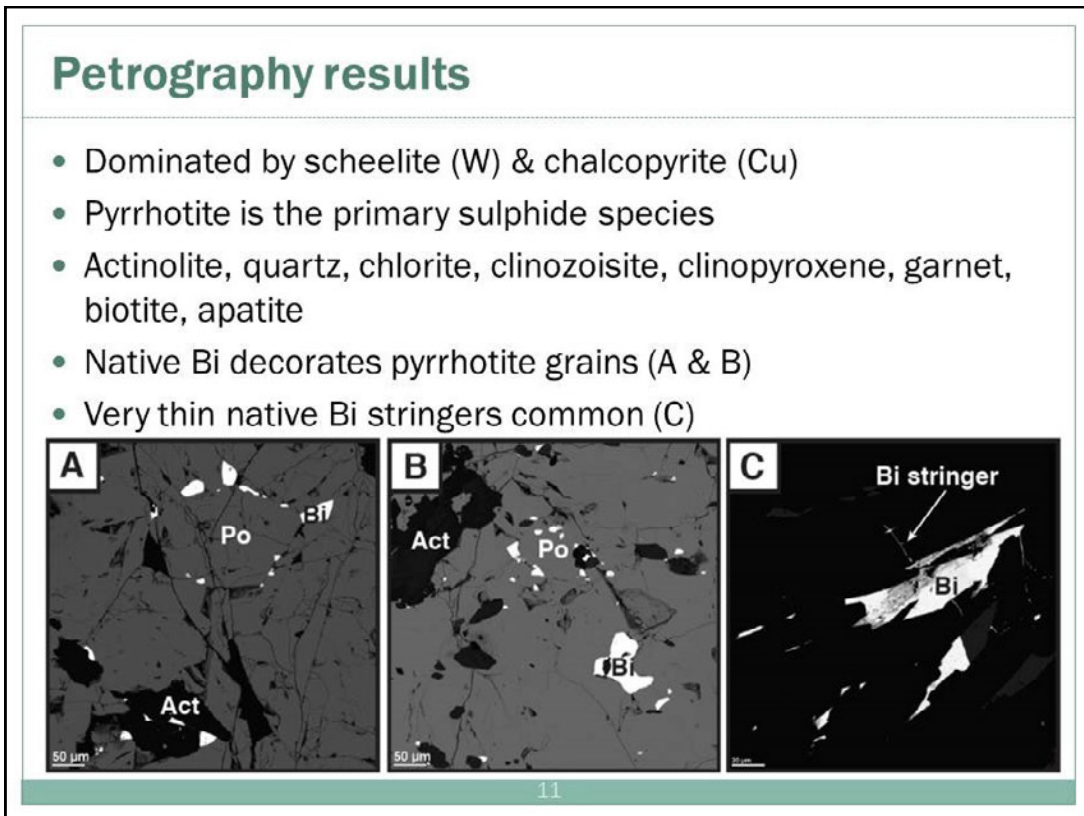
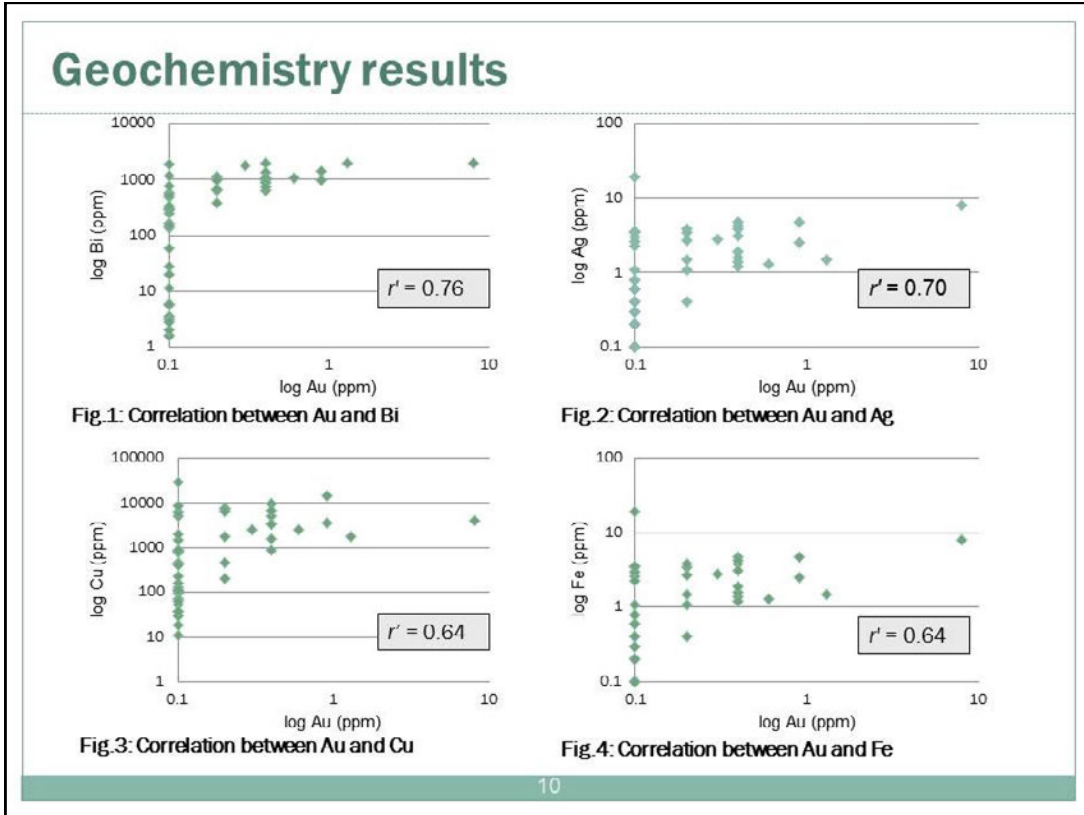
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Geochemistry results

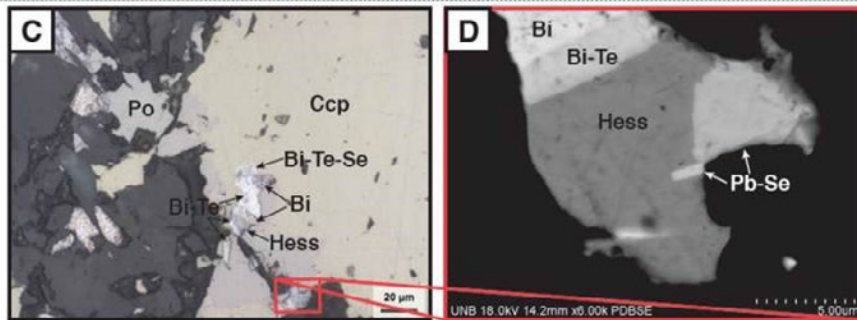
Sample#	03-DL-03	03-DL-04	03-DL-17a	03-DL-26	03-DL-27
Au (ppm)	0.9	0.6	0.9	1.3	8.0

- Spearman rank correlation coefficient
 - Measures statistical dependence of 2 variables
 - $r' = +1$: area rich in element 1 is likely to also have Au present
 - $r' = -1$: area rich in element 1 is likely to have no Au present
- Te and Se were not analyzed

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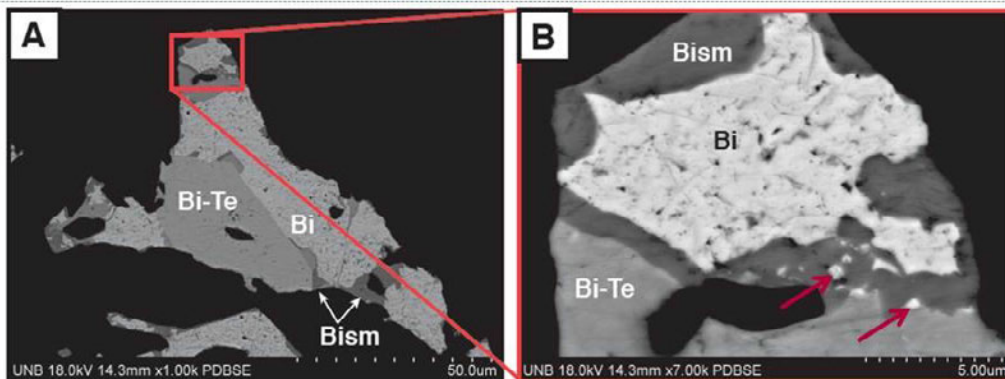
Petrography results



- Bi-sulfosalts, Bi-tellurides surround native Bi
- Native Bi & Bi-minerals found in association with zones of actinolite skarning
 - Hessite (Ag_2Te), Bi-tellurides, bismuthinite (Bi_2S_3) + sulfosalts
- No free Au present

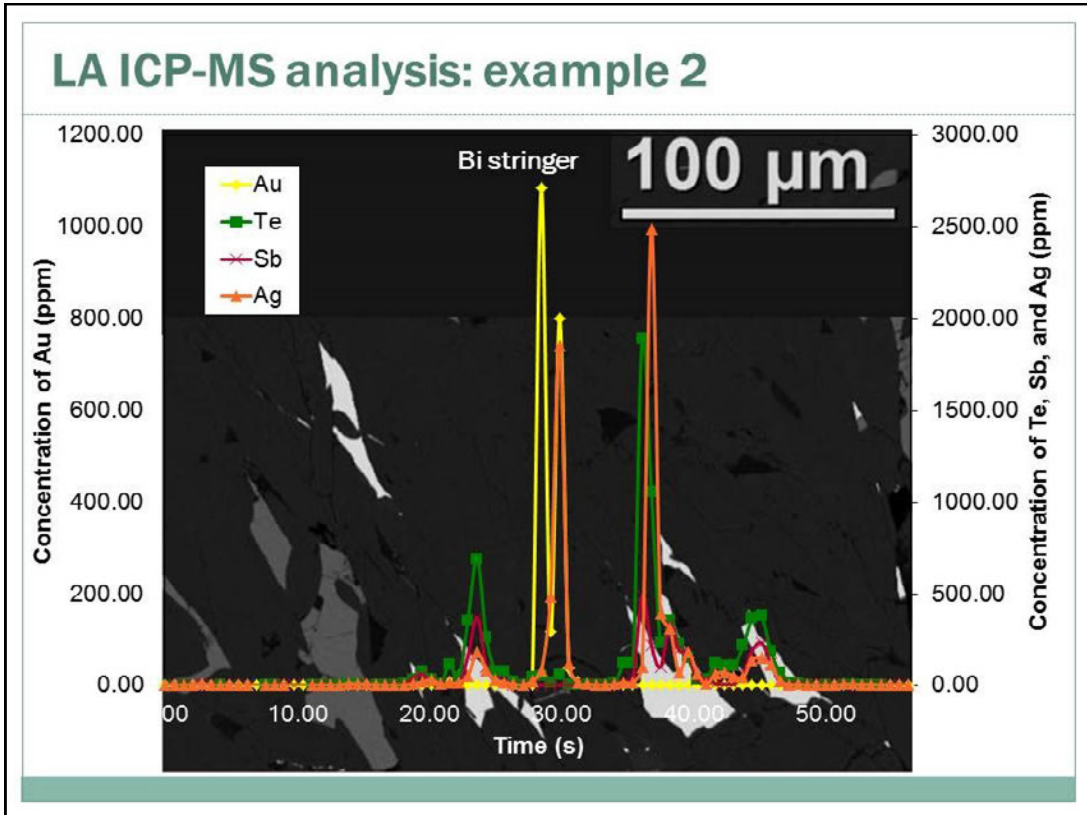
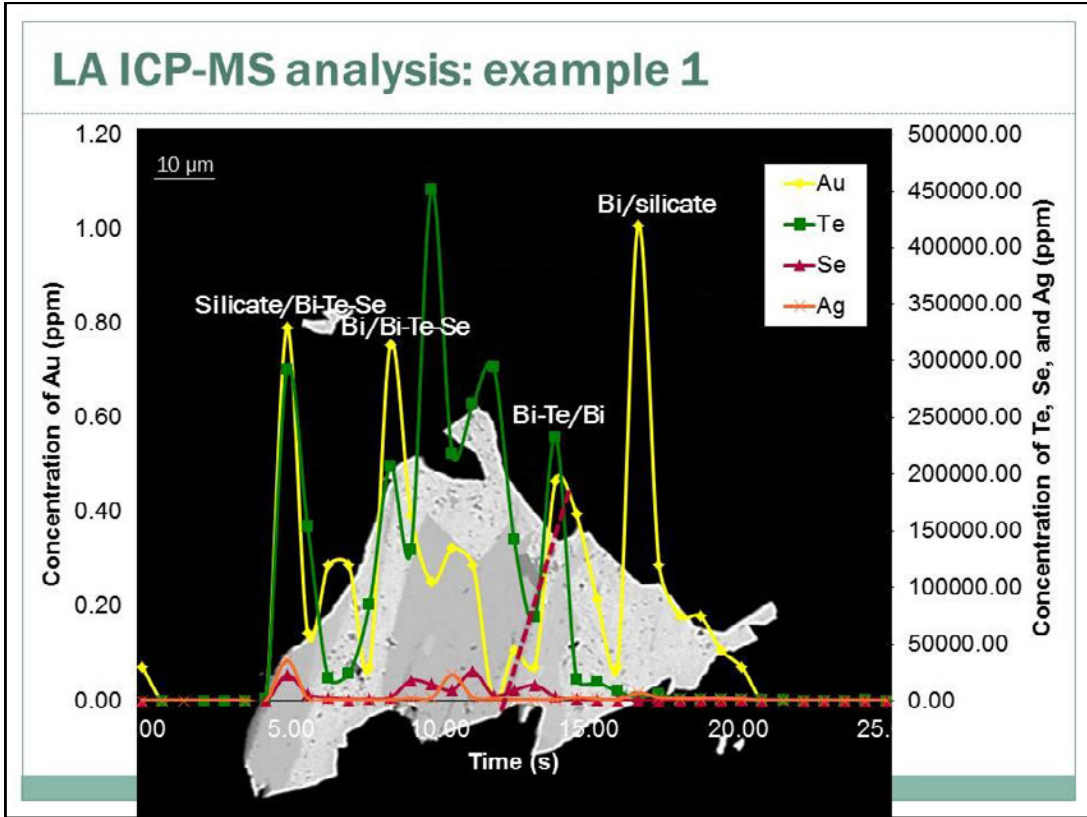
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FEG-SEM results



- Nano-inclusions in minerals, interstitial blebs, & stringers
 - Small ($<1\ \mu\text{m}$), rounded to sub-rounded
- All that were analyzed = native Bi

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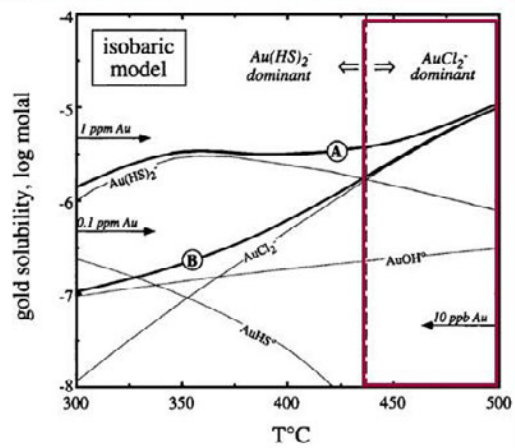


Distribution of Au

- Au concentrated on outer rim of Bi-mineral grains
- 3 possibilities:
 1. Solid solution as invisible Au within outer rim of Bi-minerals
 2. Au nano-inclusions within/on edges of native Bi & Bi-minerals
 3. Both lattice-bound & included as nano-inclusions

Au petrogenesis

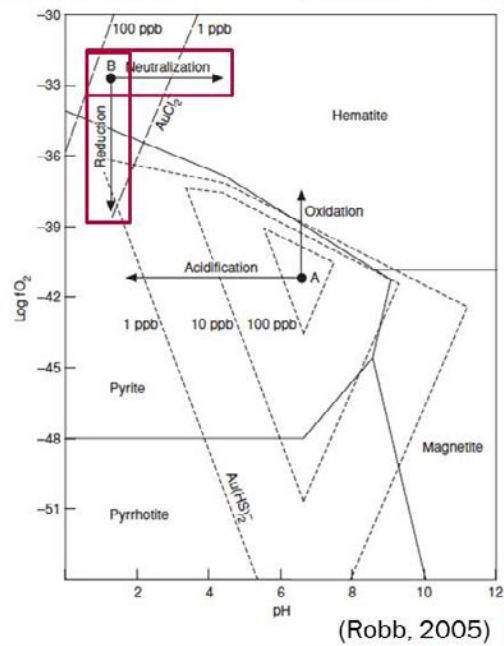
- Au dissolves mainly as AuCl_2^- at high temperatures
- Model A: high H_2S in fluid
- Solubility highest at high temperatures



(Gammons and William-Jones, 1997)

Au petrogenesis

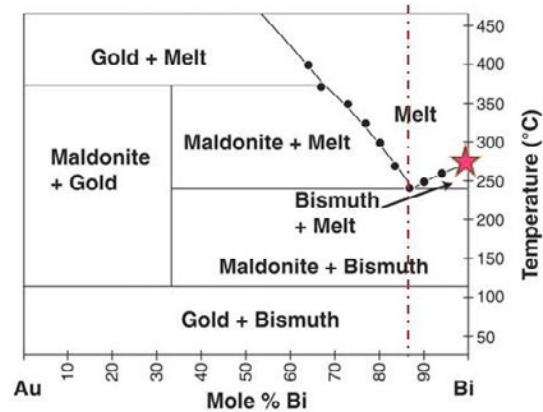
- Au dissolves mainly as AuCl_2^- at high temperatures
- Model A: high H_2S in fluid
- Solubility highest at high temperatures
- Au precipitation:
 1. Neutralization &/or
 2. Reduction
- Likely remobilized by Cl^- complexes and later fluid deposited with Bi



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Au petrogenesis

- Liquid bismuth collector model: potential explanation for Au-Bi deposits (Douglas et al., 2000)
- Au partitions into Bi-melts even from undersaturated aqueous solutions (Tooth et al., 2008)

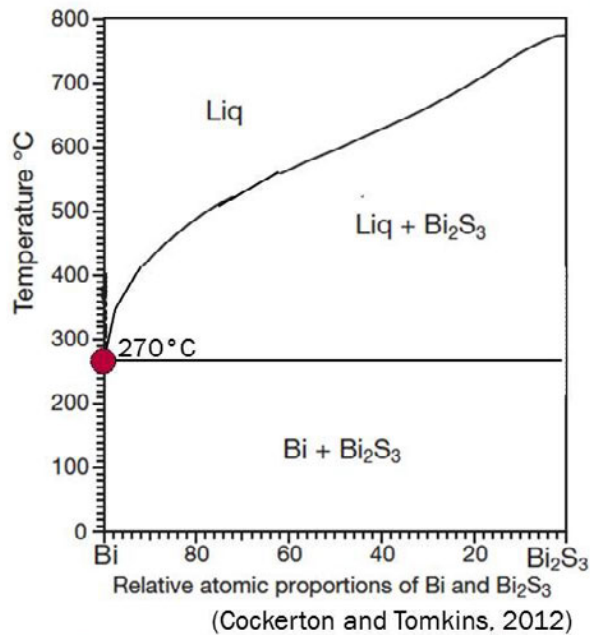


Phase diagram for Au-Bi system at 1 bar (Tooth et al., 2008). Native Bi has a melting temperature of 270°C. Au-Bi melt has a eutectic with ~12 atomic % Au.

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Au petrogenesis

- Likely aqueous Bi complex = $\text{Bi}(\text{OH})_3$
 - Precipitation of Fe^{3+} minerals = reduction of Bi^{3+} to Bi^0 (melt)
- Alteration assemblages > 300°C (Mathieson and Clark, 1987)
 - Bi- Bi_2S_3 assemblage partially liquid for much of ore formation (270°C)
 - Liquid Bi reacting with skarn fluid system
- Crystallized at eutectic or lower Bi-Au-Ag-Pb-Te-Se-S eutectic (Cockerton and Tomkins, 2012)



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Conclusions

- No free Au present
 - Solid solution of Au on rims of Bi-minerals
 - Higher concentrations of Au likely occur as nanoparticles, although none identified
- Au transported from depth as AuCl_2^-
- Incorporated in Bi-minerals due to high affinity of Bi and Au
- Area at margin of reaction front more promising for Au exploration where massive sulphides are located
 - Fluid movement and remobilization constant within front until cooled

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Paragenetic sequence

- Contact metamorphism:
 - Garnet, clinopyroxene
- Silication:
 - Actinolite, epidote, biotite, apatite, chlorite
- Main mineral precipitation:
 - Pyrrhotite, chalcopyrite, scheelite, sphalerite
 - Native Bi, bismuthinite, Bi-tellurides/Bi-Te-Se sulfosalts, hessite
 - Gold

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