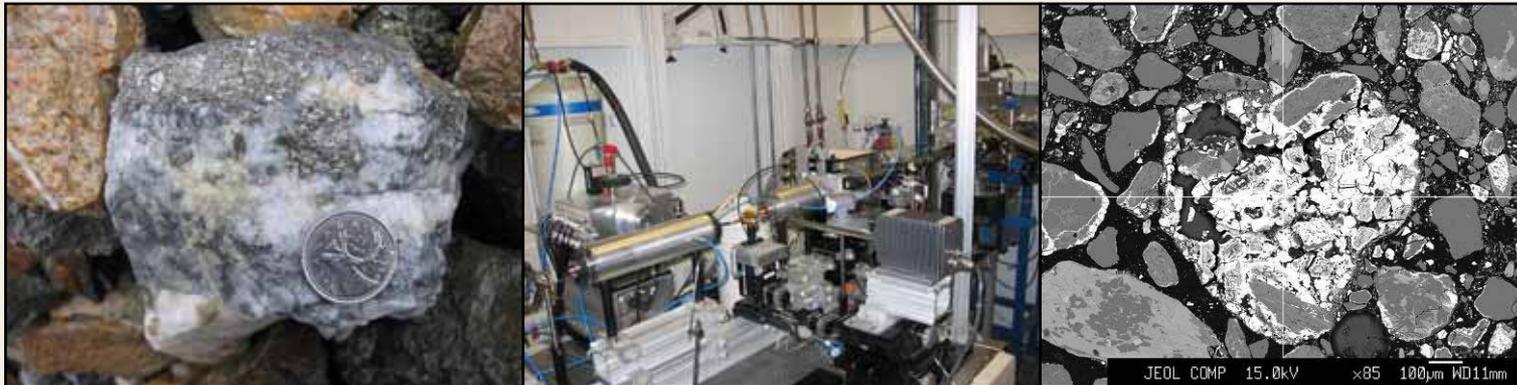




*Workshop on the Role of Geochemical Data in  
Ecological and Human Health Risk Assessment  
March 17-18, 2010, Halifax NS*

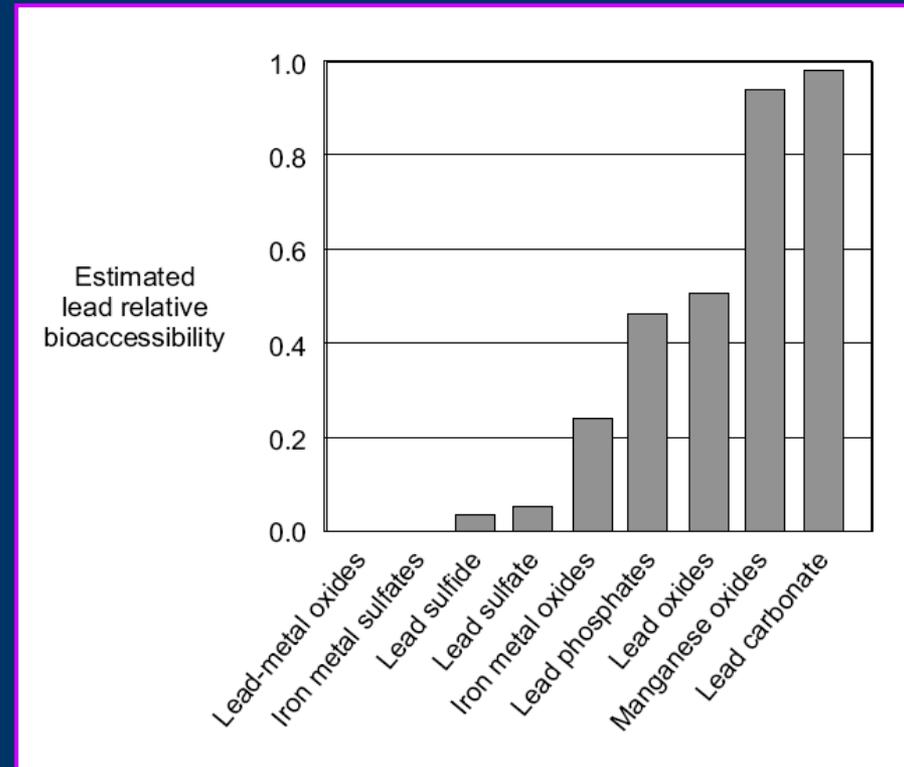
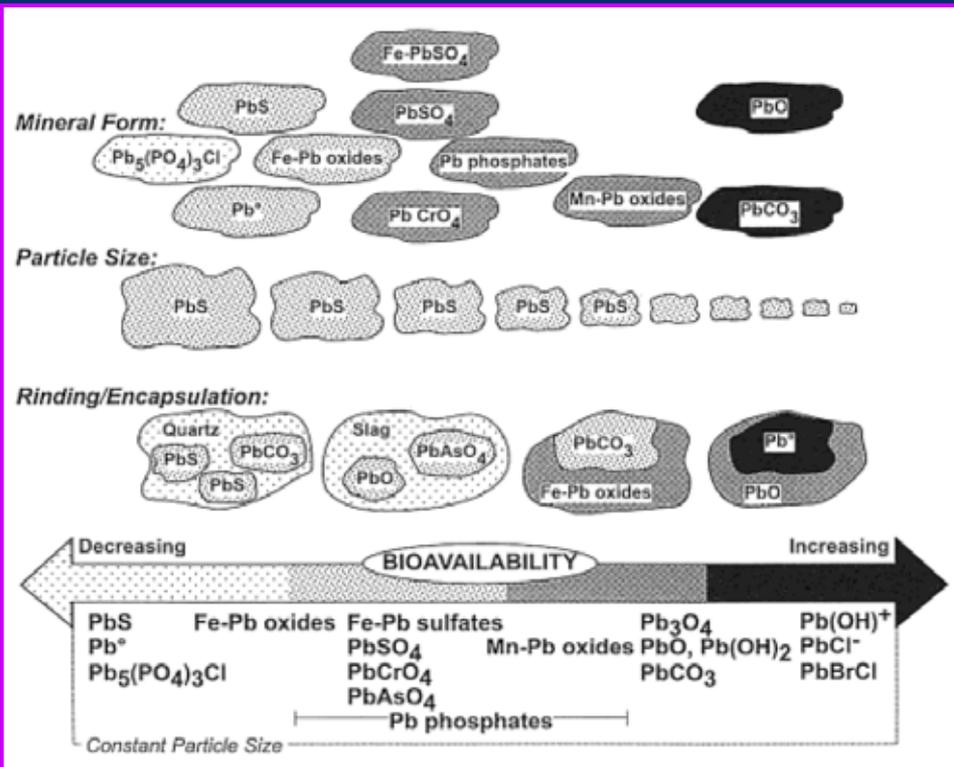


**Analytical Methods Used to Characterize the  
Solid-Phase Speciation of Metal(loid)s**

*Michael B. Parsons  
Geological Survey of Canada (Atlantic)*

# Why is Metal(loid) Speciation Important?

## Solubility and bioaccessibility of Pb minerals



Schematic of how different lead species, particle sizes, and morphologies affect lead bioavailability (*Ruby et al. 1999*).

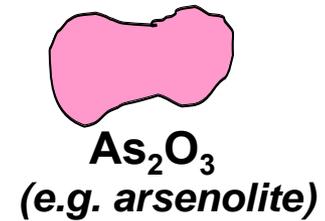
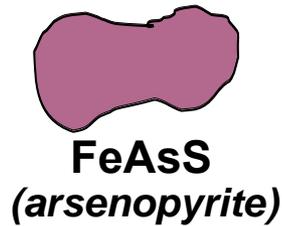
Relative bioavailability of Pb from soils (determined by swine uptake studies) as a function of the Pb-bearing minerals present in the soil (*Casteel et al. 2006*).

# Nature of As-hosting phase affects the risk of As exposure

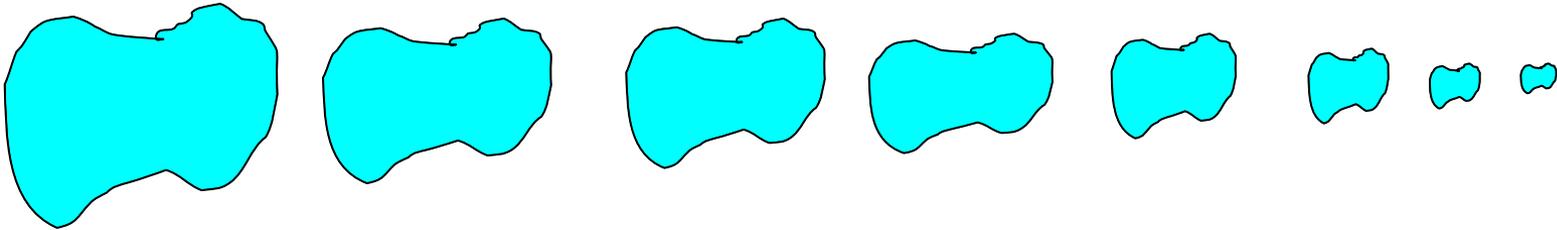
Modified from Ruby *et al.*, 1999



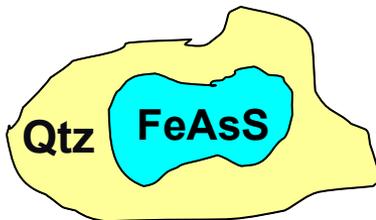
## Mineral Form



## Grain Size



## Encapsulation or Rimming of Grains



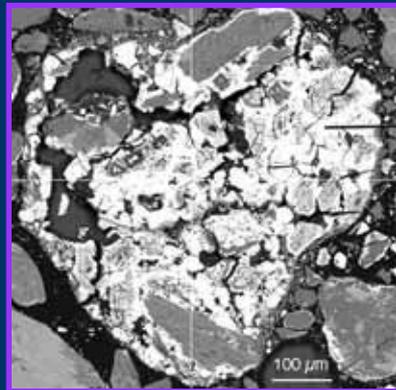
# Characterization of Mineral Hosts for As

Arsenopyrite  
(FeAsS)

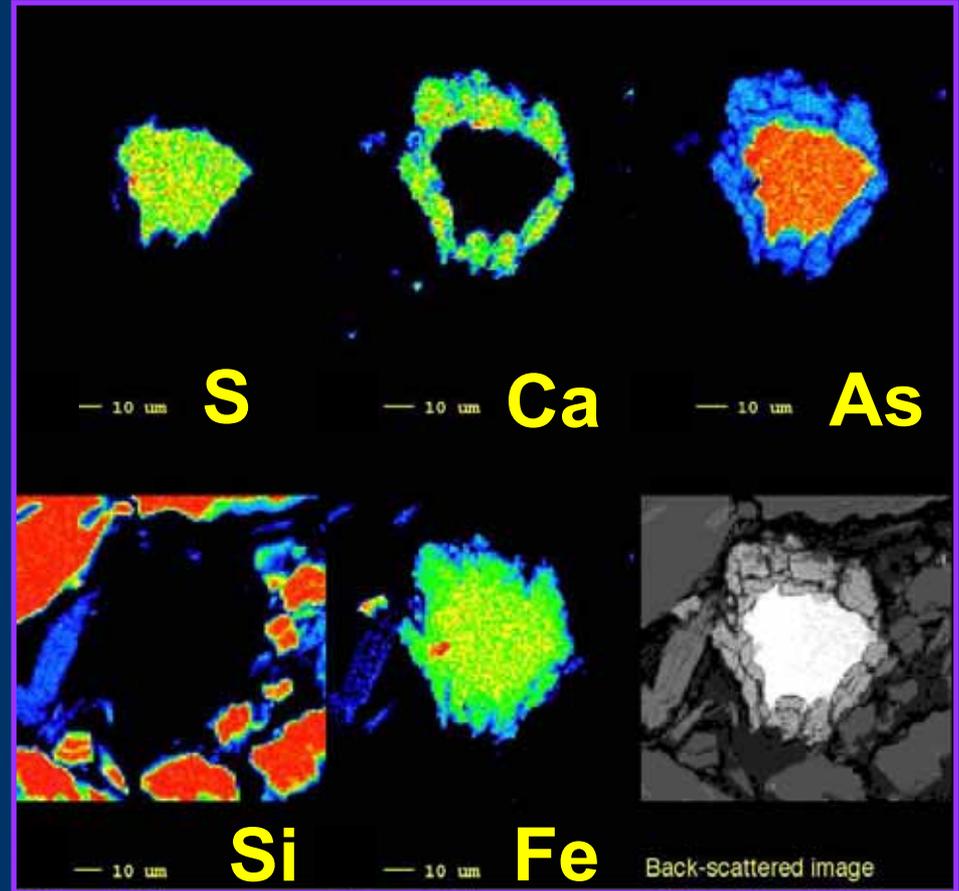


Crystalline  
scorodite  
(FeAsO<sub>4</sub>·2H<sub>2</sub>O)

Amorphous  
Fe arsenate  
cement

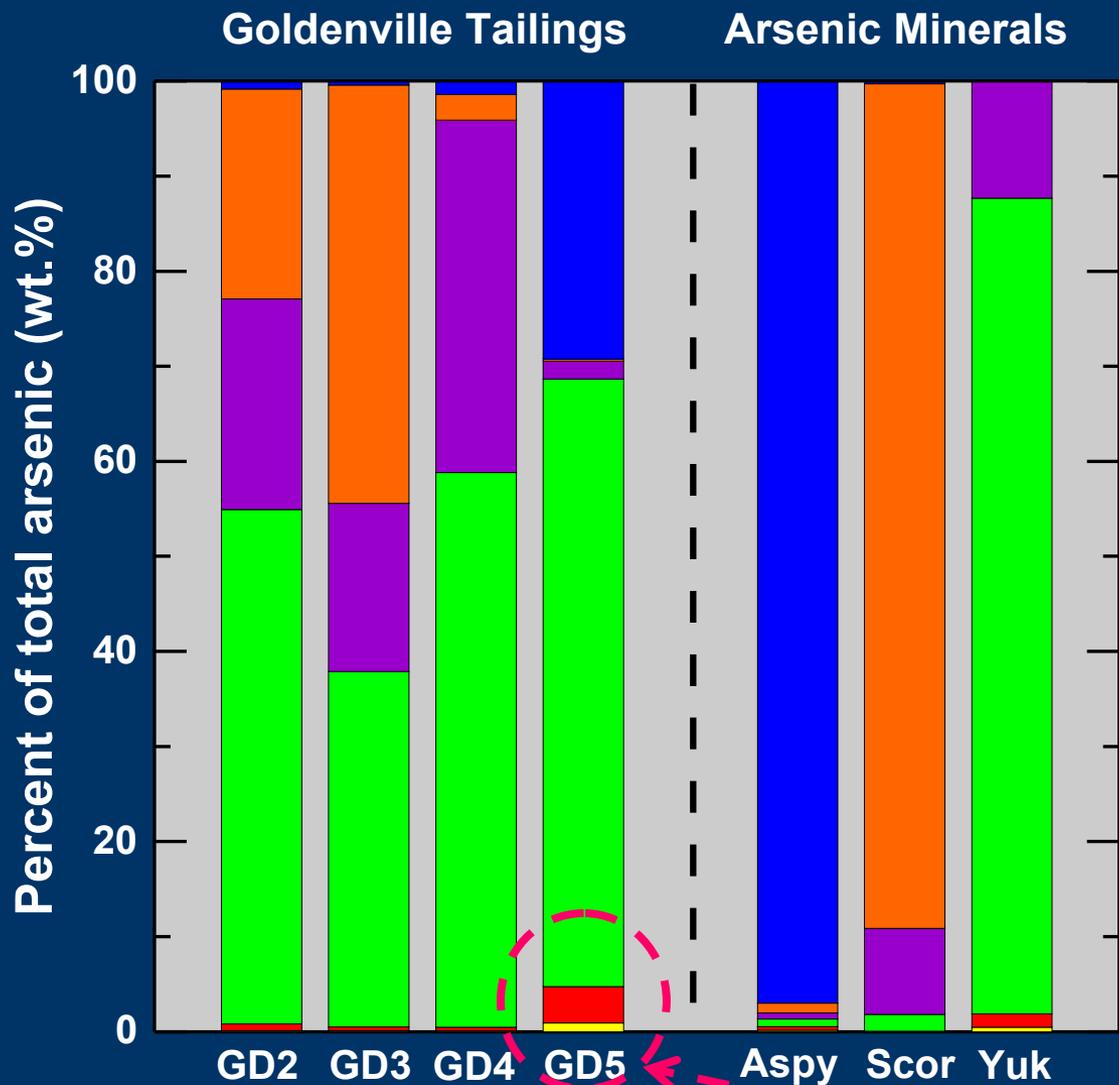


Compositional maps from electron microprobe



As-Ca bearing Fe oxyhydroxide  
reaction rim on arsenopyrite

# Bulk Speciation Methods: Sequential Chemical Extraction



## Legend

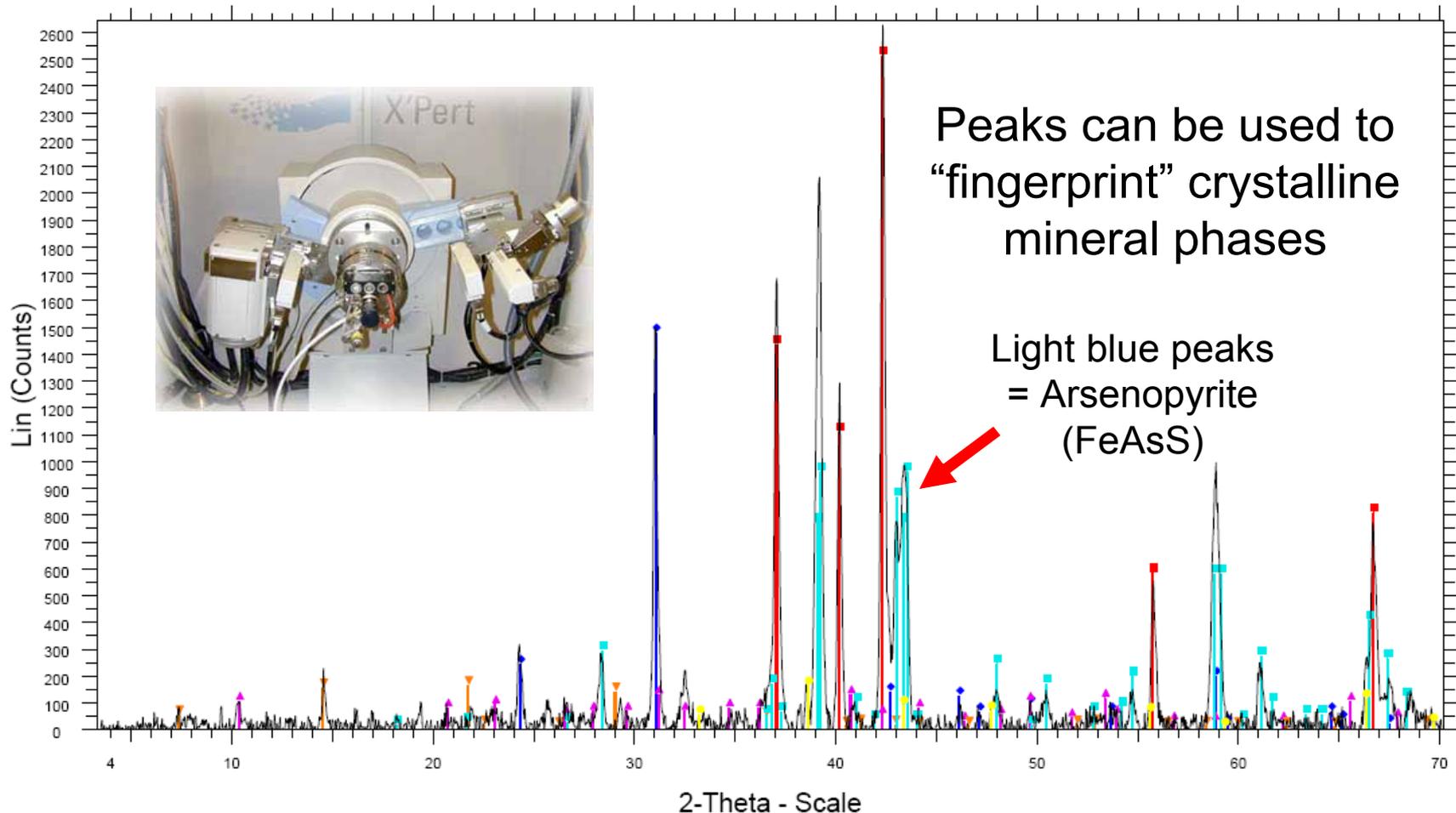
- Silicates & residuals (HF-HClO<sub>4</sub>-HNO<sub>3</sub>)
- Sulfides (*Aqua regia*)
- "Scorodite" (4 M HCl)
- Cryst. Fe oxides (1 M NH<sub>2</sub>OH + HOAc)
- Amor. Fe oxides (0.25 M NH<sub>2</sub>OH + HCl)
- Carbonates (1 M NH<sub>4</sub>OAc @ pH 5)
- Adsorbed / exchang. (1 M NH<sub>4</sub>OAc @ pH 7)

## Mineralogy Results

- GD2: Scor ≈ Amor. Fe arsenate > Amor. Fe oxide
- GD3: Scor ≈ Amor. Fe arsenate
- GD4: Amor. Fe arsenate > As-bearing Fe oxides
- GD5: Aspy ≈ Yukonite ≈ Amor. Fe oxide

Sample with highest As bioaccessibility (48%)

# Bulk Speciation Methods: X-ray Diffraction



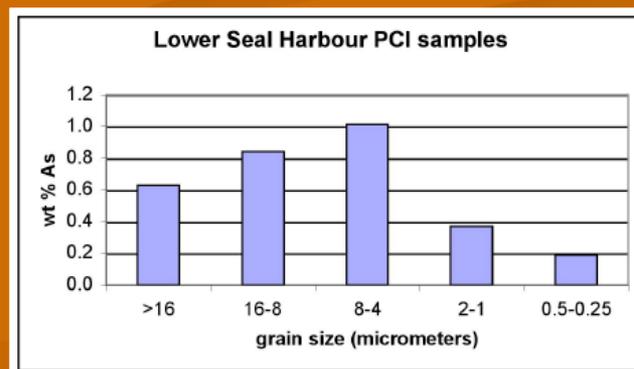
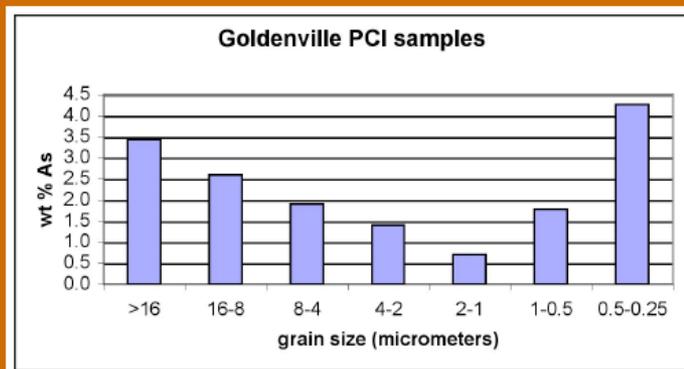
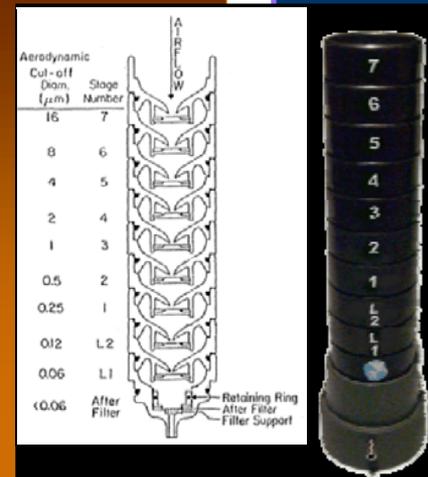
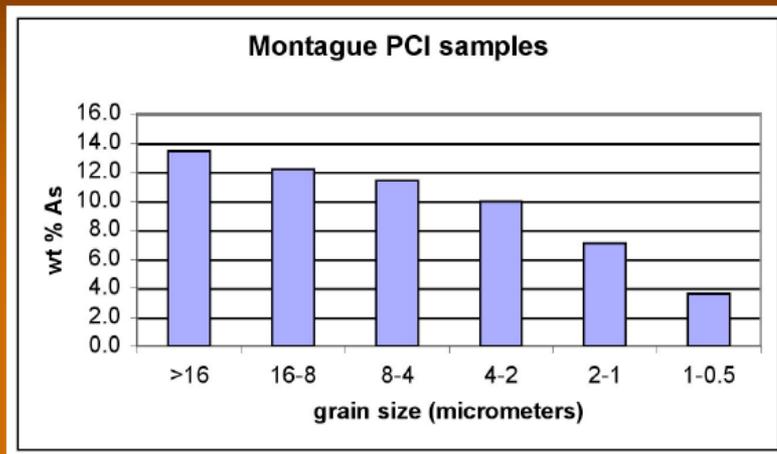
CAR05T2.mpsdpk - File: CAR05T2.mpsdpk.raw - Type: 2Th/Th locked - Start: 3.000 ° - End: 70.000 ° - Step: 0.040 ° - Step time: 2. s - Temp.: 27 °C - Time Started: 43 s - 2-Theta: 3

- 00-036-1451 (\*) - Zincite, syn - ZnO - Y: 78.11 % - d x by: 1. - WL: 1.78897 - 0 -
- 00-046-1045 (\*) - Quartz, syn - SiO<sub>2</sub> - Y: 45.91 % - d x by: 1. - WL: 1.78897 - 0 - I/c PDF 3.4 -
- 00-002-0467 (D) - Muscovite - KAl<sub>2</sub>(Si<sub>3</sub>Al)O<sub>10</sub>(OH,F)<sub>2</sub> - Y: 3.81 % - d x by: 1. - WL: 1.78897 - 0 -
- 00-012-0185 (D) - Clinocllore - (Mg,Fe,Al)<sub>6</sub>(Si,Cr)<sub>4</sub>O<sub>10</sub>(OH)<sub>8</sub> - Y: 4.87 % - d x by: 1. - WL: 1.78897 - 0 -
- 00-042-1320 (I) - Arsenopyrite - FeAsS - Y: 29.76 % - d x by: 1. - WL: 1.78897 - 0 -
- 00-042-1340 (\*) - Pyrite - FeS<sub>2</sub> - Y: 4.84 % - d x by: 1. - WL: 1.78897 - 0 - I/c PDF 1.6 -

X-ray diffractogram of arsenopyrite-bearing tailings, Caribou Gold District, NS

# Microbeam Methods: Proton-Induced X-ray Emission

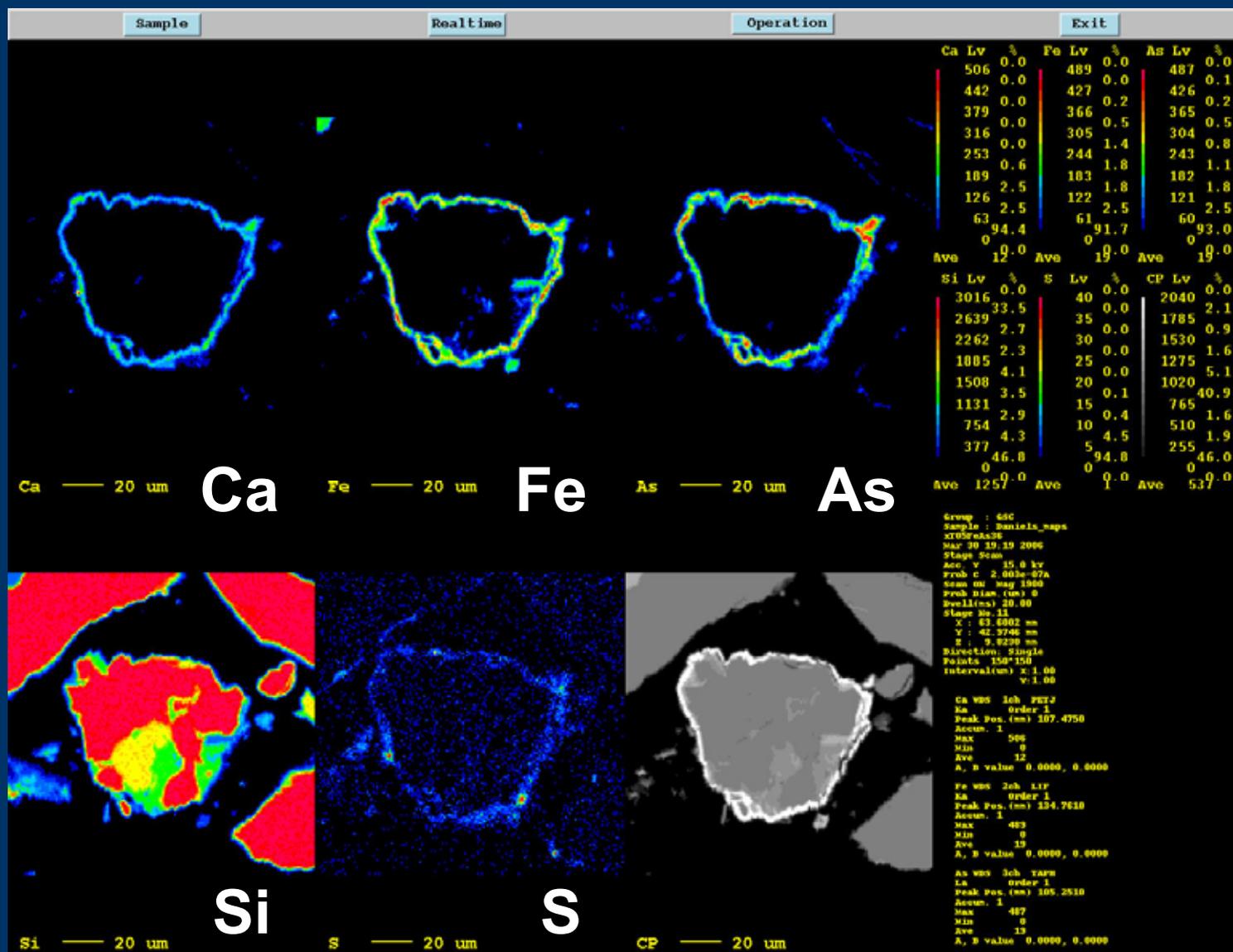
## PIXE results



*from Corriveau et al. (2010)*

# Arsenic Content of Airborne Particulates

# Microbeam Methods: Electron Microprobe Analyses



Ca-Fe-As rim on silicate mineral

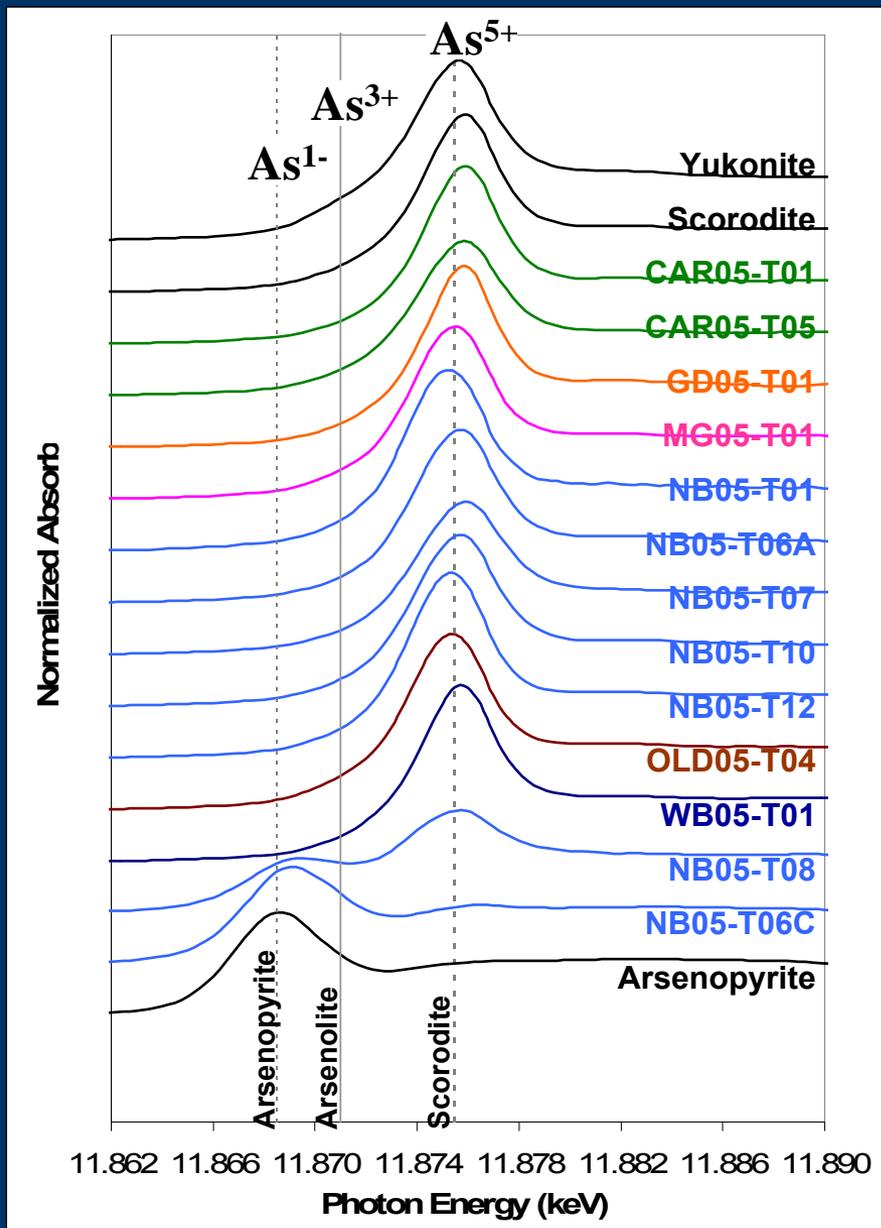
# Synchrotron Methods



Canadian Light Source

[www.lightsource.ca](http://www.lightsource.ca)

# Bulk XANES (X-ray Absorption Near-Edge Structure)



From Meunier et al. (2010)

XANES spectra of most gold mine tailings samples analyzed are similar to scorodite ( $As^{5+}$ )

North Brookfield sample NB05-T08 contains a mixture of arsenopyrite ( $As^{1-}$ ) and  $As^{5+}$

North Brookfield sample NB05-T06C spectra is similar to that of arsenopyrite

# Detached Thin Section Preparation & Location of Targets for EMPA and synchrotron microanalysis (Slide from Stephen Walker, Queen's University)

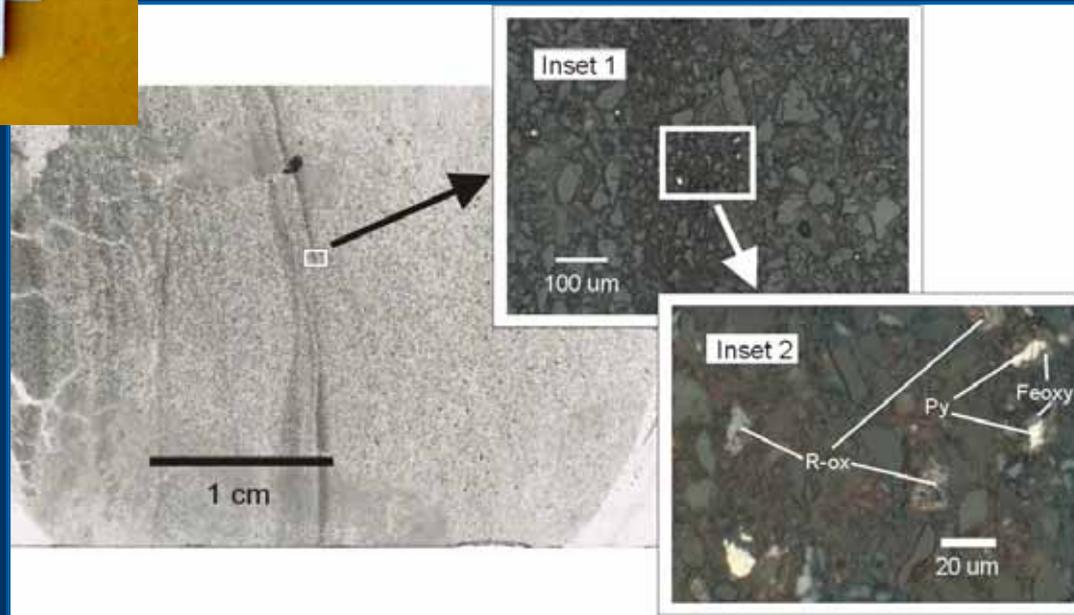


Thin-section lifted from slide & mounted on kapton tape-necessary for microXRD

Doubly polished thin section

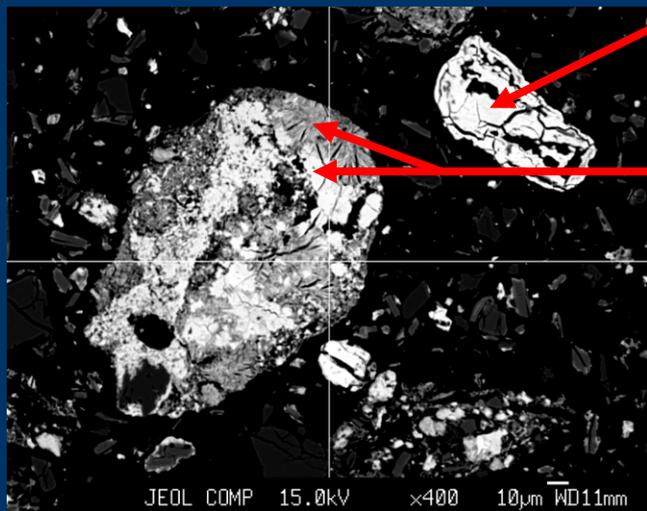
Targets selected petrographically & photographed

Other targets located using element mapping



# Combining petrography, electron microprobe (EPMA), micro-synchrotron methods

(Slide from Stephen Walker, Queen's University)



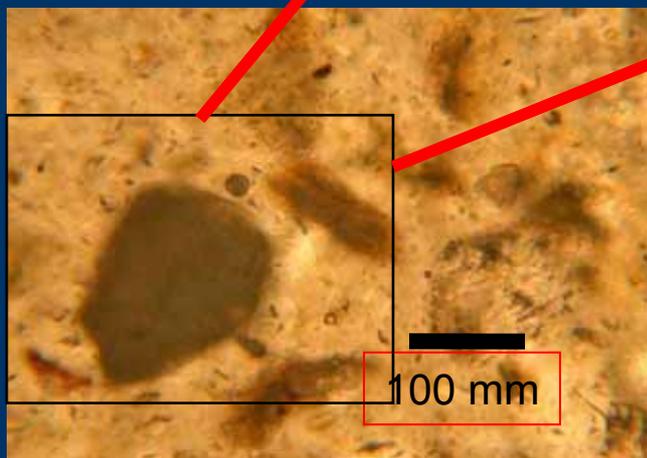
BSE image

Fe/As (molar) = 1.2  
(EPMA)

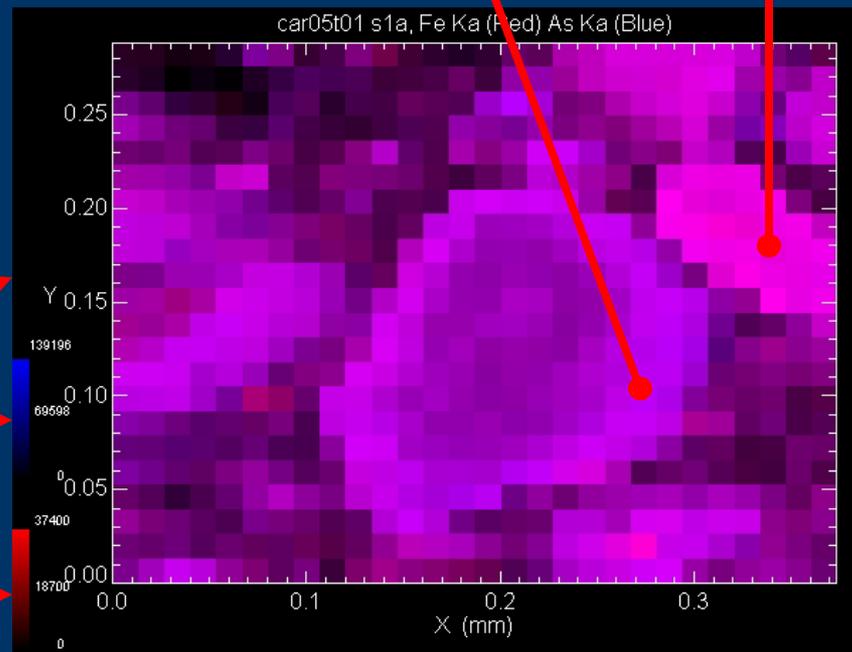
Fe/As (molar) = 0.9 to 1.0  
(EPMA)

Mixture of 2 different  
1As:1Fe arsenates

Amorphous



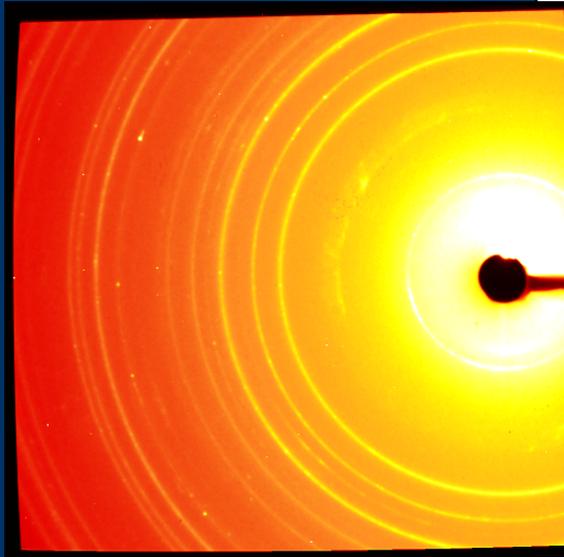
Transmitted and reflected light (20x objective)



µ-XRF map with µ-XRD targets  
and **identities**

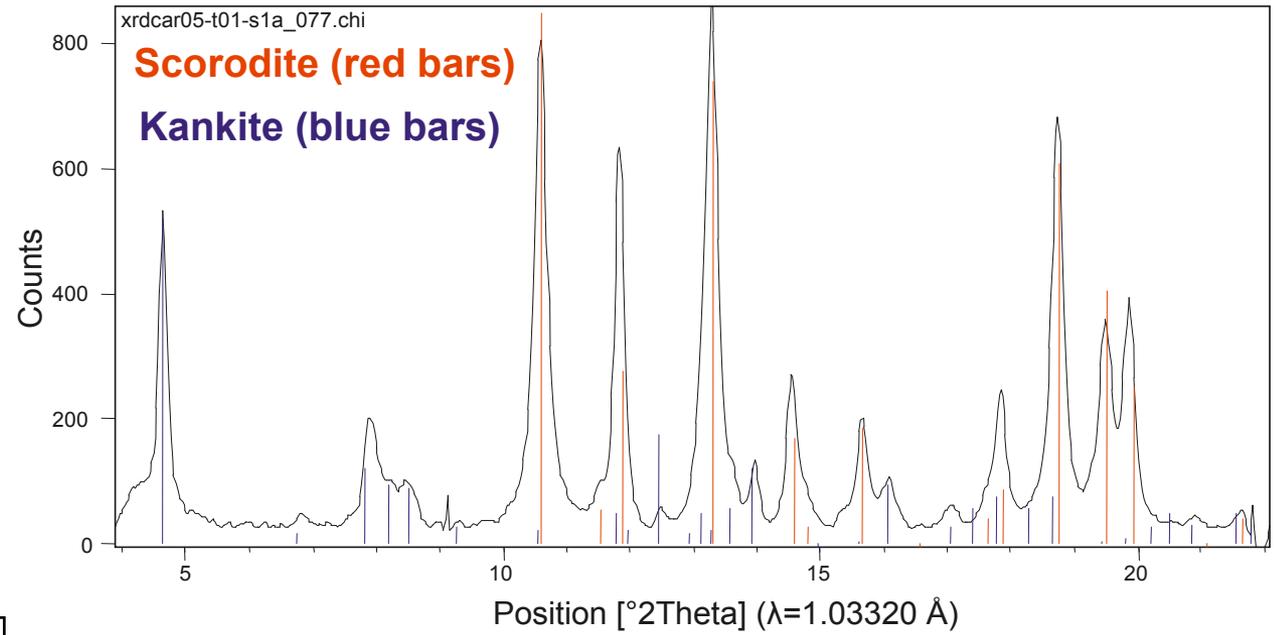
# Micro XRD (~10 μm) on larger grain identifies two microcrystalline Fe arsenate minerals

(Slide from Stephen Walker, Queen's University)



Mixture of:  
**Scorodite (FeAsO<sub>4</sub>·2H<sub>2</sub>O)**  
**Kankite (FeAsO<sub>4</sub>·3.5H<sub>2</sub>O)**

Simulated “powder” pattern (integration of 2-D pattern)



Peak List
01-070-0263; Scorodite
00-029-0694; Kankite

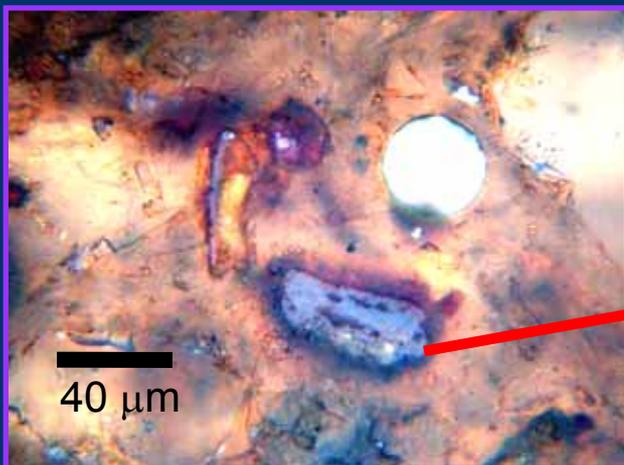
# Yukonite: an Ca-Fe arsenate mineral

(Slide from Stephen Walker, Queen's University)

Photomicrographs

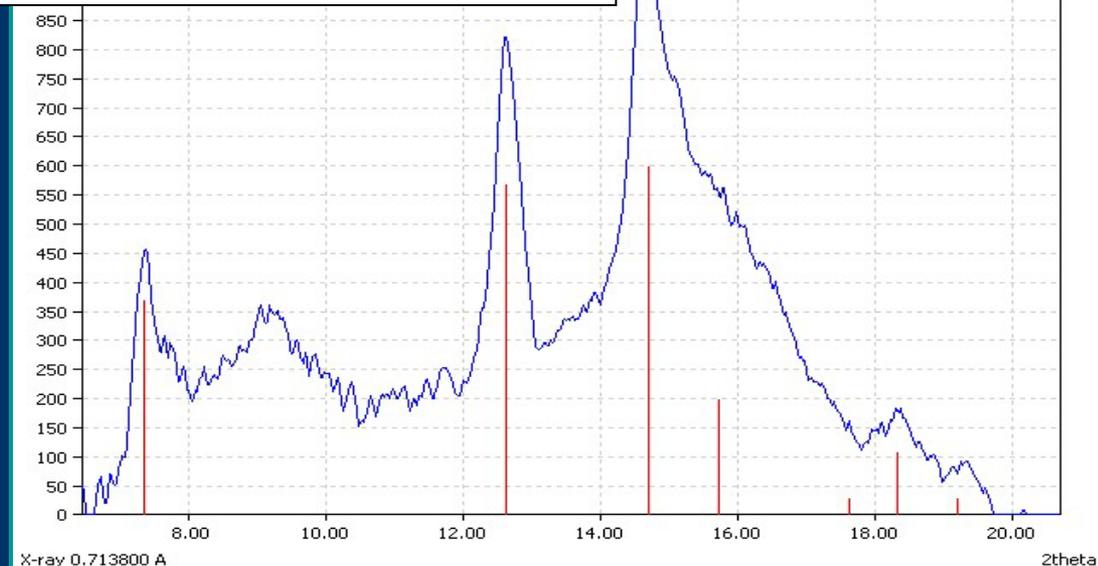


Reflected light

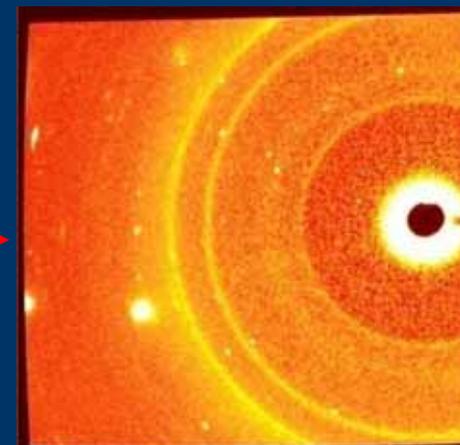


Transmitted and reflected light

## Yukonite



$\text{Fe}_2\text{O}_3 = 32.7\%$   
 $\text{As}_2\text{O}_5 = 35.6\%$   
 $\text{SO}_4 = 0.3\%$   
 $\text{CaO} = 7.4\%$

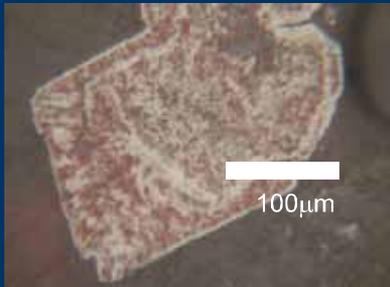


μ-XRD pattern

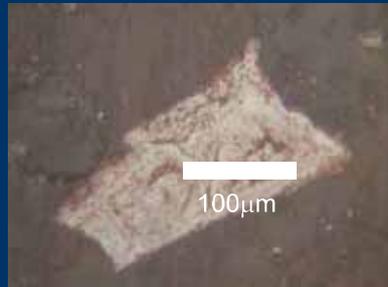
# Tailings from historical roasting of gold-ore North Brookfield Gold District

(Slide from Stephen Walker, Queen's University)

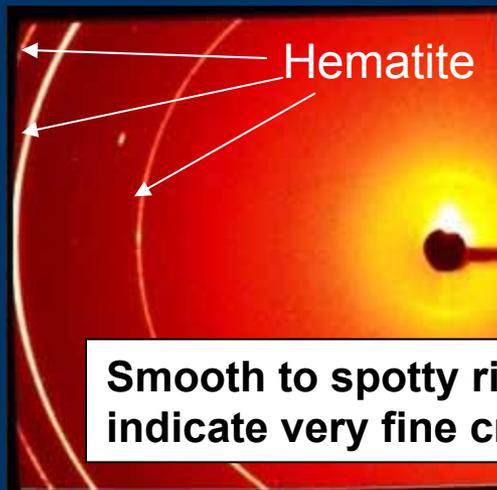
Abundant roaster derived As-bearing  
Fe-oxide (hematite  $\alpha\text{-Fe}_2\text{O}_3$  or  
maghemite  $\gamma\text{-Fe}_2\text{O}_3$ ) in some samples



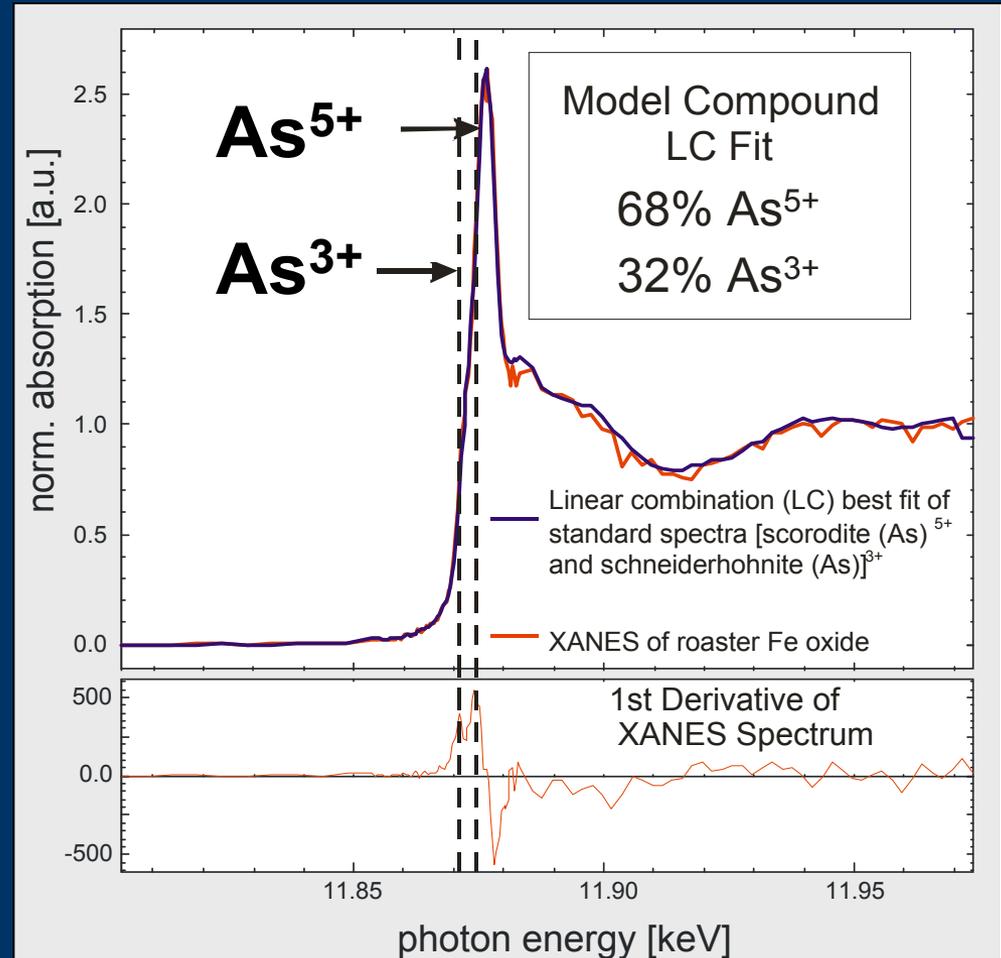
Reflected Light



Reflected Light



## $\mu\text{-XANES}$ data



Typically these roaster Fe-oxides contain  $<2\%$  As



# Key Points

- Metal(loid)s in soils, sediments, and mine wastes can be hosted in a wide range of primary and secondary phases.
- These phases have varying solubilities that strongly influence the environmental fate and bioaccessibility of metal(loid)s.
- The total concentrations of metal(loid)s in solid samples does not give sufficient information on the environmental availability of these elements, or their potential risks to human health.
- In the future, ecological and human health risk assessments should incorporate information on the solid-phase speciation of metal(loid)s to ensure that realistic management guidelines are established.



# References

- Ruby, M.V., Schoof, W.B., Brattin, W., Goldade, M., Post, G., Harnois, M., Mosby, D.E., Casteel, S.W., Berti, W., Carpenter, M., Edwards, D., Cragin, D., and Chappell, W., 1999. Advances in evaluating the oral bioavailability of inorganics in soil for use in human health risk assessment; *Environmental Science and Technology*, v. 33 (21), p. 3697–3705.
- Casteel, S.W., Weis, C.P., Henningsen, G.M., and Brattin, W.J., 2006. Estimation of relative bioavailability of lead in soil and soil-like materials using young swine; *Environmental Health Perspectives*, v. 114, p. 1162-1171.
- Corriveau, M.E., Jamieson, H.E., Parsons, M.B., Campbell, J.L., and Lanzirotti, A. 2010. Direct characterization of airborne particles associated with arsenic-rich mine tailings: Particle size, mineralogy and texture; Submitted to *Applied Geochemistry*.
- Meunier, L., Walker, S.R., Wragg, J., Parsons, M.B., Koch, I., Jamieson, H.E., Reimer, K.J. (2010) Effects of soil composition and mineralogy on the bioaccessibility of arsenic from tailings and soil in gold mine districts of Nova Scotia; *Environmental Science and Technology*, DOI: 10.1021/es9035682.

