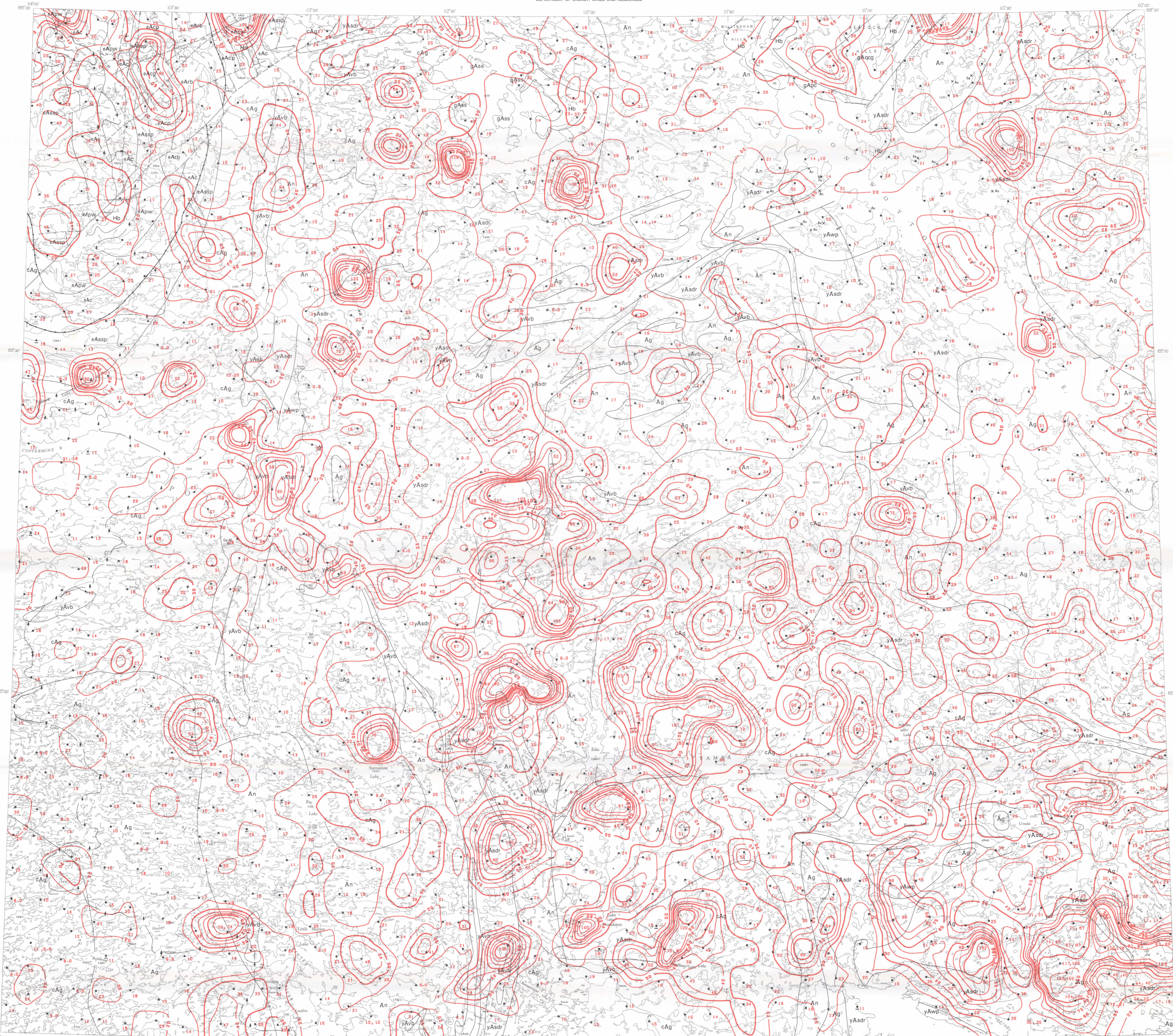


PRELIMINARY SERIES



- LEGEND**
- HADRINIAN**
- Hd Cabrolo sheets, sills
- APHERAN**
- AGS1 BROWN SOUND FORMATION: red siltstone, shale, sandstone
 - AGC PEACOCK HILLS FORMATION: shale, dolomite, sandstone
 - AGQ2 BURNSIDE RIVER FORMATION: quartzite, conglomerate
- EPWORTH GROUP**
- AVS1 TARTYKAR FORMATION: red sandstone, shale
 - AVC2 COWLES LAKE FORMATION: limestone, shale
 - AVP1 RECLUSE FORMATION: argillite, shale, greywacke
 - AVC ROCKNEST FORMATION: dolomite
 - AVS2 CONJACK FORMATION: sandstone, shale, argillite, sandstone
- PROTEROZOIC**
- Ag Quartz diorite, quartz monzonite, granodiorite, granite, in part porphyritic
 - An Granitic gneiss, migmatite, mixed gneisses involving Yellowknife rocks
 - CAG Complex of plutonic granitic rocks that may be, in part, older than Yellowknife Supergroup
- ARCHAIC**
- YAWP YELLOWKNIFE SUPERGROUP
 - YAVP Greywacke, shale
 - YASD Cordierite-andalusite bearing knotted schist and other metamorphic equivalents of YAWP
 - YAVI Intermediate to basic lava, tuff, agglomerate, and undifferentiated acidic volcanic rocks
- Geological symbols:**
- Boundary between Bear and Slave geological provinces
 - Fault, observed or assumed
 - Approximate position of faultline
 - Syncline
 - Mineral prospect showing principal element(s)
 - Lake sample site and metal concentration (sediment sieved to minus 250 mesh)
 - Lake sample site and metal concentration (sediment sieved to minus 100 mesh)
 - Geochronological concentration contours as ppm
- MINERALS**
- Copper Cu
 - Iron Fe
 - Gold Au
 - Nickel Ni
- Geology after unpublished map compiled by J. C. McElGynn, 1971
- Field work by R. J. Allan, E. M. Cameron, C. C. Durham, R. Benson, R. Collier, R. Cumming, G. Lund, D. Mann, C. Priebe, G. Thomas and R. Woronuk
- Analyses by J. J. Lynch, Alice L. MacLaurin, A. P. Lemieux and R. T. Crook
- Marginal notes by R. J. Allan and E. M. Cameron
- Geochronological contours and metal concentration numbers drawn by computer drum plotter
- Geological cartography by the Geological Survey of Canada
- Any revisions or additional geological information known to the user would be welcomed by the Geological Survey of Canada
- Base-map assembled by the Geological Survey of Canada from maps published at the same scale by the Army Survey Establishment, S.C.E. in 1965, 1964
- Copies of the topographical maps covering this map-area may be obtained from the Canada Map Office, 613 Booth Street, Ottawa, Ontario K1A 0Z9
- Mean magnetic declination 1973, 34° 20' East, decreasing 8.7' annually. Readings vary from 31° 54' to the SE corner to 37° 11' in the NW corner of the map-area
- Elevations in feet above mean sea-level

Lake Sediment Sampling The lake sediment samples were collected by post-hole auger from a helicopter. They were taken near the edge of the lake to water 3 to 4 feet deep. They comprise approximately the top 10 inches of sediment, less the surface layer. Of the variety of sediment types that may occur in lakes, the type of sample sought was of clay to silt grade and low in organic material.

Environmental Effects Certain heavy metals, such as uranium, zinc, and copper, may be enriched in sediment samples containing organic matter or iron and manganese oxides. The content of Fe and Mn has been determined for all samples, together with an index of the organic content (optical density). These data will be released later in the form of a computerized map. These data may be examined when assessing anomalous areas on this map. It should be recognized, however, that metal enrichment on these materials is a highly variable phenomenon that depends on a variety of factors, particularly the availability of metals in the environment. Also the Fe and Mn content of a sediment is not a direct measure of the amount of free oxides present. For these reasons the maps for the heavy metals are not presented on a statistically "corrected" basis. Further, since manganese and iron may be associated with these metals in the original rocks or primary mineralization, "correcting" the data may remove evidence of mineralization from the resulting map. If it is suspected that free oxides or organic material have created a false anomaly, then for those who do not have access to the original samples, the most satisfactory method of checking is to re-sample the area of the anomaly.

Sample Preparation and Analysis Sediments were dried, then sieved to minus 250 mesh to give a powder suitable for analysis. A few coarse samples were sieved to minus 100 mesh, then ball-milled before analysis.

For the zinc determination, 400 mg of the sample was digested on a water bath at 80°C for 1.5 hours with 6 ml of 4N HNO₃ plus two drops of concentrated H₂O₂. The sample was diluted to 20 ml with metal-free water, well shaken, and allowed to settle. This sample solution was aspirated through a 5 cm. single burner air-acetylene flame spectrophotometer. Measurements were made at 213mμ with a slit width equivalent to 20μ. The detection limit of the method is 2 ppm Zn, but all samples analysed contain greater than this amount.

While uranium, which is analyzed using the same acid leaching procedure, all of the zinc is not extracted from the sample. The percentage that is dissolved depends on the mineralogical form of zinc in each sample; in general it may vary between 50% and 90% of the total amount present. This method has the advantage that anomalies due to the dispersion of zinc from more easily leachable Zn minerals, perhaps site specific mineralization, are accentuated, relative to background.

Zinc in Rocks and Gneiss of the Survey Area During the 1971 orientation survey (Allan, Cameron and Durham, 1973) rock samples were collected from a number of areas within the Bear and Slave provinces. A selection of the data for these rocks and lake sediments from the same areas are given below as an aid to the interpretation of the lake sediment data. The analyses were made after an HNO₃-HCl attack. The High Lake and Hackett River areas contain zinc-bearing massive sulfide deposits; the Bull Lake area is not known to contain such deposits. The Terra Mine area contains native silver associated with nickel-cobalt-arsenide-arsenides and minor stratiform copper mineralization.

The analyses for Bull Lake show that intermediate and basic volcanic rocks contain more zinc than unmineralized acid volcanics. However, in this area, the sedimentary and granitic rocks contain more zinc than any of the volcanic rocks. The volcanic rocks associated with the deposit at High Lake have a higher level of zinc than the volcanic rocks of the Bull Lake area. The 90 percent values from this area and that at Hackett River, show that the rocks contain scattered mineralization, particularly the acidic varieties. The siliceous volcanic and sedimentary rocks at Hackett River have higher zinc concentrations than the Bull Lake acid volcanics. The zinc values for the Bull Lake porphyry may represent the general level of abundance for the intrusive rocks of the Great Bear Batholith. The Terra Mine data show evidence of slight zinc mineralization.

Province	Sample	Arithmetic Mean		Geometric Mean	Median	90 Percentile
		Zn	ppm			
High Lake:	basic volcanics	22	86	90	80	182
	intermediate volcanics	54	71	65	68	100
	acid volcanics	16	18	18	17	32
	lake sediments	31	139	120	124	-
Hackett River:	volcanics and sedimentary	27	68	42	43	146
	lake sediments	38	71	51	64	-
Bull Lake:	basic volcanics	55	48	40	42	98
	intermediate volcanics	21	59	38	50	96
	acid volcanics	23	23	23	22	32
	sedimentary rocks	61	76	76	81	96
	granulites	5	0	0	0	0
Terra Mine:	volcanic, sedimentary and intrusive rocks	65	139	74	72	275
	lake sediments	39	95	90	95	-

SLAVE PROVINCE

High Lake:

- basic volcanics
- intermediate volcanics
- acid volcanics
- lake sediments

Hackett River:

- volcanics and sedimentary
- lake sediments

Bull Lake:

- basic volcanics
- intermediate volcanics
- acid volcanics
- sedimentary rocks
- granulites
- lake sediments

BEAR PROVINCE

Bull Lake:

- porphyry
- lake sediments

Terra Mine:

- volcanic, sedimentary and intrusive rocks
- lake sediments

In the Bear Province, zinc is sometimes present as a minor constituent of the uranium and silver vein deposits. There are a number of other rock types present in the province that may be associated with zinc mineralization. Some possible associations are: arsenic mineralization with the high level granitic rocks to the west of the Wopmay Fault; volcanic-hosted silver mineralization within the mineralized sequence to the east of this fault; and lead-zinc mineralization with the Aghashan carbonate sediments.

Zinc-bearing massive sulfides of probable volcanic-exhalative origin are the most economically attractive exploration targets within the Slave Province. This mineralization is generally associated with the more acid volcanic rocks. In this province the volcanic rocks generally occur near the base of the rock sequence within the sedimentary volcanic belts—that is along the margins of these belts. These belts may be unroofed volcanic belts with the sedimentary sequence, the margins of these belts are prospective, even in the absence of mapped volcanic rocks. On the basis of orientation surveys around the High Lake and Hackett River deposits it was argued (Allan, Cameron and Durham, 1973) that this type of deposit was best outlined by a hierarchy of geochronological indicators in lake sediments:

- (1) By indicators that suggest that the area is underlain by acidic volcanic rocks (e.g. high Si or K, low Mg).
- (2) Within (1) zones underlain by exhalative facies rocks such as carbonates and iron-rich sediments (e.g. high Mn and Fe) or zones of rock alteration associated with mineralization (e.g. high Mg, high Ni).
- (3) Anomalies due to massive sulfide mineralization (e.g. high Zn, Cu, Pb or Ag).

In general, areas of ore potential may be revealed in lake sediments by broad areas of high zinc concentration or less extensive areas with sharp anomalies. In areas where deposits are found there is a tendency for the geometric mean zinc concentration of lake sediments to be higher than the corresponding means for volcanic rocks. This is because the zinc in the lake sediments has in part been derived by preferential leaching of sulfides.

Zinc in the Surficial Environment Zinc sulfides are relatively easily oxidized to soluble zinc salts such as ZnSO₄ and ZnCO₃. The element is amphoteric and forms the zinc ion (Zn²⁺) in acid solutions and the zincate ion (Zn(OH)₄²⁻) in basic solutions. In most of the survey area soils are poorly drained and the water table is close to the surface. The pH of the interstitial water is barely normally acid to neutral. In these conditions the zinc ion is highly mobile because of the least readily sorbed of the metal ions. Since zinc may travel considerable distances from ore deposits it is in excellent position to be sampled in reconnaissance surveys. Zinc may be sorbed onto the clays and the iron and manganese hydroxides of lake sediments. It has also a strong affinity for organic matter and may thus be enriched in the organic fraction of lake sediments. When considering zinc levels in sediments, it should be related to the organic content. An estimate of the organic content of each sample is being made.

Zinc in Lake Sediments, This Sheet There is a general correlation between the contour patterns for zinc and the geology of this sheet. This correlation is not in terms of rock units but there appears to be a much variation within areas mapped as one rock unit as between areas underlain by different rock units. The data given above for variation between different rock types occurring across the Bull Lake may, therefore, be valid only for the rocks of this particular area.

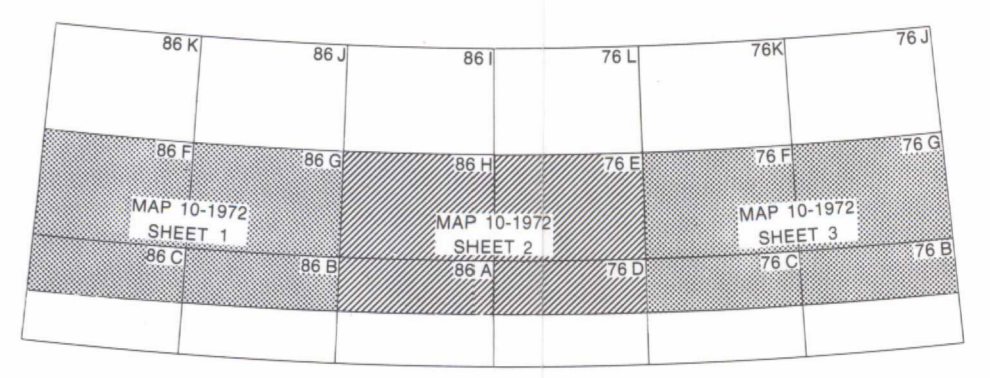
The background level of abundance for zinc across this sheet (arithmetic mean, 29.5 ppm Zn; geometric mean, 25 ppm Zn) is less than the levels obtained during the 1971 orientation survey, noted in the above table. The total area enclosed by the 150 ppm contour is relatively small. At High Lake 11 of 23 samples over 200 square miles exceeded this value and of 28 samples over 200 square miles at Hackett River. There are, however, a number of areas small and large that are anomalous relative to background. Many of these are geologically favourable areas, such as over volcanic rocks and along the margins of metasedimentary-volcanic belts. As was suggested in the section above, proper interpretation of the zinc anomalies can only be done after considering the distribution patterns of a number of other critical elements.

Some of the most prominent anomalies are related to the margins of large north-south trending sedimentary belts that occupy the Beban Lake-Point Lake-Deuparmit Lake sectors of the sheet. Bostock (1967) and Fraser (1958) who have mapped much of this belt observed a large number of anomalies. The geochronological data may help to focus attention on selected areas within the belt containing these anomalies. The other main metasedimentary belt in this sheet occurs in the north-west quadrant around Chazy Lake. The axial portion of this belt contains some of the lowest zinc levels of any part of the sheet. The areas adjacent to the margin of this belt contain some, but relatively few, anomalies.

The southeastern quadrant is relatively enriched in zinc compared to other parts of the sheet. Much of this area is underlain by granitic rocks of various types. The area should be contrasted with the relatively low zinc content of lake sediments overlying granites along the western margin of the sheet. In part, the areas of higher zinc content correlate with anomalous areas for uranium, such as seen near Yumbo Lake. Large relatively consistent anomalies that at Yumbo Lake are as likely to be caused by bedrock that has a higher dispersed metal content as by discrete mineralization. In follow-up work this question may be readily resolved by analysis of representative rock samples from the anomalous area. Some of the sharpest and highest anomalies on the sheet are in the southeast corner, south of Paul Lake to La de Grae and La de Sarrage. Seven samples have greater than 100 ppm Zn and three have more than 150 ppm. These anomalies are associated with above average contents of a number of elements, including Li, Sr, Mo, Cu, Co and Ni.

Geological Survey Paper 72-50, contains a detailed description of the experimental basis for lake sediment sampling in the Shield; the geology and metallogeny of the region; the organization, methods, and costs of the sampling operation; the methods of sample analysis. If references quoted above are not given here, they can be found in Paper 72-50.

MAP 10-1972
SHEET 2
**ZINC CONTENT OF LAKE SEDIMENTS
BEAR-SLAVE OPERATION
DISTRICT OF MACKENZIE**
Scale 1:250,000



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